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The present invention relates to reversibly variable electrochromic devices for varying the transmittance to light, such as electrochromic rearview mirrors, windows and sun roofs for motor vehicles, reversibly variable



Brief Description of the Related Technology

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Reversibly variable electrochromic devices are known in the art. In such devices, the intensity of light (e.g., visible, infrared, ultraviolet or other distinct or overlapping electromagnetic radiation) is modulated by passing the light through an electrochromic medium. The electrochromic medium is disposed between two conductive electrodes, at least one of which is typically transparent, which causes the medium to undergo reversible electrochemical reactions when potential differences are applied across the two electrodes. Some examples of these prior art devices are described in United States Patents 3,280,701 (Donnelly); 3,451,741 (Manos); 3,806,229 (Schoot); 4,712,879 (Lynam) ("Lynam I"); 4,902,108 (Byker) ("Byker I"); and I.F. Chang, "Electrochromic and Electrochemichromic Materials and Phenomena", in Nonemissive Electrooptic Displays, 155-96, A.R. Kmetz and F.K. von Willisen, eds., Plenum Press, New York (1976).

Reversibly variable electrochromic media include those wherein the electrochemical reaction takes place in a solid film or occurs entirely in a liquid solution. See e.g., Chang.

Numerous devices using an electrochromic medium, wherein the electrochemical reaction takes place entirely in a solution, are known in the art. Some examples are described in United States Patents 3,453,038 (Kissa); 5,128,799 (Byker) ("Byker II"); Donnelly; Manos; Schoot; Byker I; and commonly assigned United States Patents 5,073,012 (Lynam) ("Lynam II"); 5,115,346 (Lynam) ("Lynam III"); 5,140,455 (Varaprasad) ("Varaprasad I"); 5,142,407 (Varaprasad) ("Varaprasad II"); 5,151,816 (Varaprasad) ("Varaprasad III") and 5,239,405 (Varaprasad) ("Varaprasad IV"); and commonly assigned co-pending United States patent application Serial No. 07/935,784 (filed August 27, 1992)]. Typically, these

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electrochromic devices, sometimes referred to as electrochemichromic devices, are single-compartment, self-erasing, solution-phase electrochromic devices. See e.g., Manos, Byker I and Byker II.

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In single-compartment, self-erasing, solution-phase electrochromic devices, the intensity of the electromagnetic radiation is modulated by passing through a solution held in a compartment. The solution
10 often includes a solvent, at least one anodic compound and at least one cathodic compound. During operation of such devices, the solution is fluid, although it may be gelled or made highly viscous with a thickening agent, and the solution components, including the
15 anodic compounds and cathodic compounds, do not precipitate. See e.g., Byker I and Byker II.

Certain of these electrochemichromic devices have presented drawbacks. First, a susceptibility exists
20 for distinct bands of color to form adjacent the bus bars after having retained a colored state over a prolonged period of time. This undesirable event is known as segregation. Second, processing and manufacturing limitations are presented with
25 electrochemichromic devices containing electrochemichromic solutions. For instance, in the case of electrochemichromic devices which contain an electrochemichromic solution within a compartment or cavity thereof, the size and shape of the
30 electrochemichromic device is limited by the bulges and non-uniformities which often form in such large area electrochemichromic devices because of the hydrostatic nature of the liquid solution. Third, from a safety standpoint, in the event an electrochemichromic device
35 should break or become damaged through fracture or rupture, it is important for the device to maintain its integrity so that, if the substrates of the device are

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shattered, an electrochromic solution does not escape therefrom and that shards of glass and the like are retained and do not scatter about. In the known electrochromic devices, measures to reduce breakage or broken glass scattering include the use of tempered glass and/or a laminate assembly comprising at least two panels affixed to one another by an adhesive. Such measures control the scattering of glass shards in the event of breakage or damage due, for instance, to the impact caused by an accident.

Numerous devices using an electrochromic medium, wherein the electrochemical reaction takes place in a solid layer, are known in the art. Typically, these devices employ electrochromic solid-state thin film technology [see e.g., N.R. Lynam, "Electrochromic Automotive Day/Night Mirrors", SAE Technical Paper Series, 870636 (1987); N.R. Lynam, "Smart Windows for Automobiles", SAE Technical Paper Series, 900419 (1990); N.R. Lynam and A. Agrawal, "Automotive Applications of Chromogenic Materials", Large Area Chromogenics: Materials & Devices for Transmittance Control, C.M. Lampert and C.G. Granquist, eds., Optical Eng'g Press, Washington (1990); C.M. Lampert, "Electrochromic Devices and Devices for Energy Efficient Windows", Solar Energy Materials, 11, 1-27 (1984); United States Patents 3,521,941 (Deb); 4,174,152 (Giglia); Re. 30,835 (Giglia); 4,338,000 (Kamimori); 4,652,090 (Uchikawa); 4,671,619 (Kamimori); Lynam I; and commonly assigned United States Patents 5,066,112 (Lynam) ("Lynam IV") and 5,076,674 (Lynam) ("Lynam V")].

In solid-state thin film electrochromic devices, an anodic electrochromic layer and a cathodic electrochromic layer, each layer usually made from inorganic metal oxides, are typically separate and

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distinct from one another and assembled in a spaced-apart relationship. The solid-state thin films are often formed using techniques such as chemical vapor deposition or physical vapor deposition. Such techniques are not attractive economically, however, as they involve cost. In another type of solid-state thin film electrochromic device, two substrates are coated separately with compositions of photo- or thermo-setting monomers or oligomers to form on one of the substrates an electrochromic layer, with the electrochromic material present within the layer being predominantly an inorganic material, and on the other substrate a redox layer. [See Japanese Patent Document JP 63-262,624].

Attempts have been made to prepare electrochromic media from polymers. For example, it has been reported that electrochromic polymer layers may be prepared by dissolving in a solvent organic polymers, which contain no functionality capable of further polymerization, together with an electrochromic compound, and thereafter casting or coating the resulting solution onto an electrode. It has been reported further that electrochromic polymer layers are created upon evaporation of the solvent by pressure reduction and/or temperature elevation. [See e.g., United States Patents 3,652,149 (Rogers), 3,774,988 (Rogers) and 3,873,185 (Rogers); 4,550,982 (Hirai); Japanese Patent Document JP 52-10,745; and Y. Hirai and C. Tani, "Electrochromism for Organic Materials in Polymeric All-Solid State Systems", Appl. Phys. Lett., 43(7), 704-05 (1983)]. Use of such polymer solution casting systems has disadvantages, however, including the need to evaporate the solvent prior to assembling devices to form polymer electrochromic layers. This additional processing step adds to the cost of manufacture through increased capital expenditures and energy requirements,

involves potential exposure to hazardous chemical vapors and constrains the type of device to be manufactured.

5 A thermally cured polymer gel film containing a single organic electrochromic compound has also been reported for use in display devices. [See H. Tsutsumi et al., "Polymer Gel Films with Simple Organic Electrochromics for Single-Film Electrochromic Devices", J. Polym. Sci., 30, 1725-29 (1992) and H. Tsutsumi et al., "Single Polymer Gel Film Electrochromic Device", Electrochimica Acta, 37, 369-70 (1992)]. The gel film reported therein was said to possess a solvent-like environment around the electrochromic compounds of that film. This gel film was reported to turn brown, and ceased to perform color-bleach cycles, after only 35,200 color-bleach cycles.

SUMMARY OF THE INVENTION

20 The present invention provides electrochromic polymeric solid films ("polychromic solid films") that are prepared by an in situ curing process different from processes used to prepare the electrochromic polymer layers known to date, and employ different combinations of electrochromic compounds than those that have been placed heretofore in solid electrochromic media. The resulting polychromic solid films possess beneficial properties and characteristics, and offer superior results, compared to the known electrochromic media. For instance, polychromic solid films overcome well-known manufacturing and use concerns such as hydrostatic pressure that is particularly troublesome in large area vertically mounted panels, such as windows, or large area mirrors, such as Class 8 truck mirrors. Thus, polychromic solid films are extremely well-suited to commercial applications, like the

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5 The present invention further provides novel processes
for making polychromic solid films by transforming such
novel electrochromic monomer compositions into
polychromic solid films through exposure to
electromagnetic radiation for a time sufficient to
10 effect an in situ cure.

The present invention still further provides electrochromic devices, such as those referred to above, particularly rearview mirrors, windows and sun roofs for automobiles, which devices are stable to outdoor weathering, particularly weathering observed due to prolonged exposure to ultraviolet radiation from the sun, and are safety protected against impact from an accident. Such outdoor weathering and safety benefits are achieved by manufacturing these devices using as a medium of varying transmittance to light the polychromic solid films prepared by the in situ cure of an electrochromic monomer composition containing a monomer component that is capable of further polymerization.

The present invention provides for the first time, among other things (1) polychromic solid films that may be transformed from electrochromic monomer compositions by an in situ curing process through exposure to electromagnetic radiation, such as ultraviolet radiation; (2) a transformation during the in situ curing process from the low viscosity, typically liquid, electrochromic monomer compositions to polychromic solid films that occurs with minimum shrinkage and with good adhesion to the contacting surfaces; (3) polychromic solid films that (a) may be

manufactured to be self-supporting and subsequently laminated between conductive substrates, (b) perform well under prolonged coloration, (c) demonstrate a resistance to degradation caused by environmental conditions, such as outdoor weathering and all-climate exposure, particularly demonstrating ultraviolet stability when exposed to the sun, and (d) demonstrate a broad spectrum of color under an applied potential; (4) polychromic solid films that may be manufactured economically and are amenable to commercial processing; (5) polychromic solid films that provide inherent safety protection not known to electrochromic media heretofore; and (6) electrochromic monomer compositions that comprise anodic electrochromic compounds and cathodic electrochromic compounds, which compounds are organic or organometallic.

The self-supporting nature of polychromic solid films provides many benefits to the electrochromic devices manufactured therewith, including the elimination of a compartmentalization means, such as a sealing means, since no such means is required to confine or contain a polychromic solid film within an electrochromic device. That polychromic solid films may be manufactured to be self-supporting also enhances processibility, and vitiates obstacles well-recognized in the manufacturing of electrochromic devices containing known electrochromic media, especially those that are to be vertically mounted in their intended use.

Moreover, since the electrochromic compounds are not free to migrate within polychromic solid films, in contrast to electrochromic compounds present within a liquid solution-phase environment, polychromic solid films do not pose the segregation concern as do solution-phase electrochromic devices; rather,

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polychromic solid films perform well under prolonged coloration.

Further, from a safety perspective, in the event that
5 electrochromic devices manufactured with polychromic
solid films should break or become damaged due to the
impact from an accident, no liquid is present to seep
therefrom since the polychromic solid films of the
present invention are indeed solid. Also, the need to
10 manufacture electrochromic devices with tempered glass,
or with at least one of the substrates being of a
laminate assembly, to reduce potential lacerative
injuries is obviated since polychromic solid films,
positioned between, and in abutting relationship with,
15 the conductive surface of the two substrates, exhibit
good adhesion to the contacting surfaces. Thus,
polychromic solid films should retain any glass shards
that may be created and prevent them from scattering.
Therefore, a safety protection feature inherent to
20 polychromic solid films is also provided herein, making
polychromic solid films particularly attractive for use
in connection with electrochromic devices, such as
mirrors, windows, sun roofs, shade bands, eye glass and
the like.

25 Polychromic solid films embody a novel and useful
technology within the electrochromic art, whose utility
will become more readily apparent and more greatly
appreciated by those of skill in the art through a
30 study of the detailed description taken in conjunction
with the figures which follow hereinafter.

BRIEF DESCRIPTION OF THE FIGURES

35 Figure 1 depicts a sectional view of an electrochromic
device employing an electrochromic polymeric solid film
according to the present invention.

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Figure 2 depicts a perspective view of an electrochromic glazing assembly according to the present invention.

- 5 The depictions in these figures are for illustrative purposes and thus are not drawn to scale.

DETAILED DESCRIPTION OF THE INVENTION

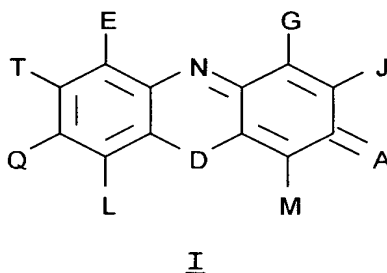
- 10 In accordance with the teaching of the present invention, polychromic solid films may be prepared by exposing an electrochromic monomer composition to electromagnetic radiation for a time sufficient to transform the electrochromic monomer composition into a
15 polychromic solid film. This in situ curing process initiates polymerization of, and typically completely polymerizes, an electrochromic monomer composition, normally in a liquid state, by exposure to electromagnetic radiation to form a polychromic solid
20 film, whose surface and cross-sections are substantially tack-free.

- The electrochromic monomer compositions are comprised of anodic electrochromic compounds, cathodic
25 electrochromic compounds, each of which are organic or organometallic compounds, a monomer component and a plasticizer. In addition, cross-linking agents, photoinitiators, photosensitizers, ultraviolet stabilizing agents, electrolytic materials, coloring
30 agents, spacers, anti-oxidizing agents, flame retarding agents, heat stabilizing agents, compatibilizing agents, adhesion promoting agents, coupling agents, humectants and lubricating agents and combinations thereof may also be added. In the preferred
35 electrochromic monomer compositions, the chosen monomer component may be a polyfunctional monomer, such as a difunctional monomer, trifunctional monomer, or a

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higher functional monomer, or a combination of monofunctional monomer and difunctional monomer or monofunctional monomer and cross-linking agent. Those of ordinary skill in the art may choose a particular monomer component or combination of monomer components from those recited in view of the intended application so as to impart the desired beneficial properties and characteristics to the polychromic solid film.

An anodic electrochromic compound suitable for use in the present invention may be selected from the class of chemical compounds represented by the following formulae:



wherein A is O, S or NRR_1 ;

wherein R and R_1 may be the same or different, and each may be selected from the group consisting of H or any straight- or branched-chain alkyl constituent having from about one carbon atom to about eight carbon atoms, such as CH_3 , CH_2CH_3 , $\text{CH}_2\text{CH}_2\text{CH}_3$, $\text{CH}(\text{CH}_3)_2$, $\text{C}(\text{CH}_3)_3$ and the like; provided that when A is NRR_1 , Q is H, OH or NRR_1 ; further provided that when A is NRR_1 , a salt may be associated therewith; still further provided that when both A and Q are NRR_1 , A and Q need not, but may, be the same functional group;

D is O, S, NR_1 or Se;

E is R_1 , COOH or CONH_2 ; or, E and T, when taken together, represent an aromatic ring

structure having six ring carbon atoms when viewed in conjunction with the ring carbon atoms to which they are attached;

G is H;

5 J is H, any straight- or branched-chain alkyl constituent having from about one carbon atom to about eight carbon atoms, such as CH_3 , CH_2CH_3 , $\text{CH}_2\text{CH}_2\text{CH}_3$, $\text{CH}(\text{CH}_3)_2$, $\text{C}(\text{CH}_3)_3$ and the like, NRR_1 , NRCR_1 , OR_1 ,

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phenyl, 2,4-dihydroxyphenyl or any halogen; or, G and J, when taken together, represent an aromatic ring structure having six ring carbon atoms when viewed in conjunction with the ring carbon atoms to which they are attached;

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L is H or OH;

M is H or any halogen;

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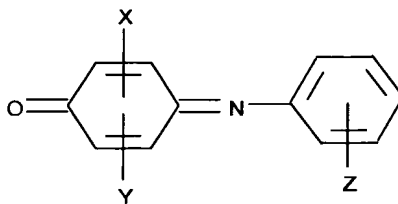
T is R_1 , phenyl or 2,4-dihydroxyphenyl; and

Q is H, OH or NRR_1 ;

provided that when L and/or Q are OH, L and/or Q may also be salts thereof; further provided that in order to render it electrochemically active in the present context, anodic electrochromic compound I has been previously contacted with a redox agent;

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II

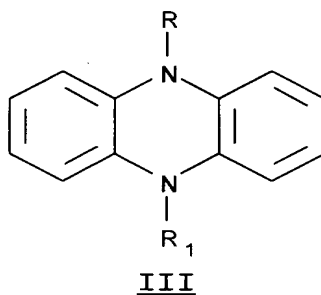
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wherein X and Y may be the same or different, and each may be selected from the group consisting of H, any halogen or NRR_1 , wherein R and R_1 may

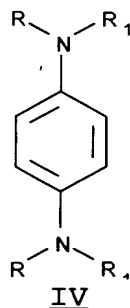
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be the same or different, and are as defined
supra; or, X and Y, when taken together,
represent an aromatic ring structure having
six ring carbon atoms when viewed in
conjunction with the ring carbon atoms to
which they are attached; and

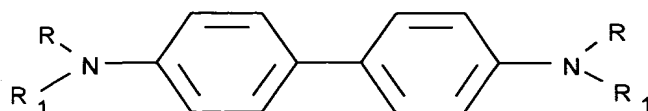
Z is OH or NRR_1 , or salts thereof;
provided that in order to render it electrochemically
active in the present context, anodic electrochromic
compound II has been previously contacted with a redox
agent;



derivatives of 5,10-dihydrophenazine
wherein R and R_1 may be the same or different, and are
defined supra;



derivatives of 1,4-phenylenediamine
wherein R and R_1 may be the same or different, and are
defined supra;

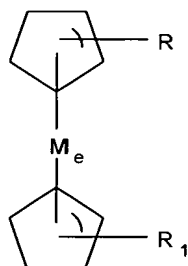


V

derivatives of benzidine

wherein R and R₁ may be the same or different, and are defined supra;

Metallocenes suitable for use as a component of the electrochromic monomer composition include, but are not limited to the following:



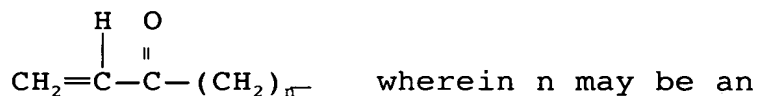
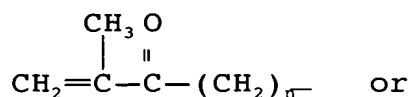
VI

metallocenes and
their derivatives

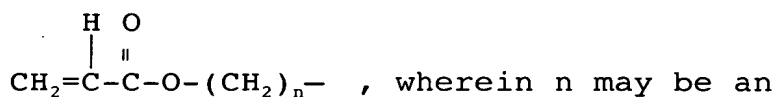
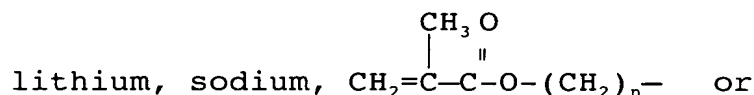
wherein R and R₁ may be the same or different, and each may be selected from the group consisting of H; any straight- or branched-chain alkyl constituent having from about 1 carbon atom to about 8 carbon atoms, such as CH₃, CH₂CH₃, CH₂CH₂CH₃, CH(CH₃)₂, C(CH₃)₃ and the like; acetyl; vinyl; allyl; -(CH₂)_n-OH, wherein n may be an integer

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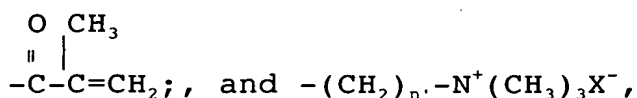
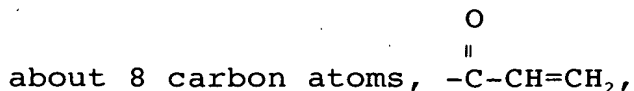
in the range of 0 to about 20;



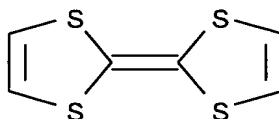
integer in the range of 0 to about 20;
 $-(\text{CH}_2)_n - \text{COOR}_2$, wherein n may be an
integer in the range of 0 to about 20
and R_2 may be any straight- or
branched-chain alkyl constituent
having from about 1 carbon atom to
about 20 carbon atoms, hydrogen,



integer from 0 to about 20,
 $-(\text{CH}_2)_{n'} - \text{OR}_3$, wherein n' may be an
integer in the range of 1 to about 12
and R_3 may be any straight- or
branched-chain alkyl constituent
having from about 1 carbon atom to



wherein n' may be an integer in the
range of 1 to about 12; X may be Cl^- ,
 Br^- , I^- , PF_6^- , ClO_4^- or BF_4^- ; and wherein
 M_e is Fe, Ni, Ru, Co, Ti, Cr, W, Mo and
the like;

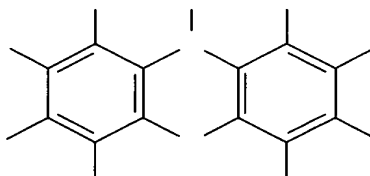


VII

tetrathiafulvalene
("TTF")

and combinations thereof.

Phenothiazines suitable for use as a component of the electrochromic monomer composition include, but are not limited to, those represented by the following structures:



phenothiazines
1-A

where R₁ may be selected from the group consisting of H; any straight- or branched-chain alkyl constituent having from about 1 carbon atom to about 10 carbon atoms; phenyl; benzyl; -(CH₂)₂-CN; -(CH₂)₂-COOH; CCH₃;;

O
||
O

allyl; -(CH₂)₂-C(=O)-O-(CH₂)_m-O-C(=O)-CH=CH₂;

-(CH₂)₂-C(=O)-O-(CH₂)_m-O-C(=O)-C(CH₃)=CH₂, wherein m' may be an

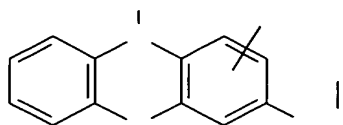
integer in the range of 1 to about 8; $-(CH_2)_2-\overset{\overset{O}{\parallel}}{C}-O-R_{18}$,

wherein R_{18} may be any straight- or branched-chain alkyl constituent having from about 1 carbon atom to about 8 carbon atoms; and

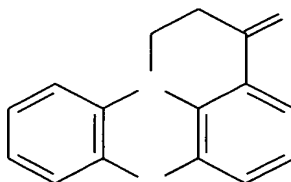
R_{10} , R_{11} , R_{12} , R_{13} , R_{14} , R_{15} , R_{16} , and R_{17} may be selected from H, Cl, Br, CF_3 , CH_3 , NO_2 , $COOH$, OH , SCH_3 , OCH_3 , O_2CCH_3 or CCH_3 ; and

R_9 and R_{17} , when taken together, form a ring with six atoms (five of which being carbon) having a carbonyl substituent on one of the carbon atoms.

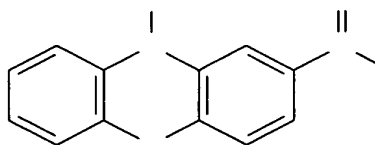
Preferred among phenothiazines 1-A is phenothiazines 2-A to 4-A as depicted in the following structure:



3-acetoxy-methyl-10H-phenothiazine
("AMPT")
2-A

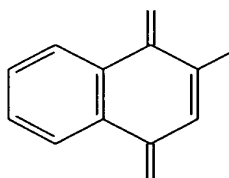


2,3-dihydro-3-keto-1H-pyrido[3,2,1-kl]phenothiazine
("C-PT")
3-A



2-acetyl-phenothiazine
("APT")
4-A

An example of a desirable quinone for use as component in the electrochromic monomer composition include, but is not limited to the following structure:



2-hydroxy-naphthoquinone
5-A

Combinations of components in the electrochromic monomer composition may be selectively chosen to achieve a desired substantially non-spectral selectivity when the electrochromic element (and the mirror in which the electrochromic element is to function) is dimmed to a colored state.

To render anodic electrochromic compounds I and II electrochemically active in the context of the present invention, a redox pre-contacting procedure must be used. Such a redox pre-contacting procedure is described in the context of preparing anodic compounds for electrochemichromic solutions in Varaprasad IV and commonly assigned co-pending United States patent application Serial No. 07/935,784 (filed August 27, 1992).

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Lacmoid (NH₂)



Resazurin



Lacmoid (OH)



Resorufin

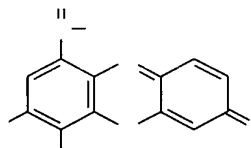


Celestine Blue



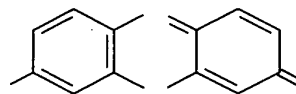
Nile Red

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XIV

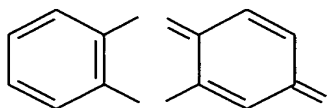
Gallocyanine



XV

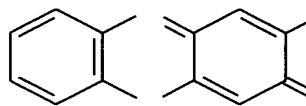
Methylene Violet
(Bernthsen) ("MVTB")

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XVI

Phenothiazine ("PT")

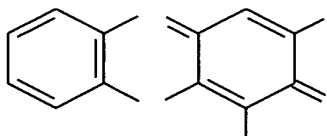


XVII

2-methyl-phenothiazin-
3-one ("MPT")

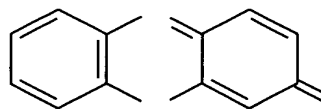
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XVIII

2-methyl-4-bromo-
phenothiazin-3-one
("BMPT")

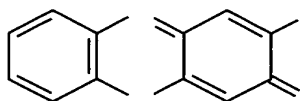


XIX

Phenoxazin-3-one
("POZ")

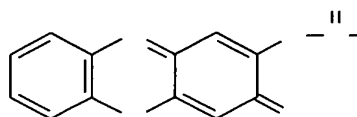
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XX

2-amino-phenoxazin-
3-one
("APOZ")



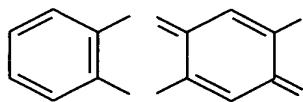
XXI

2-acetylamino-
phenoxazin-3-one
("AAPOZ")

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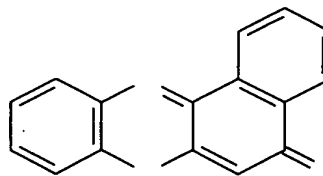
5



XXII

2-methyl-
phenoxazin-3-one
("MPOZ")

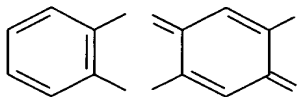
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XXIII

1,2-benzo-phen-
oxazin-3-one
("BPOZ")

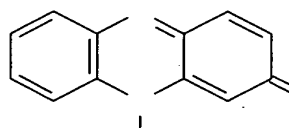
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XXIV

2-hydroxy-phenoxazin-
3-one ("HPOZ")

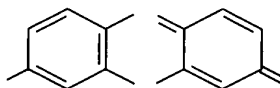
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XXV

2-keto-N-ethyl-
phenazine ("KEPA")

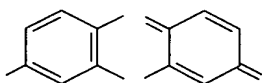
25



XXVI

Azure A
("AA")

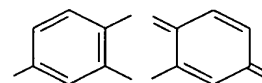
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XXVII

Azure B
("AB")

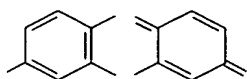
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XXVIII

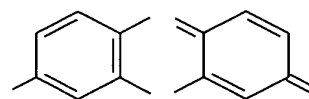
Azure C
("AC")

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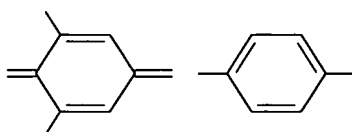
XXIX

Methylene Blue
("MB")



XXX

Thionin
("TH")

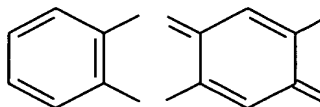


XXXI

2,6-dichloro-indophenol
("DCI")

and combinations thereof.

Among the especially preferred anodic electrochromic compounds I are MVTB (XV), PT (XVI), MPT (XVII), and POZ (XIX), with MVTB and MPT being most preferred. Also preferred is the reduced form of MPT which results from the redox pre-contacting procedure referred to above, and has been thereafter isolated. This reduced and isolated form of MPT -- RMPT [XVII(a)] -- is believed to be 2-methyl-3-hydroxyphenathiazine, which is represented by the following chemical formula



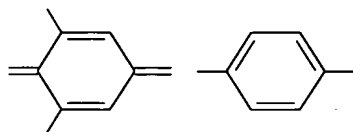
XVII(a)

("RMPT")

and salts thereof.

In addition, a preferred anodic electrochromic compound II is

5



10

XXXI

2,6-dichloro-indophenol
("DCI")

and salts thereof.

15

Likewise, preferred among anodic electrochromic compound III are 5,10-dihydro-5,10-dimethylphenazine ("DMPA") and 5,10-dihydro-5,10-diethylphenazine ("DEPA"), with DMPA being particularly preferred.

20

As a preferred anodic electrochromic compound VI, metallocenes, such as ferrocene, wherein M_e is iron and R and R_1 are each hydrogen, and alkyl derivatives thereof, may also be used advantageously in the context of the present invention.

25

The salts referred to in connection with the anodic electrochromic compounds include, but are not limited to, alkali metal salts, such as lithium, sodium, potassium and the like. In addition, when A is NRR_1 , tetrafluoroborate (" BF_4^- "), perchlorate (" ClO_4^- "), trifluoromethane sulfonate (" $CF_3SO_3^-$ "), hexafluorophosphate (" PF_6^- "), acetate (" Ac^- ") and any halogen may be associated therewith. Moreover, the ring nitrogen atom in anodic electrochromic compound I may also appear as an N-oxide.

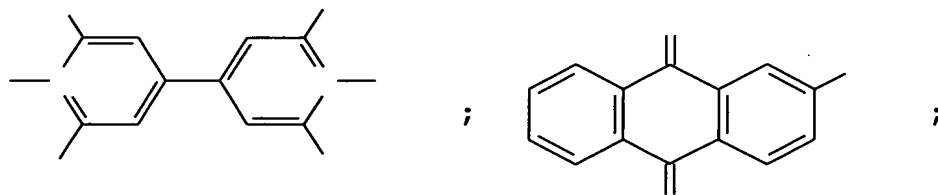
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Any one or more of anodic electrochromic compounds I,
II, III, IV, V, VI or VII may also be advantageously
combined, in any proportion, within an electrochromic
monomer composition and thereafter transformed into a
5 polychromic solid film to achieve the results so stated
herein. Of course, as regards anodic electrochromic
compounds I and II, it is necessary to contact those
compounds with a redox agent prior to use so as to
render them electrochemically active in the present
10 invention. Upon the application of a potential
thereto, such combinations of anodic electrochromic
compounds within a polychromic solid film may often
generate color distinct from the color observed from
polychromic solid films containing individual anodic
15 electrochromic compounds. A preferred combination of
anodic electrochromic compounds in this invention is
the combination of anodic electrochromic compounds III
and VI. Nonetheless, those of ordinary skill in the
art may make appropriate choices among individual
20 anodic electrochromic compounds and combinations
thereof, to prepare a polychromic solid film capable of
generating a color suitable for a particular
application.

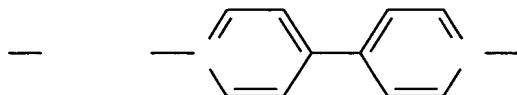
25 A choice of a cathodic electrochromic compound for use
herein should also be made. The cathodic
electrochromic compound may be selected from the class
of chemical compounds represented by the following
formulae:



XXXII
derivatives
of viologen

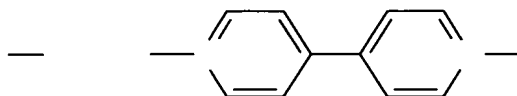
XXXIII
derivatives
of anthraquinone

5



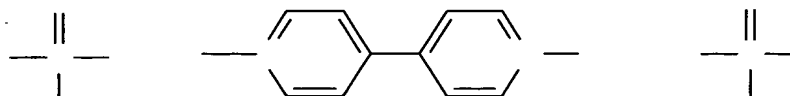
6-A
Dihydroxyalkyl Viologen Salt
("DHAVS")

10



7-A
Monohydroxy Viologen Salt
("MHVS")

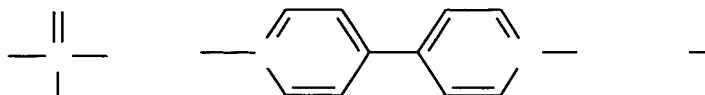
15



20

8-A
Diphosphoryldihydroxyalkyl Viologen Salt
("DPDHAVS")

25



30

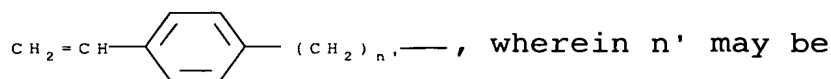
9-A
Phosphoryldihydroxyalkyl Viologen Salt
("PDHAVS")

wherein R_3 , R_4 , R_{21} , R_{22} , R_{23} and R_{24} may be the same or
different and each may be selected from the
group consisting of H, any straight- or
branched-chain alkyl constituent having from

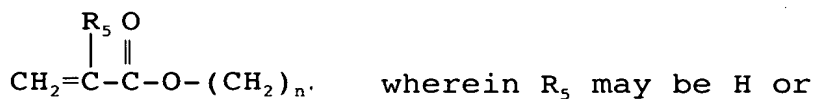
35

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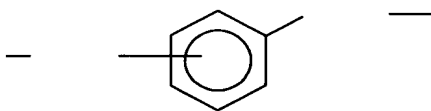
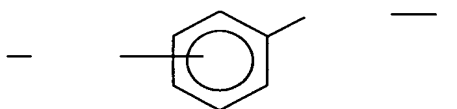
about one carbon atom to about eight carbon atoms, or any straight- or branched-chain alkyl- or alkoxy-phenyl, wherein the alkyl or alkoxy constituent contains from about one carbon atom to about eight carbon atoms;



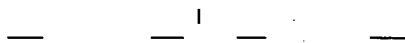
an integer in the range of 1 to 12;



CH_3 , and n' may be an integer in the range of 1 to 12; $\text{HO}-(\text{CH}_2)_{n'}-\text{---}$, wherein n' may be an integer in the range of 1 to 12; and $\text{HOOC}-(\text{CH}_2)_{n'}-\text{---}$, wherein n' may be an integer in the range of 1 to 12;



and

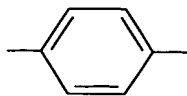


wherein q may be an integer in the range of 0 to 12; wherein each p is independently an integer from 1 to 12; and wherein X is selected from the group consisting of BF_4^- , ClO_4^- , CF_3SO_3^- , styrylsulfonate ("SS⁻"), 2-acrylamido-2-methylpropane-sulfonate, acrylate, methacrylate, 3-

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sulfopropylacrylate, 3-sulfopropyl-
methacrylate, PF_6^- , Ac^- , $\text{HO}-(\text{R}_{25})-\text{SO}_3^-$ and
 $\text{HOOC}-(\text{R}_{25})-\text{SO}_3^-$ wherein R_{25} can be any
straight- or branched-chain alkyl
constituent having from about 1 carbon atom
to about 8 carbon atoms, an aryl or a
functionalized aryl, an alkyl or aryl amide,
a branched or linear chain polymer, such as
polyvinyls, polyethers and polyesters
bearing at least one and preferably
multiple, hydroxyl and sulphonate
functionalities and any halide; and
combinations thereof.

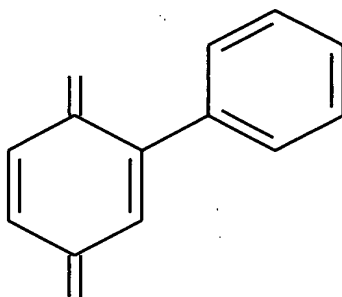
In one preferred embodiment R_{25} can be:



4-hydroxyphenylsulfonate (or its isomers)

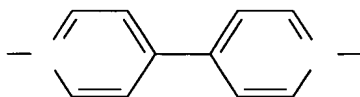
or the copolymer derived from acrylamidomethyl-
propanesulfonic acid (AMPS) and caprolactone acrylate.

Specific cathodic electrochromic compounds useful in
the context of the present invention include:



XXXIV

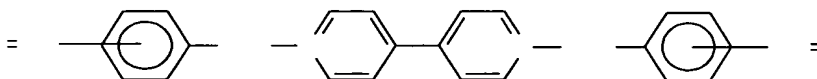
Phenyl-1,4-benzoquinone ;



;

XXXV

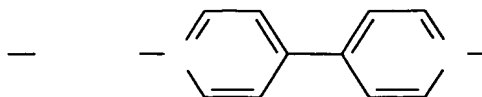
Heptylviologen
Styrylsulfonate
("HVSS")



;

XXXVI

Distyrylmethylviologen
Perchlorate
(mixed isomers)
("DSMVC10₄")



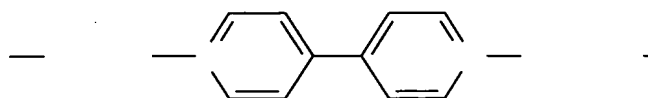
5

;

XXXVII

Ethylhydroxypropylviologen
Perchlorate
("EHPVClO₄")

10

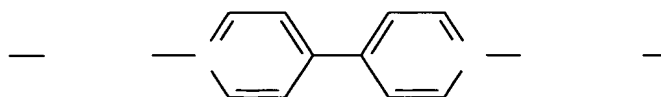


;

XXXVIII

Hydroxyundecyl Viologen Perchlorate
("HUVClO₄")

15



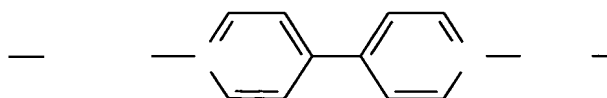
20

;

XXXIX

Hydroxyundecyl Viologen Hexafluorophosphate
("HUVPF₆")

25

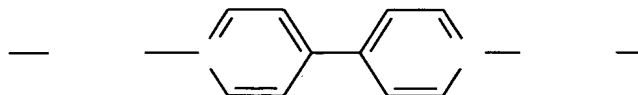


;

XXXX

Ethylhydroxyundecyl Viologen Perchlorate
("EHUVClO₄")

30



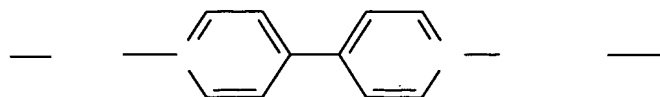
;

XXXXI

Hydroxyhexyl Viologen Hexafluorophosphate
("HHVPF₆")

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5

XXXXII
Divalericacid Viologen Hexafluorophosphate
("DVAVPF₆")

10

Preferably, R₃ and R₄ are ethyl, *n*-heptyl, hydroxyhexyl or hydroxyundecyl. Thus, when X is PF₆⁻, ClO₄⁻ or BF₄⁻, preferred cathodic electrochromic compounds are ethylviologen perchlorate ("EVClO₄"), heptylviologen tetrafluoroborate ("HVBF₄"), hydroxyundecyl viologen hexafluorophosphate ("HUVPF₆"), ethylhydroxyundecyl viologen perchlorate ("EHUVCLO₄"), hydroxyhexyl viologen hexafluorophosphate ("HHVPF₆"), and divalericacid viologen hexafluorophosphate ("DVAVPF₆").

15

20

The above anodic electrochromic compounds and cathodic electrochromic compounds may be chosen so as to achieve a desired color when the polychromic solid film in which they are present (and the device in which the polychromic solid film is contained) is colored to a dimmed state. For example, electrochromic automotive mirrors manufactured with polychromic solid films should preferably bear a blue or substantially neutral color when colored to a dimmed state. And, electrochromic optically attenuating contrast filters, such as contrast enhancement filters, manufactured with polychromic solid films should preferably bear a substantially neutral color when colored to a dimmed state.

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The plasticizer chosen for use in the present invention should maintain the homogeneity of the electrochromic monomer compositions while being prepared, used and stored, and prior to, during and after exposure to electromagnetic radiation. As a result of its

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combination within the electrochromic monomer composition or its exposure to electromagnetic radiation, the plasticizer of choice should not form by-products that are capable of hindering, or
5 interfering with, the homogeneity and the electrochemical efficacy of the resulting polychromic solid film. The occurrence of any of these undesirable events during the in situ curing process, whether at the pre-cure, cure or post-cure phase of the process
10 for preparing polychromic solid films, may interfere with the process itself, and may affect the appearance and effectiveness of the resulting polychromic solid films, and the electrochromic devices manufactured with the same. The plasticizer also may play a role in
15 defining the physical properties and characteristics of the polychromic solid films of the present invention, such as toughness, flex modulus, coefficient of thermal expansion, elasticity, elongation and the like.

20 Suitable plasticizers for use in the present invention include, but are not limited to, triglyme, tetraglyme, acetonitrile, benzylacetone, 3-hydroxypropionitrile, methoxypropionitrile, 3-ethoxypropionitrile, butylene carbonate, propylene carbonate, ethylene carbonate,
25 glycerine carbonate, 2-acetylbutyrolactone, cyanoethyl sucrose, γ -butyrolactone, 2-methylglutaronitrile, N,N'-dimethylformamide, 3-methylsulfolane, methylethyl ketone, cyclopentanone, cyclohexanone, 4-hydroxy-4-methyl-2-pentanone, acetophenone, glutaronitrile,
30 3,3'-oxydipropionitrile, 2-methoxyethyl ether, triethylene glycol dimethyl ether and combinations thereof. Particularly preferred plasticizers among that group are benzylacetone, 3-hydroxypropionitrile, propylene carbonate, ethylene carbonate, 2-acetylbutyrolactone, cyanoethyl sucrose, triethylene
35 glycol dimethyl ether, 3-methylsulfolane and combinations thereof.

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To prepare a polychromic solid film, a monomer should be chosen as a monomer component that is capable of in situ curing through exposure to electromagnetic radiation, and that is compatible with the other components of the electrochromic monomer composition at the various stages of the in situ curing process. The combination of a plasticizer with a monomer component (with or without the addition of a difunctional monomer or a cross-linking agent) should preferably be in an equivalent ratio of between about 75:25 to about 10:90 to prepare polychromic solid films with superior properties and characteristics. Of course, the art-skilled should bear in mind that the intended application of a polychromic solid film will often dictate its particular properties and characteristics, and that the choice and equivalent ratio of the components within the electrochromic monomer composition may need to be varied to attain a polychromic solid film with the desired properties and characteristics.

Among the monomer components that may be advantageously employed in the present invention are monomers having at least one reactive functionality rendering the compound capable of polymerization or further polymerization by an addition mechanism, such as vinyl polymerization or ring opening polymerization. Included among such monomers are oligomers and polymers that are capable of further polymerization. For monomers suitable for use herein, see generally those commercially available from Monomer-Polymer Labs., Inc., Philadelphia, Pennsylvania; Sartomer Co., Exton, Pennsylvania; and Polysciences, Inc., Warrington, Pennsylvania.

Monomers capable of vinyl polymerization, suitable for use herein, have as a commonality the ethylene functionality, as represented below:



wherein R_6 , R_7 , and R_8 may be the same or different, and are each selected from a member of the group consisting of hydrogen; halogen; alkyl, cycloalkyl, poly-cycloalkyl, heterocycloalkyl and alkyl and alkenyl derivatives thereof; alkenyl, cycloalkenyl, cycloalkadienyl, poly-cycloalkadienyl and alkyl and alkenyl derivatives thereof; hydroxyalkyl; hydroxyalkenyl; alkoxyalkyl; alkoxyalkenyl; cyano; amido; phenyl; benzyl and carboxylate, and derivatives thereof.

Preferred among these vinyl monomers are the ethylene carboxylate derivatives known as acrylates -- i.e., wherein at least one of R_6 , R_7 , and R_8 are carboxylate groups or derivatives thereof. Suitable carboxylate derivatives include, but are not limited to alkyl, cycloalkyl, poly-cycloalkyl, heterocycloalkyl and alkyl and alkenyl derivatives thereof; alkenyl, cycloalkenyl, poly-cycloalkenyl and alkyl and alkenyl derivatives thereof; mono- and poly-hydroxyalkyl; mono- and poly-hydroxyalkenyl; alkoxyalkyl; alkoxyalkenyl and cyano.

Among the acrylates that may be advantageously employed herein are mono- and poly-acrylates (bearing in mind that poly-acrylates function as cross-linking agents as well, see *infra*), such as 2-hydroxyethyl acrylate, 2-hydroxyethyl methacrylate, methylene glycol monoacrylate, diethylene glycol monomethacrylate, 2-hydroxypropyl acrylate, 2-hydroxypropyl methacrylate, 3-hydroxypropyl acrylate, 3-hydroxypropyl methacrylate,

dipropylene glycol monomethacrylate, 2,3-
dihydroxypropyl methacrylate, methyl acrylate, ethyl
acrylate, *n*-propyl acrylate, *i*-propyl acrylate, *n*-butyl
acrylate, *s*-butyl acrylate, *n*-pentyl acrylate, 2-
5 ethylhexyl acrylate, methyl methacrylate, ethyl
methacrylate, *n*-propyl methacrylate, *i*-propyl
methacrylate, *n*-butyl methacrylate, *s*-butyl
methacrylate, *n*-pentyl methacrylate, *s*-pentyl
methacrylate, methoxyethyl acrylate, methoxyethyl
10 methacrylate, triethylene glycol monoacrylate, glycerol
monoacrylate, glycerol monomethacrylate, allyl
methacrylate, benzyl acrylate, caprolactone acrylate,
cyclohexyl acrylate, cyclohexyl methacrylate,
2-ethoxyethyl acrylate, 2-ethoxyethyl methacrylate, 2-
15 (2-ethoxyethoxy)ethylacrylate, glycidyl methacrylate,
n-hexyl acrylate, *n*-hexyl methacrylate, isobornyl
acrylate, isobornyl methacrylate, *i*-decyl acrylate, *i*-
decyl methacrylate, *i*-octyl acrylate, lauryl acrylate,
lauryl methacrylate, 2-methoxyethyl acrylate, *n*-octyl
20 acrylate, 2-phenoxyethyl acrylate, 2-phenoxyethyl
methacrylate, stearyl acrylate, stearyl methacrylate,
tetrahydrofurfuryl acrylate, tetrahydrofurfuryl
methacrylate, tridecyl methacrylate, 1,4-butanediol
diacrylate, 1,4-butanediol dimethacrylate, 1,3-butylene
25 glycol diacrylate, 1,3-butylene glycol dimethacrylate,
diethylene glycol diacrylate, diethylene glycol
dimethacrylate, ethylene glycol diacrylate, ethoxylated
bisphenol A dimethacrylate, ethylene glycol
dimethacrylate, 1,6-hexanediol diacrylate, 1,6-
30 hexanediol dimethacrylate, neopentyl glycol diacrylate,
neopentyl glycol dimethacrylate, polyethylene glycol
diacrylate, polyethylene glycol dimethacrylate,
tetraethylene glycol diacrylate, tetraethylene glycol
dimethacrylate, triethylene glycol diacrylate,
35 triethylene glycol dimethacrylate, tripropylene glycol
diacrylate, dipentaerythritol pentaacrylate,
ethoxylated pentaerythritol triacrylate,

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pentaerythritol tetraacrylate, pentaerythritol triacrylate, trimethylolpropane triacrylate, trimethylolpropane trimethacrylate, tris(2-hydroxyethyl)-isocyanurate triacrylate, 5 tris(2-hydroxyethyl)-isocyanurate trimethacrylate, polyethylene glycol monoacrylate, polyethylene glycol monomethacrylate, polypropylene glycol monoacrylate, polypropylene glycol monomethacrylate, hydroxyethyl cellulose acrylate, hydroxyethyl cellulose 10 methacrylate, methoxy poly(ethyleneoxy) ethylacrylate, methoxy poly(ethyleneoxy) ethylmethacrylate and combinations thereof. For a further recitation of suitable acrylates for use herein, see those acrylates available commercially from Monomer-Polymer Labs, Inc.; 15 Polysciences, Inc. and Sartomer Co. Also, those of ordinary skill in the art will appreciate that derivatized acrylates in general should provide beneficial properties and characteristics to the resulting polychromic solid film.

20 Other monomers suitable for use herein include styrenes, unsaturated polyesters, vinyl ethers, acrylamides, methyl acrylamides and the like.

25 Other monomers capable of addition polymerization include isocyanates, polyols, amines, polyamines, amides, polyamides, acids, polyacids, compounds comprising an active methylene group, ureas, thiols, etc. Preferably, such monomers have a functionality of 30 2 or greater. For example, the monomer composition can include isocyanates such as hexamethylene diisocyanate (HDI); toluene diisocyanate (TDI including 2, 4 and 2, 6 isomers); diphenylmethane diisocyanate (MDI); isocyanate tipped prepolymers such as those prepared 35 from a diisocyanate and a polyol; condensates produced from hexamethylene diisocyanate including biuret type and trimer type (also known as isocyanurate), as is

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known in the urethane chemical art. A recitation of various monomers suitable to use in the electrochromic monomer composition is given in the following Table 1.

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Type	Trade Name	Product No:	Supplier	Location
Isocyanate	Tolonate	HDT (Isocyanurate)	Rhone-Poulenc Inc.	Princeton, NJ
Isocyanate	Tolonate	HDB (Biuret)	Rhone-Poulenc Inc.	Princeton, NJ
Isocyanate	ISONATE	modified MDI	Dow Chemical	Midland, MI
Isocyanate	PAPI	polymeric MDI	Dow Chemical	Midland, MI
Isocyanate	RUBINATE	9043 MDI	ICI	Sterling Heights, MI
Isocyanate	DESMODUR	N-100	Miles	Pittsburgh, PA
Isocyanate	TYCEL	7351	Liofol Co.	Cary, NC
Polyol	VORANOL	polyether polyols	Dow Chemical	Midland, MI
Polyol	VORANOL	copolymer polyols	Dow Chemical	Midland, MI
Polyol	ARCOL	E-786	Arco Chemical	Hinsdale, IL
Polyol	ARCOL	LHT-112	Arco Chemical	Hinsdale, IL
Polyol	ARCOL	E-351	Arco Chemical	Hinsdale, IL
Polyol	LEXOREZ	1931-50	Inolex Chemical Co.	Philadelphia, PA
Polyol	LEXOREZ	1842-90	Inolex Chemical Co.	Philadelphia, PA
Polyol	LEXOREZ	1405-65	Inolex Chemical Co.	Philadelphia, PA
Polyol	LEXOREZ	1150-110	Inolex Chemical Co.	Philadelphia, PA
Polyol	DESMOPHEN	1700	Miles	Pittsburgh, PA
Tin Catalyst	DABCO	T-9	Air Products and Chemical Inc.	Allentown, PA
Tin Catalyst	DABCO	T-1	Air Products and Chemical Inc.	Allentown, PA
Tin Catalyst	DABCO	T-120	Air Products and Chemical Inc.	Allentown, PA

Table 1. Monomers suitable to use in the electrochromic monomer composition

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D.E.R. 736 epoxy resin (epichlorohydrin-polyglycol reaction product), D.E.R. 755 epoxy resin (diglycidyl ether of bisphenol A-diglycidyl ether of polyglycol) and D.E.R. 732 epoxy resin (epichlorohydrin-polyglycol reaction product), and the NOVOLAC epoxy resins such as D.E.N. 431, D.E.N. 438 and D.E.N. 439 (phenolic epoxides), and those epoxides commercially available from Shell Chemical Co., Oak Brook, Illinois, like the "EPON" resins 825 and 1001F (epichlorohydrin-bisphenol A type epoxy resins).

Other commercially available epoxide monomers that are particularly well-suited for use herein include those commercially available under the "ENVIBAR" tradename from Union Carbide Chemicals and Plastics Co., Inc., Danbury, Connecticut, such as "ENVIBAR" UV 1244 (cycloalkyl epoxides).

In addition, derivatized urethanes, such as acrylated (e.g., mono- or poly-acrylated) urethanes; derivatized heterocycles, such as acrylated (e.g., mono- or poly-acrylated) heterocycles, like acrylated epoxides, acrylated lactones, acrylated lactams; and combinations thereof, capable of undergoing addition polymerizations, such as vinyl polymerizations and ring opening polymerizations, are also well-suited for use herein.

Many commercially available ultraviolet curable formulations are well-suited for use herein as a monomer component in the electrochromic monomer composition. Among those commercially available ultraviolet curable formulations are acrylated urethanes, such as the acrylated alkyl urethane formulations commercially available from Sartomer Co., including Low Viscosity Urethane Acrylate (Flexible) (CN 965), Low Viscosity Urethane Acrylate (Resilient)

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(CN 964), Urethane Acrylate (CN 980), Urethane Acrylate/TPGDA (CN 966 A80), Urethane Acrylate/IBOA (CN 966 J75), Urethane Acrylate/EOEOEA (CN 966 H90), Urethane Acrylate/TPGDA (CN 965 A80), Urethane Acrylate/EOTMPTA (CN 964 E75), Urethane Acrylate/EOEOEA (CN 966 H90), Urethane Acrylate/TPGDA (CN 963 A80), Urethane Acrylate/EOTMPTA (CN 963 E75), Urethane Acrylate (Flexible) (CN 962), Urethane Acrylate/EOTMPTA (CN 961 E75), Urethane Acrylate/EOEOEA (CN 961 H90), Urethane Acrylate (Hard) (CN 955), Urethane Acrylate (Hard) (CN 960) and Urethane Acrylate (Soft) (CN 953), and acrylated aromatic urethane formulations, such as those sold by Sartomer Co., may also be used herein, including Hydrophobic Urethane Methacrylate (CN 974), Urethane Acrylate/TPGDA (CN 973 A80), Urethane Acrylate/IBOA (CN 973 J75), Urethane Acrylate/EOEOEA (CN 973 H90), Urethane Acrylate (Flexible) (CN 972), Urethane Acrylate (Resilient) (CN 971), Urethane Acrylate/TPGDA (CN 971 A80), Urethane Acrylate/TPGDA (CN 970 A60), Urethane Acrylate/EOTMPTA (CN 970 E60) and Urethane Acrylate/EOEOEA (CN 974 H75). Other acrylated urethane formulations suitable for use herein may be obtained commercially from Monomer-Polymer Labs, Inc. and Polysciences, Inc.

Other ultraviolet curable formulations that may be used herein are the ultraviolet curable acrylated epoxide formulations commercially available from Sartomer Co., such as Epoxidized Soy Bean Oil Acrylate (CN 111), Epoxy Acrylate (CN 120), Epoxy Acrylate/TPGDA (CN 120 A75), Epoxy Acrylate/HDDA (CN 120 B80), Epoxy Acrylate/TMPTA (CN 120 C80), Epoxy Acrylate/GPTA (CN 120 D80), Epoxy Acrylate/Styrene (CN 120 S85), Epoxy Acrylate (CN 104), Epoxy Acrylate/GPTA (CN 104 D80), Epoxy Acrylate/HDDA (CN 104 B80), Epoxy Acrylate/TPGDA (CN 104 A80), Epoxy Acrylate/TMPTA (CN 104 C75), Epoxy Novolac Acrylate/TMPTA (CN 112 C60), Low Viscosity

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Epoxy Acrylate (CN 114), Low Viscosity Epoxy Acrylate/EOTMPTA (CN 114 E80), Low Viscosity Epoxy Acrylate/GPTA (CN 114 D75) and Low Viscosity Epoxy Acrylate/TPGDA (CN 114 A80).

5 In addition, "SARBOX" acrylate resins, commercially available from Sartomer Co., like Carboxylated Acid Terminated (SB 400), Carboxylated Acid Terminated (SB 401), Carboxylated Acid Terminated (SB 500),
10 Carboxylated Acid Terminated (SB 500E50), Carboxylated Acid Terminated (SB 500K60), Carboxylated Acid Terminated (SB 501), Carboxylated Acid Terminated (SB 510E35), Carboxylated Acid Terminated (SB 520E35) and Carboxylated Acid Terminated (SB 600) may also be advantageously employed herein.

15 Also well-suited for use herein are ultraviolet curable formulations like the ultraviolet curable conformational coating formulations commercially available under the "QUICK CURE" trademark from the
20 Specialty Coating Systems subsidiary of Union Carbide Chemicals & Plastics Technology Corp., Indianapolis, Indiana, and sold under the product designations B-565, B-566, B-576 and BT-5376; ultraviolet curing adhesive formulations commercially available from Loctite Corp.,
25 Newington, Connecticut under the product names UV OPTICALLY CLEAR ADH, MULTI PURPOSE UV ADHESIVE, "IMPRUV" LV POTTING COMPOUND and "LOCQUIC" ACTIVATOR 707; ultraviolet curable urethane formulations commercially available from Norland Products, Inc., New
30 Brunswick, New Jersey, and sold under the product designations "NORLAND NOA 61", "NORLAND NOA 65" and "NORLAND NOA 68"; and ultraviolet curable acrylic formulations commercially available from Dymax Corp., Torrington, Connecticut, including "DYMAX LIGHT-WELD
35 478".

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By employing polyfunctional monomers, like difunctional monomers, or cross-linking agents, cross-linked polychromic solid films may be advantageously prepared. Such cross-linking tends to improve the physical properties and characteristics (e.g., mechanical strength) of the resulting polychromic solid films. Cross-linking during cure to transform the electrochromic monomer composition into a polychromic solid film may be achieved by means of free radical ionic initiation by the exposure to electromagnetic radiation. This may be accomplished by combining together all the components of the particular electrochromic monomer composition and thereafter effecting cure. Alternatively, cross-links may be achieved by exposing to electromagnetic radiation the electrochromic monomer composition for a time sufficient to effect a partial cure, whereupon further electromagnetic radiation and/or a thermal influence may be employed to effect a more complete in situ cure and transformation into the polychromic solid film.

Suitable polyfunctional monomers for use in preparing polychromic films should have at least two reactive functionalities, and may be selected from, among others, ethylene glycol diacrylate, ethylene glycol dimethacrylate, 1,2-butylene dimethacrylate, 1,3-butylene dimethacrylate, 1,4-butylene dimethacrylate, propylene glycol diacrylate, propylene glycol dimethacrylate, diethylene glycol diacrylate, dipropylene glycol diacrylate, divinyl benzene, divinyl toluene, diallyl tartrate, allyl maleate, divinyl tartrate, triallyl melamine, glycerine trimethacrylate, diallyl maleate, divinyl ether, diallyl monomethylene glycol citrate, ethylene glycol vinyl allyl citrate, allyl vinyl maleate, diallyl itaconate, ethylene glycol diester of itaconic acid, polyester of maleic anhydride with triethylene glycol, polyallyl glucoses (e.g.,

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covalent bonding, in a polymer backbone. The SORBALITE™ clear thin polymer film when placed on a surface of the substrate closest to the source of UV radiation (such as the sun), efficiently absorbs UV light below about 370 nm with minimal effect on the visible region. Thickness of the SORBALITE™ film is desirably in the range of about 0.1 microns to 1000 microns (or thicker); preferably less than 100 microns; more preferably less than about 25 microns, and most preferably less than about 10 microns. Also, UV absorbing thin films or additives such as cerium oxide, iron oxide, nickel oxide and titanium oxide or such oxides with dopants can be used to protect the electrochromic device from UV degradation. Further as described above, UV absorbing chromophores can be incorporated, such as by covalent bonding, into the solid polymer matrix to impart enhanced resilience to UV radiation. Also near-infrared radiation absorbing species may be incorporated into the solid polymer matrix.

The density of the cross-link within the resulting polychromic solid film tends to increase with the amount and/or the degree of functionality of polyfunctional monomer present in the electrochromic monomer composition. Cross-linking density within a polychromic solid film may be achieved or further increased by adding to the electrochromic monomer composition cross-linking agents, which themselves are incapable of undergoing further polymerization. In addition to increasing the degree of cross-linking within the resulting polychromic solid film, the use of such cross-linking agents in the electrochromic monomer composition may enhance the prolonged coloration performance of the resulting polychromic solid film. Included among such cross-linking agents are polyfunctional hydroxy compounds, such as glycols and

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glycerol, polyfunctional primary or secondary amino compounds and polyfunctional mercapto compounds. Among the preferred cross-linking agents are pentaerythritol, 2-ethyl-2-(hydroxymethyl)-1,3-propanediol, the poly (caprolactone) diols having molecular weights of 1,250, 2,000 and 3,000, and polycarbonate diol available from Polysciences, Inc. and the polyfunctional hydroxy compounds commercially available under the "TONE" tradename from Union Carbide Chemicals and Plastics Co. Inc., Danbury, Connecticut, such as ϵ -caprolactone triols (known as "TONE" 0301, "TONE" 0305 and "TONE" 0310). Among the preferred glycols are the poly(ethylene glycols), like those sold under the "CARBOWAX" tradename by the Industrial Chemical division of Union Carbide Corp., Danbury, Connecticut such as "CARBOWAX" PEG 200, PEG 300, PEG 400, PEG 540 Blend, PEG 600, PEG 900, PEG 1000, PEG 1450, PEG 3350, PEG 4600, and PEG 8000, with "CARBOWAX" PEG 1450 being the most preferred among this group, and those available from Polysciences, Inc.

Polychromic solid films that perform well under prolonged coloration may be prepared from electrochromic monomer compositions that contain as a monomer component at least some portion of a polyfunctional monomer -- e.g., a difunctional monomer. By preferably using polyfunctional monomers having their functional groups spaced apart to such an extent so as to enhance the flexibility of the resulting polychromic solid film, polychromic solid films may be prepared with a minimum of shrinkage during the transformation process and that also perform well under prolonged coloration.

While it is preferable to have electrochromic monomer compositions which contain a monomer component having polyfunctionality in preparing polychromic solid films

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that perform well under prolonged coloration, electrochromic monomer compositions that exhibit enhanced resistance to shrinkage when transformed into polychromic solid films preferably contain certain monofunctional monomers. In this regard, depending on the specific application, some physical properties and characteristics of polychromic solid films may be deemed of greater import than others. Thus, superior performance in terms of resistance to shrinkage during in situ curing of the electrochromic monomer composition to the polychromic solid film may be balanced with the prolonged coloration performance of the resulting polychromic solid film to achieve the properties and characteristics desirable of that polychromic solid film.

Those of ordinary skill in the art may make appropriate choices among the herein described monomers -- monofunctional and polyfunctional, such as difunctional -- and cross-linking agents to prepare a polychromic solid film having beneficial properties and characteristics for the specific application by choosing such combinations of a monofunctional monomer to a polyfunctional monomer or a monofunctional monomer to a cross-linking agent in an equivalent ratio of about 1:1 or greater.

In the preferred electrochromic monomer compositions, photoinitiators or photosensitizers may also be added to assist the initiation of the in situ curing process. Such photoinitiators or photosensitizers enhance the rapidity of the curing process when the electrochromic monomer compositions are exposed to electromagnetic radiation. These materials include, but are not limited to, radical initiation type and cationic initiation type polymerization initiators such as benzoin derivatives, like the *n*-butyl, *i*-butyl and

ethyl benzoin alkyl ethers, and those commercially available products sold under the "ESACURE" tradename by Sartomer Co., such as "ESACURE" TZT (trimethyl benzophenone blend), KB1 (benzildimethyl ketal), KB60 (60% solution of benzildimethyl ketal), EB3 (mixture of benzoin *n*-butyl ethers), KIP 100F (α -hydroxy ketone), KT37 (TZT and α -hydroxy ketone blend), ITX (*i*-propylthioxanthone), X15 (ITX and TZT blend), and EDB [ethyl-4-(dimethylamino)-benzoate]; those commercially available products sold under the "IRGACURE" and "DAROCURE" tradenames by Ciba Geigy Corp., Hawthorne, New York, specifically "IRGACURE" 184, 907, 369, 500, 651, 261, 784 and "DAROCURE" 1173 and 4265, respectively; the photoinitiators commercially available from Union Carbide Chemicals and Plastics Co. Inc., Danbury, Connecticut, under the "CYRACURE" tradename, such as "CYRACURE" UVI-6974 (mixed triaryl sulfonium hexafluoroantimonate salts) and UVI-6990 (mixed triaryl sulfonium hexafluorophosphate salts); and the visible light [blue] photoinitiator, *dl*-camphorquinone.

Of course, when those of ordinary skill in the art choose a commercially available ultraviolet curable formulation, it may no longer be desirable to include as a component within the electrochromic monomer composition an additional monomer to that monomer component already present in the commercial formulation. And, as many of such commercially available ultraviolet curable formulations contain a photoinitiator or photosensitizer, it may no longer be desirable to include this optional component in the electrochromic monomer composition. Nevertheless, a monomer, or a photoinitiator or a photosensitizer, may still be added to the electrochromic monomer composition to achieve beneficial results, and particularly when specific properties and

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characteristics are desired of the resulting polychromic solid film.

5 With an eye toward maintaining the homogeneity of the electrochromic monomer composition and the polychromic solid film which results after in situ cure, those of ordinary skill in the art should choose the particular components dispersed throughout, and their relative quantities, appropriately. One or more compatibilizing agents may be optionally added to the electrochromic monomer composition so as to accomplish this goal. Such compatibilizing agents include, among others, combinations of plasticizers recited herein, a monomer component having polyfunctionality and cross-linking agents that provide flexible cross-links. See supra.

10 Further, monomer compositions can be formed comprising both organic and inorganic monomers. For example, inorganic monomers such as tetraethylorthosilicate, titanium isopropoxide, metal alkoxides, and the like may be included in the monomer composition, and formation of the solid matrix (be it an organic polymer matrix, an inorganic polymer matrix or an organic/inorganic polymer matrix) can proceed via a variety of reaction mechanisms, including hydrolysis/condensation reactions. Also, transition metal-peroxy acid products (such as tungsten peroxy acid product) can be reacted with alcohol to form a peroxy-transition metal derivative (such as peroxytungstic ester derivative), with a recitation of such species being found in US Patent 5,457,218 entitled "Precursor and Related Method of Forming Electrochromic Coatings", invented by J. Cronin et al and issued October 10, 1995, the disclosure of which is hereby incorporated by reference herein, and can be used as a component of the electrochromic monomer composition. Also, the polychromic solid films may

optionally be combined with inorganic and organic films such as those of metal oxides (e.g., WO_3 , NiO , IrO_2 , etc.) and organic films such a polyaniline. Examples of such films are found in copending US Patent Application 08/429,643 filed April 27, 1995, copending
5 US Patent Application 08/547,578 filed October 24, 1995, and copending US Patent Application 08/330,090 filed October 26, 1994, the disclosures of which are hereby incorporated by reference herein. Also, the
10 devices of this present invention can benefit from the use of elemental semiconductors layers or stacks, PRM, anti-wetting adaption, synchronous manufacturing, multi-layer transparent conducting stacks incorporating a thin metal layer overcoated with a conducting metal oxide (such as a high reflectivity reflector comprising
15 around 1000 Å of silver metal or aluminum metal, overcoated with about 1500 Å of ITO and with a reflectivity greater than 70%R and a sheet resistance below 5 ohms/square), conducting seals, variable intensity band pass filters, isolation valve vacuum
20 backfilling, cover sheets and on demand displays such as are disclosed in copending US Patent Application 08/429,643 filed April 27, 1995, the disclosure of which is hereby incorporated by reference herein. Also, as further disclosed in copending US Patent
25 Application 08/429,643, the solid polymer films of this present invention may comprise within their structure electrochromatically active phthalocyanine-based and/or phthalocyanine-derived moieties including transition metal phthalocyanines such as zirconium phthalocyanine
30 and molybdenum phthalocyanine. Also, the solid polymer films of this invention can be combined with an electron donor (e.g. TiO_2) - spacer (salicylic acid or phosphoric acid bound to the TiO_2) - electron acceptor (a viologen bound to the salicylic acid or to the
35 phosphoric acid) heterodyad such as described also in US Patent Application 08/429,643. Such donor-spacer-

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acceptor solid films can function as an electrochromic solid film in combination with the polychromic solid films of the present invention. Further, such as described in US Patent Application 08/429,643, such chemically modified nanoporous-nanocrystalline films, such as of TiO_2 with absorbed redox chromophores, can be used in a variety of electrochromic devices and device constructions, including rearview mirrors, glazings, architectural and vehicular glazings, displays, filters, contrast enhancement filters and the like.

Many electrochromic compounds absorb electromagnetic radiation in the about 290 nm to about 400 nm ultraviolet region. Because solar radiation includes an ultraviolet region between about 290 nm to about 400 nm, it is often desirable to shield such electrochromic compounds from ultraviolet radiation in that region. By so doing, the longevity and stability of the electrochromic compounds may be improved. Also, it is desirable that the polychromic solid film itself be stable to electromagnetic radiation, particularly in that region. This may be accomplished by adding to the electrochromic monomer composition an ultraviolet stabilizing agent (and/or a self-screening plasticizer which may act to block or screen such ultraviolet radiation) so as to extend the functional lifetime of the resulting polychromic solid film. Such ultraviolet stabilizing agents (and/or self-screening plasticizers) should be substantially transparent in the visible region and function to absorb ultraviolet radiation, quench degradative free radical reaction formation and prevent degradative oxidative reactions.

As those of ordinary skill in the art will readily appreciate, the preferred ultraviolet stabilizing agents, which are usually employed on a by-weight basis, should be selected so as to be compatible with

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the other components of the electrochromic monomer composition, and so that the physical, chemical or electrochemical performance of, as well as the transformation into, the resulting polychromic solid film is not adversely affected.

Although many materials known to absorb ultraviolet radiation may be employed herein, preferred ultraviolet stabilizing agents include "UVINUL" 400 [2,4-dihydroxy-benzophenone (manufactured by BASF Corp., Wyandotte, Michigan)], "UVINUL" D 49 [2,2'-dihydroxy-4,4'-dimethoxybenzophenone (BASF Corp.)], "UVINUL" N 35 [ethyl-2-cyano-3,3-diphenylacrylate (BASF Corp.)], "UVINUL" N 539 [2-ethylhexyl-2-cyano-3,3'-diphenylacrylate (BASF Corp.)], "UVINUL" M 40 [2-hydroxy-4-methoxybenzophenone (BASF Corp.)], "UVINUL" M 408 [2-hydroxy-4-octoxybenzophenone (BASF Corp.)], "TINUVIN" P [2-(2'-hydroxy-5'-methylphenyl)-triazole] (Ciba Geigy Corp.)], "TINUVIN" 327 [2-(3',5'-di-t-butyl-2'-hydroxyphenyl)-5-chloro-benzotriazole (Ciba Geigy Corp.)], "TINUVIN" 328 [2-(3',5'-di-n-pentyl-2'-hydroxyphenyl)-benzotriazole (Ciba Geigy Corp.)] and "CYASORB UV" 24 [2,2'-dihydroxy-4-methoxy-benzophenone (manufactured by American Cyanamid Co., Wayne, New Jersey)], with "UVINUL" M 40, "UVINUL M" 408, "UVINUL" N 35 and "UVINUL" N 539 being the most preferred ultraviolet stabilizing agents when used in a by-weight range of about 0.1% to about 15%, with about 4% to about 10% being preferred.

Since solar radiation includes an ultraviolet region only between about 290 nm and 400 nm, the cure wave length of the electrochromic monomer composition, the peak intensity of the source of electromagnetic radiation, and the principle absorbance maxima of the ultraviolet stabilizing agents should be selected to provide a rapid and efficient transformation of the

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electrochromic monomer compositions into the polychromic solid films, while optimizing the continued long-term post-cure stability to outdoor weathering and all-climate exposure of polychromic solid films.

5 An electrolytic material may also be employed in the electrochromic monomer composition to assist or enhance the conductivity of the electrical current passing through the resulting polychromic solid film. The electrolytic material may be selected from a host of
10 known materials, preferred of which are tetraethylammonium perchlorate, tetrabutylammonium tetrafluoroborate, tetrabutylammonium hexafluorophosphate, tetrabutylammonium trifluoromethane sulfonate, lithium salts and
15 combinations thereof, with tetrabutylammonium hexafluorophosphate and tetraethylammonium perchlorate being the most preferred.

20 In addition, adhesion promoting agents or coupling agents may be used in the preferred electrochromic monomer compositions to further enhance the degree to which the resulting polychromic solid films adhere to the contacting surfaces. Adhesion promoting or
25 coupling agents, which promote such enhanced adhesion, include silane coupling agents, and commercially available adhesion promoting agents like those sold by Sartomer Co., such as Alkoxylated Trifunctional Acrylate (9008), Trifunctional Methacrylate Ester (9010 and 9011), Trifunctional Acrylate Ester (9012),
30 Aliphatic Monofunctional Ester (9013 and 9015) and Aliphatic Difunctional Ester (9014). Moreover, carboxylated vinyl monomers, such as methacrylic acid, vinyl carboxylic acid and the like may be used to further assist the development of good adhesion to the
35 contacting surfaces.

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And, coloring agents, spacers, anti-oxidizing agents, flame retarding agents, heat stabilizing agents and combinations thereof may be added to the electrochromic monomer compositions, choosing of course those materials in appropriate quantities depending upon the specific application of the resulting polychromic solid film. For instance, a blue-tinted electrochromic automotive mirror, such as described herein, may be prepared by dispersing within the electrochromic monomer composition a suitable ultraviolet stable coloring agent, such as "NEOZAPON" BLUE TM 807 (a phthalocyanine blue dye, available commercially from BASF Corp., Parsippany, New Jersey) and "NEOPEN" 808 (a phthalocyanine blue dye, available commercially from BASF Corp.).

Polychromic solid films may be prepared within an electrochromic device by introducing an electrochromic monomer composition to a film forming means, such as the vacuum backfilling technique, which fills a cavity of an assembly by withdrawing into the cavity the electrochromic monomer composition while the assembly is in an environment of reduced atmospheric pressure [see e.g., Varaprasad II], the two hole filling technique, where the electrochromic monomer composition is dispensed under pressure into the assembly through one hole while a gentle vacuum is applied at the other hole [see e.g., Varaprasad III], or with the sandwich lamination technique, which contemporaneously creates and fills a cavity of an assembly by placing on one or both substrates either a thermoplastic sealing means to act as a spacing means [see commonly assigned United States Patent 5,233,461 (Dornan)] or glass beads of nominal diameter, and then exposing to electromagnetic radiation at least one clear substrate of the assembly constructed by any of the above manufacturing techniques (containing the low viscosity electrochromic

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monomer composition) for a time sufficient to transform the electrochromic monomer composition into a polychromic solid film.

5 In connection with such film forming means, spacers, such as glass beads, may be dispensed across the conductive surface of one or both substrates, or dispersed throughout the electrochromic monomer composition which may then be dispensed onto the conductive surface of one or both substrates, to assist
10 in preparing a polychromic solid film which contacts, in abutting relationship, the conductive surface of the two substrates. Similarly, a pre-established spacing means of solid material, such as tape, pillars, walls, ridges and the like, may also be employed to assist in
15 determining the interpane distance between the substrates in which a polychromic solid film may be prepared to contact, in abutting relationship with, the conductive surface of the two substrates.

20 Polychromic solid films may also be prepared separately from the electrochromic device, and thereafter placed between, and in abutting relationship with, the conductive surface of the two substrates used in constructing the device. Many known film manufacturing
25 processes may be employed as a film forming means to manufacture polychromic solid films. Included among these processes are calendering, casting, rolling, dispensing, coating, extrusion and thermoforming. For a non-exhaustive description of such processes, see
30 Modern Plastics Encyclopedia 1988, 203-300, McGraw-Hill Inc., New York (1988). For instance, the electrochromic monomer composition may be dispensed or coated onto the conductive surface of a substrate, using conventional techniques, such as curtain coating,
35 spray coating, dip coating, spin coating, roller coating, brush coating or transfer coating.

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As described above, polychromic solid films may be prepared as a self-supporting solid film which may thereafter be contacted with conductive substrates.

5 For instance, an electrochromic monomer composition may be continuously cast or dispensed onto a surface, such as a fluorocarbon surface and the like, to which the polychromic solid film, transformed therefrom by exposure to electromagnetic radiation, does not adhere. In this way, polychromic solid films may be continuously prepared, and, for example, reeled onto a take-up roller and stored for future use. Thus, when a particular electrochromic device is desired, an appropriately shaped portion of the stored polychromic solid film may be cut from the roll using a die, laser, 10 hot wire, blade or other cutting means. This now custom-cut portion of polychromic solid film may be contacted with the conductive substrates to form an electrochromic device. 15

20 For example, the custom-cut portion of the polychromic solid film may be laminated between the conductive surface of two transparent conductive coated substrates, such as ITO or tin oxide coated glass substrates, two ITO or tin oxide coated "MYLAR" 25 [polyethylene terephthalate film (commercially available from E.I. du Pont de Nemours and Co., Wilmington, Delaware)] substrates or one ITO or tin oxide coated glass substrate and one ITO or tin oxide coated "MYLAR" substrate. To this end, it may be 30 desirable to allow for residual cure in the stored polychromic solid film so that adhesion to the conductive substrates in the laminate to be formed is facilitated and optimized.

35 In this regard, a polychromic solid film may be prepared by the film forming means of extrusion or

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calendaring wherein the electrochromic monomer composition is transformed into the polychromic solid film by exposure to electromagnetic radiation prior to, contemporaneously with, or, if the electrochromic monomer composition is sufficiently viscous, after
5 passing through the extruder or calendar. Thereafter, the polychromic solid film may be placed between, and in abutting relationship with, the conductive surface of the substrates, and then construction of the electrochromic device may be completed.

10 While preparing polychromic solid films, the viscosity of the electrochromic monomer composition may be controlled to optimize its dispensibility by adjusting the temperature of (1) the electrochromic monomer
15 composition itself, (2) the substrates on which the electrochromic monomer composition may be placed to assemble the electrochromic device or (3) the processing equipment used to prepare polychromic solid films (if the polychromic film is to be prepared
20 independently from the substrates of the electrochromic devices). For example, the temperature of the electrochromic monomer composition, the substrates or the equipment or combinations thereof may be elevated to decrease the viscosity of the electrochromic monomer
25 composition. Similarly, the uniformity on the substrate of the dispensed electrochromic monomer composition may be enhanced using lamination techniques, centrifuge techniques, pressure applied from the atmosphere (such as with vacuum bagging),
30 pressure applied from a weighted object, rollers and the like.

The substrates employed in the electrochromic devices
35 of the present invention may be constructed from materials that are substantially inflexible as well as flexible depending on the application to which they are

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to be used. In this regard, the substrates may be constructed from substantially inflexible substrates, such as glass, laminated glass, tempered glass, optical plastics, such as polycarbonate, acrylic and polystyrene, and flexible substrates, such as "MYLAR" film. Also, the glass substrates suitable for use herein may be tinted specialized glass which is known to significantly reduce ultraviolet radiation transmission while maintaining high visible light transmission. Such glass, often bearing a blue colored tint, provides a commercially acceptable silvery reflection to electrochromic automotive mirrors even when the polychromic solid film is prepared containing an ultraviolet stabilizing agent or other component which may have a tendency to imbue a yellowish appearance to the polychromic solid film. Preferably, blue tinted specialized glass may be obtained commercially from Pittsburgh Plate Glass Industries, Pittsburgh, Pennsylvania as "SOLEXTRA" 7010; Ford Glass Co., Detroit, Michigan as "SUNGLAS" Blue; or Asahi Glass Co., Tokyo, Japan under the "SUNBLUE" tradename. Whether the chosen substrate is substantially inflexible or flexible, a transparent conductive coating, such as indium tin oxide ("ITO") or doped-tin oxide, is coated on a surface of the substrate making that surface suitable for placement in abutting relationship with a polychromic solid film.

The choice of substrate may influence the choice of processing techniques used to prepare the polychromic solid film or the type of electrochromic device assembled. For example, when assembling an electrochromic device from flexible substrates, an electrochromic monomer composition may be advantageously applied to such flexible substrates using a roll-to-roll system where the flexible substrates are released from rolls (that are aligned

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and rotate in directions opposite to one another), and brought toward one another in a spaced-apart relationship. In this way, the electrochromic monomer composition may be dispensed or injected onto one of the flexible substrates at the point where the two flexible substrates are released from their respective rolls and brought toward one another, while being contemporaneously exposed to electromagnetic radiation for a time sufficient to transform the electrochromic monomer composition into a polychromic solid film.

The dispensing of the electrochromic monomer composition may be effected through a first injection nozzle positioned over one of the rolls of flexible substrate. A weathering barrier forming material, such as a curing epoxide like an ultraviolet curing epoxide, may be dispensed in an alternating and synchronized manner onto that flexible substrate through a second injection nozzle positioned adjacent to the first injection nozzle. By passing in the path of these nozzles as a continuously moving ribbon, a flexible substrate may be contacted with the separate polymerizable compositions in appropriate amounts and positions on the flexible substrate.

In manufacturing flexible electrochromic assemblies having a dimension the full width of the roll of flexible substrate, a weathering barrier forming material may be dispensed from the second injection nozzle which may be positioned inboard (typically about 2 mm to about 25 mm) from the leftmost edge of the roll of flexible substrate. The first injection nozzle, positioned adjacent to the second injection nozzle, may dispense the electrochromic monomer composition onto most of the full width of the roll of flexible substrate. A third injection nozzle, also dispensing weathering barrier forming material, may be positioned

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adjacent to, but inboard from, the rightmost edge of that roll of flexible substrate (typically about 2 mm to about 25 mm). In this manner, and as described above, a continuous ribbon of a flexible electrochromic assembly may be formed (upon exposure to
5 electromagnetic radiation) which, in turn, may be taken up onto a take-up roller. By so doing, a flexible electrochromic assembly having the width of the roll of flexible substrate, but of a particular length, may be obtained by unrolling and cutting to length an
10 electrochromic assembly of a particular size.

Should it be desirable to have multiple flexible electrochromic assemblies positioned in the same taken-up roll, multiple nozzles may be placed appropriately
15 at positions throughout the width of one of the rolls of flexible substrate, and the dispensing process carried out accordingly.

In that regard, a small gap (e.g., about 5 mm to about
20 50 mm) should be maintained where no dispensing occurs during the introduction of the electrochromic monomer composition and the weathering barrier forming material onto the substrate so that a dead zone is created where neither the electrochromic monomer composition nor the
25 weathering barrier forming material is present. Once the weathering barrier and polychromic solid film have formed (see infra), the electrochromic assembly may be isolated by cutting along the newly created dead zones of the flexible assemblies. This zone serves
30 conveniently as a cutting area to form electrochromic assemblies of desired sizes.

And, the zones outboard of the respective weathering barriers serve as convenient edges for attachment of a
35 means for introducing an applied potential to the flexible electrochromic assemblies, such as bus bars.

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Similarly, the bisection of the dead zones establishes a convenient position onto which the bus bars may be affixed.

While each of the weathering barrier forming material and the electrochromic monomer composition may be transformed into a weathering barrier and a polychromic solid film, respectively, by exposure to electromagnetic radiation, the required exposures to complete the respective transformations may be independent from one another. The weathering barrier forming material may also be thermally cured to form the weathering barrier.

The choice of a particular electromagnetic radiation region to effect in situ cure may depend on the particular electrochromic monomer composition to be cured. In this regard, typical sources of electromagnetic radiation, such as ultraviolet radiation, include: mercury vapor lamps; xenon arc lamps; "H", "D", "X", "M", "V" and "A" fusion lamps (such as those commercially available from Fusion UV Curing Systems, Buffalo Grove, Illinois); microwave generated ultraviolet radiation; solar power and fluorescent light sources. Any of these electromagnetic radiation sources may use in conjunction therewith reflectors and filters, so as to focus the emitted radiation within a particular electromagnetic region. Similarly, the electromagnetic radiation may be generated directly in a steady fashion or in an intermittent fashion so as to minimize the degree of heat build-up. Although the region of electromagnetic radiation employed to in situ cure the electrochromic monomer compositions into polychromic solid films is often referred to herein as being in the ultraviolet region, that is not to say that other regions of radiation within the electromagnetic

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spectrum may not also be suitable. For instance, in certain situations, visible radiation may also be advantageously employed.

5 Bearing in mind that some or all of the components of the electrochromic monomer composition may inhibit, retard or suppress the in situ curing process, a given source of electromagnetic radiation should have a sufficient intensity to overcome the inhibitive effects of those components so as to enable to proceed
10 successfully the transformation of the electrochromic monomer composition into the polychromic solid film. By choosing a lamp with a reflector and, optionally, a filter, a source which itself produces a less advantageous intensity of electromagnetic radiation may
15 suffice. In any event, the chosen lamp preferably has a power rating of at least about 100 watts per inch (about 40 watts per cm), with a power rating of at least about 300 watts per inch (about 120 watts per cm) being particularly preferred. Most preferably, the
20 wavelength of the lamp and its output intensity should be chosen to accommodate the presence of ultraviolet stabilizing agents incorporated into electrochromic monomer compositions. Also, a photoinitiator or photosensitizer, if used, may increase the rate of in
25 situ curing or shift the wavelength within the electromagnetic radiation spectrum at which in situ curing will occur in the transformation process.

30 During the in situ curing process, the electrochromic monomer composition will be exposed to a source of electromagnetic radiation that emits an amount of energy, measured in KJ/m², determined by parameters including: the size, type and geometry of the source; the duration of the exposure to electromagnetic
35 radiation; the intensity of the radiation (and that portion of radiation emitted within the region

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appropriate to effect curing); the absorbance of electromagnetic radiation by any intervening materials, such as substrates, conductive coatings and the like; and the distance the electrochromic monomer composition lies from the source of radiation. Those of ordinary skill in the art will readily appreciate that the polychromic solid film transformation may be optimized by choosing appropriate values for these parameters in view of the particular electrochromic monomer composition.

The source of electromagnetic radiation may remain stationary while the electrochromic monomer composition passes through its path. Alternatively, the electrochromic monomer composition may remain stationary while the source of electromagnetic radiation passes thereover or therearound to complete the transformation into a polychromic solid film. Still alternatively, both may traverse one another, or for that matter remain stationary, provided that the electrochromic monomer composition is exposed to the electrochromic radiation for a time sufficient to effect such in situ curing.

Commercially available curing systems, such as the Fusion UV Curing Systems F-300 B [Fusion UV Curing Systems, Buffalo Grove, Illinois], Hanovia UV Curing System [Hanovia Corp., Newark, New Jersey] and RC-500 A Pulsed UV Curing System [Xenon Corp., Woburn, Massachusetts], are well-suited to accomplish the transformation. Also, a Sunlighter UV chamber fitted with low intensity mercury vapor lamps and a turntable may accomplish the transformation.

Electromagnetic radiation in the near-infrared and far-infrared (including short and long wavelengths from 3 microns to 30 microns and beyond) regions of the

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electromagnetic spectrum can be used, as can radiation in other regions such as microwave radiation. Thus, for electrochromic monomer compositions responsive to energy input that includes thermal energy, radiant heaters that emit in the infrared region and couple energy into the monomer composition can be used. For compositions responsive to microwave energy, a microwave generator can be used. Also, for systems that respond, for example, to a combination of energy inputs from different regions of the electromagnetic spectrum, a combined energy radiator can be used. For example, the Fusion UV Curing System, Sunlight UV Chamber, Hanovia UV Curing System, and RC-500A Pulsed UV Curing System described above emit energy efficiently in both the ultraviolet region and the infrared region, and thus effect a cure both by photoinitiation and thermally. For systems responsive to thermal influences, ovens, lehrs, conveyerized ovens, induction ovens, heater banks and the like can be used to couple energy into the electrochromic monomer composition by convection, conduction and/or radiation. Also, chemical initiators and catalysts, photo initiators, latent curing agents (such as are described in copending US Patent Application 08/547,578, the disclosure of which is hereby incorporated by reference herein) and similar chemical accelerants can be used to assist conversion of the electrochromic monomer composition into a cross-linked solid polymer matrix. By customizing and selecting the components of the electrochromic monomer composition, cure can be retarded/suppressed until after the composition is applied within the cavity of the electrochromic cell. Thereafter, by exposure to electromagnetic radiation or thermal influence, cure to the solid polymer matrix polychromic film can be accelerated. Since devices will not typically be consumer used until at least days (often weeks or

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passes using a less intense energy source, may be preferable over a rapid cure using a more intense energy source, for example, to minimize shrinkage during the transformation process. Also, it is desirable to use a source of electromagnetic radiation that is delivered in an intermittent fashion, such as by pulsing or strobing, so as to ensure a thorough and complete cure without causing excessive heat build-up.

In transforming electrochromic monomer compositions into polychromic solid films, shrinkage may be observed during and after the transformation process of the electrochromic monomer composition into a polychromic solid film. This undesirable event may be controlled or lessened to a large extent by making appropriate choices among the components of the electrochromic monomer composition. For instance, appropriately chosen polyfunctional monomers or cross-linking agents may enhance resistance to shrinkage during the transformation process. In addition, a conscious control of the type and amount of plasticizer used in the electrochromic monomer composition may also tend to enhance resistance to shrinkage. While shrinkage may also be observed with polychromic solid films that have been subjected to environmental conditions, especially conditions of environmental accelerated aging, such as thermal cycling and low temperature soak, a conscious choice of components used in the electrochromic monomer composition may tend to minimize this event as well. In general, shrinkage may be decreased as the molecular weight of the monomer employed is increased, and by using index matched inert fillers, such as glass beads or fibres.

Electrochromic devices may be manufactured with polychromic solid films of a particular thickness by preparing partially-cured polychromic solid films

between the glass substrates of electrochromic assemblies with spacers or a thermoplastic spacing means having been placed on one or both of the substrates. This partially-cured polychromic solid film should have a thickness slightly greater than that which the resulting polychromic solid film will desirably assume in the completed device. The electrochromic assemblies should then be subjected to compression, such as that provided by an autoclave/vacuum bagging process, and thereafter be exposed to electromagnetic radiation to complete the transformation into a polychromic solid film with the desired film thickness.

Figures 1 and 2 show an electrochromic device assembled from the polychromic solid films of the present invention. The electrochromic assembly 1 includes two substantially planar substrates 2, 3 positioned substantially parallel to one another. It is preferable that these substrates 2, 3 be positioned as close to parallel to one another as possible so as to avoid double imaging, which is particularly noticeable in mirrors, especially when the electrochromic media -- i.e., the polychromic solid film -- is colored to a dimmed state.

A source of an applied potential need be introduced to the electrochromic assembly 1 so that polychromic solid film 6 may color in a rapid, intense and uniform manner. That source may be connected by electrical leads 8 to conducting strips, such as bus bars 7. The bus bars 7 may be constructed of a metal, such as copper, stainless steel, aluminum or solder, or of conductive frits and epoxides, and should be affixed to a conductive coating 4, coated on a surface of each of the substrates 2, 3. An exposed portion of the conductive coating 4 should be provided for the bus

bars 7 to adhere by the displacement of the coated substrates 2, 3 in opposite directions relative to each other -- lateral from, but parallel to --, with polychromic solid film 6 positioned between, and in abutting relationship with, the conductive surface of the two substrates.

As noted above, coated on a surface of each of these substrates 2, 3 is a substantially transparent conductive coating 4. The conductive coating 4 is generally from about 300 Å to about 10,000 Å in thickness, having a refractive index in the range of about 1.6 to about 2.2. Preferably, a conductive coating 4 with a thickness of about 1,200 Å to about 2,300 Å, having a refractive index of about 1.7 to about 1.9, is chosen depending on the desired appearance of the substrate when the polychromic solid film situated therebetween is colored.

The conductive coating 4 should also be highly and uniformly conductive in each direction to provide a substantially uniform response as to film coloring once a potential is applied. The sheet resistance of these transparent conductive substrates 2, 3 may be below about 100 ohms per square, with about 6 ohms per square to about 20 ohms per square being preferred. Such substrates 2, 3 may be selected from among those commercially available as glass substrates, coated with indium tin oxide ("ITO") from Donnelly Corporation, Holland, Michigan, or tin oxide-coated glass substrates sold by the LOF Glass division of Libbey-Owens-Ford Co., Toledo, Ohio under the tradename of "TEC-Glass" products, such as "TEC 10" (10 ohms per square sheet resistance), "TEC 12" (12 ohms per square sheet resistance), "TEC 15" (15 ohms per square sheet resistance) and "TEC 20" (20 ohms per square sheet resistance) tin oxide-coated glass. Moreover, tin

oxide coated glass substrates, commercially available from Pittsburgh Plate Glass Industries, Pittsburgh, Pennsylvania under the "SUNGATE" tradename, may be advantageously employed herein. Also, substantially transparent conductive coated flexible substrates, such as ITO deposited onto substantially clear or tinted "MYLAR", may be used. Such flexible substrates are commercially available from Southwall Corp., Palo Alto, California.

The conductive coating 4 coated on each of the substrates 2, 3 may be constructed from the same material or different materials, including tin oxide, ITO, ITO-FW, ITO-HW, ITO-HWG, doped tin oxide, such as antimony-doped tin oxide and fluorine-doped tin oxide, doped zinc oxide, such as antimony-doped zinc oxide and aluminum-doped zinc oxide, with ITO being preferred.

The substantially transparent conductive coated substrates 2, 3 may be of the full-wave length-type ("FW") (about 6 ohms per square to about 8 ohms per square sheet resistance), the half-wave length-type ("HW") (about 12 ohms per square to about 15 ohms per square sheet resistance) or the half-wave length green-type ("HWG") (about 12 ohms per square to about 15 ohms per square sheet resistance). The thickness of FW is about 3,000 Å in thickness, HW is about 1,500 Å in thickness and HWG is about 1,960 Å in thickness, bearing in mind that these substantially transparent conductive coated substrates 2, 3 may vary as much as about 100 to about 200 Å. HWG has a refractive index of about 1.7 to about 1.8, and has an optical thickness of about five-eighths wave to about two-thirds wave. HWG is generally chosen for electrochromic devices, especially reflective devices, such as mirrors, whose desired appearance has a greenish hue in color when a potential is applied.

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Optionally, and for some applications desireably, the spaced-apart substantially transparent conductive coated substrates 2, 3 may have a weather barrier 5 placed therebetween or therearound. The use of a weather barrier 5 in the electrochromic devices of the present invention is for the purpose of preventing environmental contaminants from entering the device during long-term use under harsh environmental conditions rather than to prevent escape of electrochromic media, such as with an electrochemichromic device. Weather barrier 5 may be made from many known materials, with epoxy resins coupled with spacers, plasticized polyvinyl butyral (available commercially under the "SAFLEX" tradename from Monsanto Co., St. Louis, Missouri), ionomer resins (available commercially under the "SURLYN" tradename from E.I. du Pont de Nemours and Co., Wilmington, Delaware) and "KAPTON" high temperature polyamide tape (available commercially from E.I. du Pont de Nemours and Co., Wilmington, Delaware) being preferred. In general, it may be desirable to use within the electrochromic device, and particularly for weather barrier 5, materials such as nitrile containing polymers and butyl rubbers that form a good barrier against oxygen permeation from environmental exposure.

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A further recitation of weather barrier materials and types (including single and double weather barrier constructions) is found in copending US Patent Application 08/547,578 filed October 24, 1995, the disclosure of which is hereby incorporated by reference herein, including flexible weather barrier materials that are beneficial when the polychromic solid film devices of this invention are exposed to wide and rapid oscillation between temperature extremes, such as the thermal shocks experienced during normal use in or on a vehicle in regions of climate extremes. Also, devices,

such as electrochromic rearview mirrors utilizing a polychromic solid film, can be constructed suitable for use on automobiles, and suitable to withstand accelerated aging testing such as boiling in water for an extended period (such as 96 hours or longer);
5 exposure to high temperature/high humidity for an extended period (for example, 85°C/85% relative humidity for 720 hours or longer); exposure within a steam autoclave for extended periods (for example, 121°C; 15-18 psi steam for 144 hours or longer).

In the sandwich lamination technique, see supra, it is the thickness of the polychromic solid film itself, especially when a highly viscous electrochromic monomer composition is used, optionally coupled with either
15 spacers or a thermoplastic spacing means, assembled within the electrochromic devices of the present invention that determines the interpane distance of the spaced-apart relationship at which the substrates are positioned. This interpane distance may be influenced
20 by the addition of spacers to the electrochromic monomer composition, which spacers, when added to an electrochromic monomer composition, assist in defining the film thickness of the resulting polychromic solid film. And, the thickness of the polychromic solid film
25 may be about 10 μm to about 1000 μm , with about 20 μm to about 200 μm being preferred, a film thickness of about 37 μm to about 74 μm being particularly preferred, and a film thickness of about 53 μm being most preferred depending of course on the chosen
30 electrochromic monomer composition and the intended application.

By taking appropriate measures, electrochromic devices
35 manufactured with polychromic solid films may operate so that, upon application of a potential thereto, only selected portions of the device -- i.e., through the

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polychromic solid film -- will color in preference to the remaining portions of the device. In such segmented electrochromic devices, lines may be scored or etched onto the conductive surface of either one or both of substrates 2, 3, in linear alignment so as to cause a break in electrical continuity between regions immediately adjacent to the break, by means such as chemical etching, mechanical scribing, laser etching, sand blasting and other equivalent means. By so doing, an addressable pixel may be created by the break of electrical continuity when a potential is applied to a pre-determined portion of the electrochromic device. The electrochromic device colors in only that pre-determined portion demonstrating utility, for example, as an electrochromic mirror, where only a selected portion of the mirror advantageously colors to assist in reducing locally reflected glare or as an electrochromic information display device.

To prepare an electrochromic device containing a polychromic solid film, the electrochromic monomer composition may be dispensed onto the conductive surface of one of the substrates 2 or 3. The conductive surface of the other substrate may then be placed thereover so that the electrochromic monomer composition is dispersed uniformly onto and between the conductive surface of substrates 2, 3.

This assembly may then be exposed, either in a continuous or intermittent manner, to electromagnetic radiation, such as ultraviolet radiation in the region between about 200 nm to about 400 nm for a period of about 2 seconds to about 10 seconds, so that the electrochromic monomer composition is transformed by in situ curing into polychromic solid film 6. The intermittent manner may include multiple exposures to such energy.

Once the electrochromic device is assembled with polychromic solid film 6, a potential may be applied to the bus bars 7 in order to induce film coloring. The applied potential may be supplied from a variety of sources including, but not limited to, any source of alternating current ("AC") or direct current ("DC") known in the art, provided that, if an AC source is chosen, control elements, such as diodes, should be placed between the source and each of the conductive coatings 4 to ensure that the potential difference between the conductive coatings 4 does not change polarity with variations in polarity of the applied potential from the source. Suitable DC sources are storage batteries, solar thermal cells, photovoltaic cells or photoelectrochemical cells.

An electrochromic device, such as an electrochromic shade band where a gradient opacity panel may be constructed by positioning the bus bars 7 along the edges of the substrates in such a way so that only a portion -- e.g., the same portion -- of each of the substrates 2, 3 have the bus bars 7 affixed thereto. Thus, where the bus bars 7 are aligned with one another on opposite substrates 2, 3, the introduction of an applied potential to the electrochromic device will cause intense color to be observed in only that region of the device onto which an electric field has been created -- i.e., only that region of the device having the bus bars 7 so aligned. A portion of the remaining bleached region will also exhibit color extending from the intensely colored region at the bus bar/non-bus bar transition gradually dissipating into the remaining bleached region of the device.

The applied potential generated from any of these sources may be introduced to the polychromic solid film of the electrochromic device in the range of about

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0.001 volts to about 5.0 volts. Typically, however, a potential of about 0.2 volts to about 2.0 volts is preferred, with about 1 volt to about 1.5 volts particularly preferred, to permit the current to flow across and color the polychromic solid film 6 so as to lessen the amount of light transmitted therethrough. The extent of coloring -- i.e., high transmittance, low transmittance and intermediate transmittance levels -- at steady state in a particular device will often depend on the potential difference between the conductive surface of the substrates 2, 3, which relationship permits the electrochromic devices of the present invention to be used as "gray-scale" devices, as that term is used by those of ordinary skill in the art.

A zero potential or a potential of negative polarity (i.e., a bleaching potential) may be applied to the bus bars 7 in order to induce high light transmittance through polychromic solid film 6. A zero potential to about -0.2 volts will typically provide an acceptable response time for bleaching; nevertheless, increasing the magnitude of the negative potential to about -0.7 volts will often enhance response times. And, a further increase in the magnitude of that potential to about -0.8 volts to about -0.9 volts, or a magnitude of even more negative polarity as the art-skilled should readily appreciate, may permit polychromic solid film 6 to form a light-colored tint while colored to a partial- or fully-dimmed state.

In electrochromic devices where the polychromic solid film is formed within the assembly by exposure to electromagnetic radiation, the performance of the device may be enhanced by applying the positive polarity of the potential to the substrate that faced the electromagnetic radiation during the transformation

process. Thus, in the case of electrochromic mirrors manufactured in such a manner, the positive polarity of the potential should be applied to the conductive surface of the clear, front glass substrate, and the negative polarity of the potential applied to the conductive surface of the silvered, rear glass substrate, to observe such a beneficial effect.

In the context of an electrochromic mirror assembly, a reflective coating, having a thickness in the range of 250 Å to about 2,000 Å, preferably about 1,000 Å, should thereafter be applied to one of the transparent conductive coated glass substrates 2 or 3 in order to form a mirror. Suitable materials for this layer are aluminum, palladium, platinum, titanium, gold, chromium, silver and stainless steel, with silver being preferred. As an alternative to such metal reflectors, multi-coated thin film stacks of dielectric materials or a high index single dielectric thin film coating may be used as a reflector. Alternatively, one of the conductive coatings 4 may be a metallic reflective layer which serves not only as an electrode, but also as a mirror.

It is clear from the teaching herein that should a window, sun roof or the like be desirably constructed, the reflective coating need only be omitted from the assembly so that the light which is transmitted through the transparent panel is not further assisted in reflecting back therethrough.

Similarly, an electrochromic optically attenuating contrast filter may be manufactured in the manner described above, optionally incorporating into the electrochromic assembly an anti-reflective means, such as a coating, on the front surface of the outermost substrate as viewed by an observer (see e.g., Lynam V);

an anti-static means, such as a conductive coating, particularly a transparent conductive coating of ITO, tin oxide and the like; index matching means to reduce internal and interfacial reflections, such as thin films of an appropriately selected optical path length; and/or light absorbing glass, such as glass tinted to a neutral density, such as "GRAYLITE" gray tinted glass (commercially available from Pittsburgh Plate Glass Industries, Pittsburgh, Pennsylvania) and "SUNGLAS" Gray gray tinted glass (commercially available from Ford Glass Co., Detroit, Michigan), to augment contrast enhancement. Moreover, polymer interlayers, which may be tinted gray, such as those used in electrochromic constructions as described in Lynam III, may be incorporated into such electrochromic optically attenuating contrast filters.

Electrochromic optical attenuating contrast filters may be an integral part of a device or may be affixed to an already constructed device, such as cathode ray tube monitors. For instance, an optical attenuating contrast filter may be manufactured from a polychromic solid film and then affixed, using a suitable optical adhesive, to a device that should benefit from the properties and characteristics exhibited by the polychromic solid film. Such optical adhesives maximize optical quality and optical matching, and minimize interfacial reflection, and include plasticized polyvinyl butyral, various silicones, polyurethanes such as "NORLAND NOA 65" and "NORLAND NOA 68", and acrylics such as "DYMEX LIGHT-WELD 478". In such contrast filters, the electrochromic compounds are chosen for use in the polychromic solid film so that the electrochromic assembly may color to a suitable level upon the introduction of an applied potential thereto, and no undesirable spectral bias is exhibited. Preferably, the polychromic solid film should dim

Polychromic solid films may be used in electrochromic devices, particularly glazings and mirrors, whose functional surface is substantially planar or flat, or that are curved with a convex curvature, a compound curvature, a multi-radius curvature, a spherical curvature, an aspheric curvature, or combinations of such curvature. For example, flat electrochromic automotive mirrors may be manufactured using the polychromic solid films of the present invention. Also, convex electrochromic automotive mirrors may be manufactured, with radii of curvature typically in the range of about 25" to about 250", preferably in the range of about 35" to about 100", as are conventionally known. In addition, multi-radius automotive mirrors, such as those described in United States Patent 4,449,786 (McCord), may be manufactured using the polychromic solid films of the present invention. Multi-radius automotive mirrors may be used typically on the driver-side exterior of an automobile to extend the driver's field of view and to enable the driver to safely see rearward and to avoid blind-spots in the rearward field of view. Generally, such mirrors comprise a higher radius (even flat) region closer to the driver and a lower radius (i.e., more curved) region outboard from the driver that serves principally as the blind-spot detection zone in the mirror. Indeed, such polychromic solid film-containing electrochromic multi-radius automotive mirrors may benefit from the prolonged coloration performance of polychromic solid films and/or from the ability to address individual segments in such mirrors.

The demarcation means may also include an etching of a deletion line or an otherwise established break in the electrical continuity of the transparent conductors used in such mirrors so that either one or both regions may be individually or mutually addressed. Optionally, this deletion line may itself be colored black. Thus, the outboard, more curved region may operate independently from the inboard, less curved region to provide an electrochromic mirror that operates in a segmented arrangement. Upon the introduction of an applied potential, either of such regions may color to a dimmed intermediate reflectance level, independent of the other region, or, if desired, both regions may operate together in tandem.

An insulating demarcation means, such as demarcation lines, dots and/or spots, may be placed within electrochromic devices, such as mirrors, glazings, optically attenuating contrast filters and the like, to assist in creating the interpane distance of the device and to enhance overall performance, in particular the uniformity of coloration across large area devices. Such insulating demarcation means, constructed from, for example, epoxy coupled with glass spacer beads, plastic tape or die cut from plastic tape, may be placed onto the conductive surface of one or more substrates by silk-screening or other suitable technique prior to assembling the device. The insulating demarcation means may be geometrically positioned across the panel, such as in a series of parallel, uniformly spaced-apart lines, and may be clear, opaque, tinted or colorless and appropriate

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location within the electrochromic device. By using such a screen-on technique as described above, a more defined and predictable layout of the insulating demarcation means may be achieved. Further, such spacers may be a rigid insoluble spacer material such as glass or be rigid soluble spacer material (such as a polymer such as polycarbonate, polymethylmethacrylate, polystyrene and the like) capable of dissolving in the plasticizer of the monomer composition. For example, rigid, soluble polymer spacer beads can be sprinkled across the conductive surface of a substrate and so help define an interpane spacing when the device is first assembled. Then, when the monomer composition is dispensed into the interpane spacing (after the establishment of the interpane spacing with the assistance of soluble polymer spacers), then over time the soluble spacer beads dissolve in the plasticizer, preferably prior to in situ conversion to the solid polychromic film.

Using such insulating demarcation means, one or both of the substrates, either prior to or after assembly in the device, may be divided into separate regions with openings or voids within the insulating demarcation means interconnecting adjacent regions so as to permit a ready introduction of the electrochromic monomer composition into the assembly.

A demarcation means may be used that is conductive as well, provided that it is of a smaller thickness than the interpane distance and/or a layer of an insulating material, such as a non-conductive epoxy, urethane or acrylic, is applied thereover so as to prevent conductive surfaces from contacting one another and thus short-circuiting the electrochromic assembly. Such conductive demarcation means include conductive frits, such as silver frits like the # 7713 silver

conductive frit available commercially from E.I. de Pont de Nemours and Co., Wilmington, Delaware, conductive paint or ink and/or metal films, such as those disclosed in Lynam IV. Use of a conductive demarcation means, such as a line of the # 7713 silver conductive frit, having a width of about 0.09375" and a thickness of about 50 μ m, placed on the conductive surface of one of the substrates of the electrochromic device may provide the added benefit of enhancing electrochromic performance by reducing bus bar-to-bus bar overall resistance and thus enhancing uniformity of coloration, as well as rapidity of response, particularly over large area devices.

Fabrication of electrochromic multi-radius/aspheric or spherical/convex mirrors can benefit from single or tandem bending such as is described in copending US Patent Application 08/547,578, the disclosure of which is hereby incorporated by reference herein. Convex or multi-radius minilites/shapes can, for example, be individually bent [and thereafter ITO coated or metal reflector coated (such as with a chromium metal reflector, a chromium undercoat, rhodium overcoat metal reflector, a chromium undercoat/aluminum overcoat reflector, or their like, such as is described in US Patent Application 08/547,578 and then the individual bent minilites/shapes can be selectively sorted so that the best matched pairs from a production batch can be selected. For example, bent convex or aspheric minilites/shapes can be bent in production lots such as of 100 pieces or thereabouts. Thereafter, each individual bent minilite/shape is placed in a vision system where the reflection of a pattern of dots, squares, lines, circles, ovals (or the like) is photographed using a digital camera and the position of individual dots, etc. in the pattern, as reflected off the individual minilite/shape being measured, is

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pattern's reflection off the convex (or, if desired, the concave) surface of the bent part. If desired, optical calculations can be made that allow determination of the actual profile of the bent glass based upon measurements taken and calculated from the pattern's reflection.

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Also, an aspheric electrochromic (or a convex electrochromic) mirror can be used as an interior rearview mirror, and can be packaged as a clip-on to an existing vehicular rearview mirror in a manner that is similar to aftermarket wide angle mirrors conventionally known. Such interior aspheric/convex electrochromic mirrors can optionally be solar powered or be powered by a battery pack, for ease of installation in the vehicular aftermarket. Should it be desirable to minimize weight for convex or aspheric inside or outside mirrors, then thin glass (in the thickness range of about 1 mm to about 1.8 mm, or even thinner) can be used for one or both of the substrates used in a laminate electrochromic assembly. Use of such thin glass is described in copending US Patent Application 08/547,578 filed October 24, 1995, the disclosure of which is hereby incorporated by reference herein. Also, cutting of convex and especially aspheric glass can benefit from computer numerical controlled (CNC) cutting where a cutting head is moved under digital computer control. In this regard, a multi-axis CNC cutter is preferred where the cutting head (which may be a diamond tool or wheel, a laser beam, a water jet, an abrasive water jet, or the like) can be moved in three dimensions. Most preferably, and especially for cutting aspheric bent glass, a cutter that moves in three-dimensions but that keeps the cutting tool (such as a diamond wheel) normal (i.e., with a cutting wheel axis at or close to 90° to the tangential plane of the bent glass surface) is

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preferred. For example, a cutting machine such as available from LASER Maschinenbau GmbH & Company KG, Grossbetlingen, Germany can be used to cut aspheric glass. In such a system, the bent glass lite/minilite from which the shape is to be cut is mounted on a support arm that is movable in three dimensions and that generally moves in three dimensions either CNC driven, or by following a cam, along the three-dimensional profile of the aspheric shape being cut. Also, the cutting wheel can be adjusted so that its angle relative to a tangent to the glass at point of cut is close to 90° (and not less than about 70° ; not less than about 80° more preferred and not less than about 85° most preferred). In this manner, movement of the cutting support under the cutting wheel, in combination with adjustment of the pitch of the cutting wheel itself, maintains as close to normal (i.e., 90°) the cutting angle as possible, and thus achievement of a clean, efficient cut and breakout of the shape. While particularly beneficial for aspheric shapes where the radius can change from about 2000 mm to below 600 mm, and smaller, across the surface of the shape, cutting of convex glass can also benefit from maintenance of a near normal cutting angle for the cutting tool (i.e., cutting wheel).

Optionally, a machine vision system can be utilized to determine the surface profile of the glass to be cut and the data as to the profile is fed back to the cutter's CNC controller to properly orientate the glass under the cutting head. Use of a vision system, such as is described above, to scan and measure the system profile of the glass to be cut can be thus used to determine how much offset there is on the radius of the glass relative to the cutting head. CNC controlled sensors can be automatically adjusted on every cutting cycle based on the information received from the vision

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compending US Patent Application 08/429,643 filed April 27, 1995, the disclosure of which is hereby incorporated by reference herein. Preferably, the heater is combined with the mirror reflector mounting plate (also known in the automotive mirror art as the mirror backing plate or the mirror backplate). More preferably, the heater and/or the mirror backing plate is formed (such as by injection molding, extrusion and the like) of a conductive polymer material such as a polymer resin incorporating conductive carbon or conductive metal flakes or fibrils (such as of copper, brass, aluminum, steel or equivalent metal). Most preferably, the heater and the mirror backing plate are formed and attached to the mirror element in an integral molding operation as follows. The mirror reflector glass (that preferably is an electrochromic mirror cell but that, optionally, can be a conventional mirror reflector such as chromed glass) is placed in a mold. A heater (such as a positive temperature coefficient heater pad, or a pad formed from a conductive polymer resin that incorporates metal or carbon conducting particles, or a pad formed from a resin that is intrinsically self-conducting in its resin structure such as a polyaniline resin), is either injection molded onto the rearmost glass surface of the glass reflector element (optionally, with an adhesion promoting primer already applied to the rearmost glass surface and/or with a heat transfer agent applied to the rearmost glass surface), or is attached to the rearmost glass surface (or is already pre-attached to the rearmost glass surface) using a double-sticky tape or a hot melt adhesive (preferably, also conducting and/or of high heat transfer efficiency such as aluminum foil). Finally, a plastic resin is injection molded to form the mirror backing plate (and, optionally, the bezel around the outer perimeter of an electrochromic sideview mirror sub-assembly as is

commonly known in the electrochromic rearview mirror
art). The backing plate for the mirror element is the
means for attachment to the electrical or manually
operated actuator within the complete outside sideview
mirror assembly that enables the driver to change the
orientation of the mirror reflector when mounted on the
vehicle and to select the mirror's alignment relative
to the driver and thus select the rearward view that
suits that particular driver's needs for field of view
rearward. By integral molding, the conventionally
separate steps of separately attaching a heater pad to
the mirror glass and then attaching a separately formed
backing plate can be reduced to a single integral
molding step where components, including the mirror
glass, are loaded into a mold, plastic resin is
injected or plastic resins are co-injected, and a
complete sub-assembly (including heater, connectors,
busbars, wire leads/wire harnesses, heat distributors,
thermistors, thermal cut-off switches, backing plate,
bezel, etc.) is unloaded from the tool after completion
of the integral molding step.

Further, vehicle warning indicia such as the familiar
"OBJECTS MAY BE CLOSER THAN THEY APPEAR" can be created
(such as by silk-screening, dispensing, printing, etc.)
using a conductive material (such as a conductive ink,
conductive paint, conductive polymer and the like). In
this way, electrical conductivity is maintained across
the full surface of the inward facing surface of the
rear substrate (frequently called the third surface).
Where a metal reflector (such as a chromium layer or an
underlayer of chromium overcoated with a higher
reflecting metal layer such as of silver, aluminum or
rhodium) is used as a third surface reflector, the
metal reflector can first be deposited (such as by
sputter deposition utilizing planar magnetron or rotary
magnetron cathodes) onto the conductive surface of TEC

glass (or any other transparent conductive coated surface). Next, the metal reflector can be selectively removed to form the desired indicia (i.e., a "HEATED" symbol, a manufacturer's date code and ID, a hazard warning indicia such is commonly found on signal mirrors such as are available on MY97 Ford Bronco and Ford Expedition vehicles available from Ford Motor Company, Detroit, Michigan and as described in United States Patent 5,207,492 invented by Roberts and issued May 1993, the disclosure of which is hereby incorporated by reference herein). The metal reflector can be removed using chemical etching through a mask or directly using laser scribing (such as with a YAG laser), by controlled sandblasting, and the like. By selectively removing the overlayering metal reflector but leaving the underlying transparent conductor largely intact, electrical conductivity across the third surface (i.e., the inward facing surface of the rear substrate) is largely undistributed, and electrochromic coloration is correspondingly uniform. Should it be desired to read an indicia on a third surface, then backlighting can be provided on the fourth surface (i.e., the non-inward facing surface of the rear substrate) that can be viewed by reading through the indicia created on the third surface by removing a third surface metal reflector. Also, optionally, a conductive indicia of a non-dark color (such as brilliant white) could be created on the second surface (i.e., the inward facing surface of the front substrate) of the laminate electrochromic assembly. Thus, when the electrochromic medium colors, the indicia remains visible as a color contrast against the colored dimmed state of the electrochromic medium. Preferably, and as stated above, the indicia is created from conducting or at least partially conducting material (such as can be achieved using conductive carbon fillers). Alternately, non-conducting non-dark

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described in US Patent 4,973,844 titled "Vehicular
Moisture Sensor and Mounting Apparatus Therefor",
invented by O'Farrell et al. and issued November 27,
1990, the disclosure of which is hereby incorporated
herein by reference), or it may include a non-
5 windshield-contacting rain sensor (such as is described
in PCT International Application PCT/US94/05093
entitled "Multi-Function Light Sensor for Vehicle"
invented by Dennis J. Hegyl, published as WO 94/27262
on November 24, 1994, the disclosure of which is hereby
10 incorporated by reference herein). The rearview mirror
assembly can include a display function (or multiple
display functions).

These displays may perform a single display function or
15 multiple display functions such as providing indication
of an additional vehicle function, such as a compass
mirror display function, a temperature display
function, status of inflation of tires display
function, a passenger air bag disable display function,
20 an automatic rain sensor operation display function,
telephone dial information display function, highway
status information display function, blind spot
indicator display function, or the like. Such display
may be an alpha-numerical display or a multi-pixel
25 display, and maybe fixed or scrolling. Such an
automatic rain sensor operation display function may
include a display function related to both a
windshield-contacting and a non-windshield-contacting
rain sensor, including, for example, where the
30 circuitry to control the rain sensor, electrochromic
dimming of a variable reflectance electrochromic
mirror, and any other mirror-mounted electronic feature
are commonly housed in or on a rearview mirror assembly
and wholly or partially share components on a common
35 circuit board. The blind spot detection display or the
automatic rain sensor operation display may alternate

with other display functions by a display toggle which may be manually operated, time-shared, voice-actuated, or under the control of some other sensed function, such as a change in direction of the vehicle or the like. Should a rain sensor control be associated with, incorporated in, or coupled to the interior rearview mirror assembly, the rain sensor circuitry, in addition to providing automatic or semi-automatic control over operation of the windshield wipers (on the front and/or rear windshield of the vehicle), can control the defogger function to defog condensed vapor on an inner cabin surface of a vehicle glazing (such as the inside surface of the front windshield, such as by operating a blower fan, heater function, air conditioning function, or their like), or the rain sensor control can close a sunroof or any other movable glazing should rain conditions be detected. As stated above, it may be advantageous for the rain sensor control (or any other feature such as a head-lamp controller, a remote keyless entry receiver, a cellular phone including its microphone, a vehicle status indicator and the like) to share components and circuitry with the electrochromic mirror function control circuitry and electrochromic mirror assembly itself. Also, a convenient way to mount a non-windshield-contacting rain sensor such as described by Hegyl is by attachment, such as by snap-on attachment, as a module to the mirror channel mount such as is described in US Patent 5,576,678 entitled "Mirror Support Bracket," invented by R. Hook et al. and issued November 19, 1996, the disclosure of which is hereby incorporated by reference herein. The mirror mount and/or windshield button may optionally be specially adapted to accommodate a non-windshield-mounting rain sensor module. Such mounting as a module is readily serviceable and attachable to a wide variety of interior mirror assemblies (both electrochromic and non-electrochromic such as prismatic, manually adjusted

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mirror assemblies), and can help ensure appropriate alignment of the non-windshield-mounted variety of rain sensor to the vehicle windshield insofar that the module attached to the mirror mount remains fixed whereas the mirror itself (which typically attaches to the mirror channel mount via a single or double ball joint) is movable so that the driver can adjust its field of view. Also, should smoke from cigarettes and the like be a potential source of interference to the operation of the non-windshield-contacting rain sensor, then a mirror-attached housing can be used to shroud the rain sensor unit and shield it from smoke (and other debris). Optionally, such ability to detect presence of cigarette smoke can be used to enforce a non-smoking ban in vehicles, such as is commonly requested by rental car fleet operators. Also, when a rain sensor (contacting or non-contacting) is used to activate the wiper on the rear window (rear backlight) of the vehicle, the sensor can be conveniently packaged and mounted with the CHMSL (center high mounted stop light) stop light assembly commonly mounted on the rear window glass or close to it. Mounting of the rain sensor with the CHMSL stop light can be aesthetically appealing and allow sharing of components/wiring/circuitry.

The electrochromic solid films can be used with interior rearview mirrors equipped with a variety of features such as a high/low (or daylight running beam/low) headlamp controller, a hands-free phone attachment, a video camera for internal cabin surveillance and/or video telephone function, seat occupancy detection, map reading lights, compass/temperature display, fuel level and other vehicle status display, a trip computer, an intrusion detector and the like. Again, such features can share components and circuitry with the electrochromic mirror

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second push changes the temperature display to metric units (i.e., to degrees Celsius), on third push changes to Imperial units (i.e., degrees Fahrenheit) and on fourth push turns off the compass/temperature display, with the microprocessor controlling the logic of the display. Alternately, a single switch actuation turns on the display in Imperial units, the second actuation changes it to metric units, and third actuation turns the display off. Further, the displays and functions described herein can find utility also on outside rearview mirrors. For example, a transducer that receives and/or transmits information to a component of an intelligent highway system (such as is known in the automotive art) can be incorporated into an interior and/or outside rearview mirror assembly. Thus, for example, a transmitter/receiver for automatic toll booth function could be mounted at/within/on an outside sideview mirror assembly. A digital display of the toll booth transaction can be displayed by a display incorporated in the interior rearview mirror assembly. Optionally, a micro printer incorporated within the rearview mirror can print a receipt of the transaction. Similarly, for safety and security on the highways, GPS information, state of traffic information, weather information, telephone number information, and the like may be displayed and transmitted/received via transducers located at, within, or on an interior rearview mirror assembly and/or an outside sideview mirror assembly. Also, the interior rearview mirror assembly can include a link to the Worldwide Web via the INTERNET. Such as via a modem/cellular phone mounted within the vehicle, and preferably, mounted at, within or on the interior rearview mirror assembly. Thus, the driver can interact with other road users, can receive/transmit messages including E-mail, can receive weather and status of highway

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Further, a trainable garage door opener such as a universal garage door opener such as is available from Prince Corporation, Holland, Michigan under the tradename HOMELINK™, or the transmitter for a universal home access system that replaces the switch in a household garage that opens/closes the garage door with a smart switch that is programmable to a household specific code that is of the rolling code type, such as is available from TRW Automotive, Farmington Hills, Michigan under the tradename KWIKLINK™ may be mounted at, within, or on the interior mirror (or, if desired, the outside sideview mirror). Switches to operate such devices (typically up to three separate push type switches, each for a different garage door/security gate/household door) can be mounted on the mirror assembly, preferably user actuatable from the front face of the mirror housing. Preferably, the universal garage door opener HOMELINK™ unit or the universal home access KWIKLINK™ unit is mounted at, within or on the interior rearview mirror assembly. Optionally, such a unit could be mounted at, within or on an outside sideview mirror assembly.

The KWIKLINK™ Universal Home Access System (which operates on a rolling code, such as is commonly known in the home/vehicle security art) comprises a vehicle mounted transmitter and a receiver located in the garage. The KWIKLINK™ system is a low-current device that can be, optionally, operated off a battery source, such as a long life lithium battery. It is also compact and lightweight as executed on a single-or double-sided printed circuit board. The KWIKLINK™ printed circuit board can be mounted within the mirror housing (optionally adhered to a shock absorber

comprising a double-sticky tape anti-scatter layer on the rear of the reflector element (prismatic or electrochromic) such as is described in US Patent 5,572,354 entitled "Rear Mirror Assembly", invented by J. Desmond et al. and issued November 5, 1996, the disclosure of which is hereby incorporated by reference herein or may be accommodated within a detachable module, such as the pod described in US Patent 5,576,678 entitled "Mirror Support Bracket", invented by R. Hook et al. and issued November 19, 1996, the disclosure of which is hereby incorporated by reference herein, and with the detachable module attached to the mirror mount or to the mirror button. Mounting the KWIKLINK™ unit in a detachable module has advantages, particularly for aftermarket supply where a battery operated KWIKLINK™ unit can be supplied within a pod housing (with the necessary user actuatable button or buttons mounted on the pod and with the battery being readily serviceable either by access through a trap door and/or by detaching the pad from the mirror mount). By supplying a battery-operated, stand-alone, snap-on, detachable KWIKLINK™ mirror mount pod, the KWIKLINK™ home access system can be readily and economically provided to a broad range of mirrors (including non-electrical mirrors such as base prismatic mirrors, and electrical mirrors such as lighted prismatic mirrors and electo-optic mirrors, such as electrochromic mirrors). Further, a solar panel can be installed on the pod housing to recharge the battery.

Also, the pod module assembly may have a windshield button mount attached thereto or incorporated therein and have, in addition, a structure that replicates the windshield button standard on most vehicles manufactured in the United States. Thus, when a consumer purchases such an aftermarket product, the

consumer simply removes the existing interior rearview mirror assembly from the windshield button it is attached to in the vehicle. Then, the consumer attaches the pod module windshield button mount to the windshield button attached to the windshield (this can be achieved either by sliding on and securing with a screwdriver, or by snap-on in a manner conventional in the mirror mounting art). Finally, the consumer now attaches the rearview mirror assembly to the windshield button replication structure that is part of the aftermarket pod module. Since the windshield button shape is largely an industry standard (but the interior rearview mirror mount that attaches thereto is typically not standard), by using this "button on a button" pod module design, an aftermarket product (such as a pod module incorporating a home access transmitter, a universal garage door opener, a security monitor such as a pyroelectric intrusion detector (such as disclosed in copending US Patent Application 08/720,237 filed September 26, 1996, the disclosure of which is hereby incorporated by reference herein), a remote keyless entry receiver, a compass, a temperature and/or clock function and the like) may be readily installed by the vehicle owner, and the existing rearview minor assembly can be readily remounted in the vehicle.

Also, a cellular phone can be incorporated into the interior mirror assembly with its antenna, optionally, incorporated into the outside sideview mirror assembly or into the inside rearview mirror assembly. Such mounting within the mirror assemblies has several advantages including that of largely hiding the cellular phone and antenna from ready view by a potential thief. Further, a seat occupancy detector coupled to an air bag deployment/disable monitor can be located at, within or on the interior rearview mirror

assembly. The seat occupancy detector can be a video microchip or CCD camera seat occupancy detector, an ultrasonic detector or a pyroelectric detector, or their combination. Moreover, where more than one rearview mirror is being controlled or operated, or when several vehicle accessories are linked to, for example, an electrochromic interior or outside mirror, interconnections can be multiplexed, as is commonly known in the automotive art. Moreover, where it is desired to display external outdoor temperature within the interior cabin of the vehicle, a temperature sensor (such as a thermocouple or thermistor) can be mounted at, within or on an outside sideview mirror assembly (for example, it can protrude into the slipstream below the lower portion of the sideview mirror housing in a manner that is aesthetically and styling acceptable to the automakers and to the consumer) and with the temperature sensor output connected, directly or by multiplexing to a display (such as a vacuum fluorescent display) located in the interior cabin of the vehicle.

Preferably, the external temperature display is located at, within or on the interior rearview mirror assembly, optionally in combination with another display function such as a compass display (see US Patent Application No: not yet assigned, entitled "Vehicle Blind Spot Detection System" invented by K. Schofield et al., and filed February 12, 1997), or as a stand-alone pod attached as a module to a mirror support support member (see US Patent No. 5,576,687). Most preferably, the interior and outside mirror assemblies are supplied by the same supplier, using just-in-time sequencing methods, such as is commonly known in the automotive supply art and as is commonly used such as for supply of seats to vehicles. Just-in-time and/or sequencing techniques can be used to supply a specific option (for example, the option of configuring an external

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remote keyless entry control, and the like), it is useful to equip such assemblies with a standard connector (for example, a 10-pin parallel connector) so that a common standard wiring harness can be provided across an automaker's entire product range. Naturally, 5 multiplexing within the vehicle can help alleviate the need for more pins on such a connector, or allow a given pin or set of pins control more than one function.

10 Polychromic solid films can be used in added feature interior rearview mirror assemblies including those that include a loudspeaker (such as for a vehicle audio system, radio or the like, or for a cellular phone including a video cellular phone). Such loudspeaker 15 may be a high frequency speaker that is mounted at, within, or on the interior rearview mirror assembly (such as within the mirror housing or attached as a module-type pod to the mirror mount such as is described supra) and with its audio output, preferably, 20 directed towards the front windshield of the vehicle so that the windshield itself at least partially reflects the audio output of the speaker (that preferably is a tweeter speaker, more preferably is a compact (such as about 1" X 1" X 1" in dimensions or smaller), and most 25 preferably utilizes a neodymium magnet core) back into the interior cabin of the vehicle. The interior rearview mirror assembly can also include a microphone and a digital (or a conventional magnetic tape) 30 recorder that can be used by vehicle occupants to record messages and the like. A display can be provided that receives paging information from a pager incorporated in the interior rearview mirror assembly and that displays messages to the driver (preferably via a scrolling display) or to other occupants. The 35 interior rearview mirror assembly can include a digital storage of information such as phone numbers, message

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reminders, calendar information and the like, that can automatically, or on demand, display information to the driver.

5 Each of the documents cited in the present teaching is herein incorporated by reference to the same extent as if each document had individually been incorporated by reference.

10 In view of the above description of the instant invention, it is evident that a wide range of practical opportunities is provided by the teaching herein. The following examples illustrate the benefits and utility of the present invention and are provided only for purposes of illustration, and are not to be construed
15 so as to limit in any way the teaching herein.

EXAMPLES

20 In each of these examples, we selected random assemblies, fractured the substrates of the assemblies and scraped the polychromic solid film that had formed during the transformation process within the assembly from the broken substrate.

25 Scatter Safety Aspect of Electrochromic Devices Manufactured With Polychromic Solid Films

To demonstrate the safety performance of the electrochromic devices manufactured according to the these examples, we simulated the impact of an accident
30 by impacting the glass substrates of randomly selected devices with a solid object so as to shatter the glass substrates of those devices. We observed that in each instance the shattered glass was held in place by the polychromic solid film such that glass shards from the
35 broken substrates did not separate and scatter from the device.

**Stability and Cyclability of Electrochromic
Devices Manufactured with Polychromic Solid Films**

In general, we observed good cycle stability, heat
stability, performance under prolonged coloration and
5 ultraviolet stability of the electrochromic devices
manufactured with the polychromic solid films of the
present invention.

To demonstrate the cycle stability, ultraviolet
10 stability and thermal stability of some of the
electrochromic devices manufactured with the
polychromic solid films of the present invention, we
subjected electrochromic mirrors to (1) 15 seconds
color -- 15 seconds bleach cycles at both room
15 temperature and about 50 °C; (2) ultraviolet stability
tests by exposing the electrochromic mirrors to at
least about 900 KJ/m² using a Xenon Weatherometer as per
SAE J1960 and (3) thermal stability tests at about 85
°C.

20 In these mirrors, we observed no change of
electrochromic performance or degrading of the
electrochromic devices after more than about 100,000
cycles (15 seconds color -- 15 seconds bleach) at room
25 temperature and more than about 85,000 cycles (15
seconds color -- 15 seconds bleach) at about 50 °C, and
after exposure to about 900 KJ/m² of ultraviolet
radiation and to about 85 °C for about 360 hours
indicating excellent cycle stability and
30 weatherability.

Example 1

In this example, we chose a RMPT-HVBF₄ electrochromic
35 pair, in conjunction with a commercially available
ultraviolet curable formulation, to illustrate the

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beneficial properties and characteristics of the polychromic solid films and electrochromic interior automotive mirrors manufactured therewith.

A. Synthesis and Isolation of RMPT

We synthesized 2-methyl-phenothiazine-3-one according to the procedure described in European Patent Publication EP 0 115 394 (Merck Frosst Canada). To reduce MPT to RMPT, we followed the redox procedure described in commonly assigned co-pending United States patent application Serial No. 07/935,784 (filed August 27, 1992).

B. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition according to the present invention comprising about 3.7% HVBF₄ (as a cathodic compound), about 1.6% RMPT (as an anodic compound), both homogeneously dispersed in a combination of about 47.4% propylene carbonate (as the plasticizer) and, as a monomer component, about 52.6% "IMPRUV" (an ultraviolet curable formulation). We thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

C. Mirror Assembly with Electrochromic Monomer Composition

We assembled interior automotive mirrors from HW-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15.ohms per square. The dimensions of the mirror assemblies were about 2.5"

x 10" x 37 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 37 μ m also applied.

We placed into the mirror assemblies the electrochromic monomer composition of Example 1(B), supra, by the vacuum backfilling technique [as described in Varaprasad III, supra].

D. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromatic Solid Film

Once the electrochromic monomer composition of Example 1(B), supra, was uniformly applied within the mirror assemblies of Example 1(C), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B. While the belt advanced initially at a rate of about twenty-five feet per minute, we exposed the assemblies to ultraviolet radiation generated by the D fusion lamp of the F-300 B. We passed the assemblies under the fusion lamp light eight times at that rate, pausing momentarily between passes to allow the assemblies to cool, then eight times at a rate of about fifteen feet per minute again pausing momentarily between passes to allow the assemblies to cool and finally three times at a rate of about ten feet per minute with the aforementioned pausing between passes.

E. Use of Electrochromic Mirror

We applied a potential of about 1.3 volts to one of the electrochromic mirrors, and thereafter observed this mirror to color rapidly and uniformly to a bluish purple color.

In addition, we observed that the high reflectance at the center portion of the mirror was about 71%

reflectance which decreased to a low reflectance of about 10.8% when about 1.3 volts was applied thereto. The response time for the reflectance to change from about 70% to about 20% when that potential was applied thereto was about 3.7 seconds. We made this
5 determination by following the SAE J964a standard procedure of the Society of Automotive Engineers, with a reflectometer -- set in reflectance mode -- equipped with a light source (known in the art as Illuminant A) and a photopic detector assembly interfaced with a data
10 acquisition system.

We also observed that the mirror bleached from about 20% reflectance to about 60% reflectance in a response time of about 7.1 seconds under about a zero applied
15 potential. We noted the bleaching to be uniform, and the bleached appearance to be silvery.

Example 2

20 In this example, we chose a RMPT-HVBF₄ electrochromic pair, in conjunction with a combination of commercially available ultraviolet curable formulations, to illustrate the beneficial properties and characteristics of the polychromic solid film and the
25 electrochromic interior automotive mirrors manufactured therewith by using the sandwich lamination technique.

A. Preparation of Electrochromic Monomer Composition

30 We prepared an electrochromic monomer composition comprising by weight about 2.6% HVBF₄ (as a cathodic compound), about 1.2% RMPT (as an anodic compound), both homogeneously dispersed in a combination of about
35 40% propylene carbonate (as a plasticizer) and, in combination as a monomer component, about 50% "QUICK

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CURE" B-565 (an acrylated urethane/ultraviolet curable formulation) and about 10% "ENVIBAR" UV 1244 (a cycloalkyl epoxide/ultraviolet curable formulation). We thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

10 In this example, we assembled interior automotive mirrors by dispensing a portion of the electrochromic monomer composition of Example 2(A), supra, onto the conductive surface of a tin oxide-coated glass substrate (the other surface of the substrate being
15 silver-coated so as to form a mirror) onto which we also placed 37 μ m glass beads, and then positioned thereover the conductive surface of a clear, tin oxide-coated glass substrate. These glass substrates, commercially available under the trade name "TEC-Glass" products as "TEC-20" from Libbey-Owens-Ford Co.,
20 Toledo, Ohio, having dimensions of about 3" x 6", were assembled to form an interpane distance between the glass substrates of about 37 μ m. In this way, the electrochromic monomer composition was located between
25 the conductive surface of the two glass substrates of the mirror assemblies.

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

30 Once the electrochromic monomer composition of Example 2(A), supra, was uniformly applied within the mirror assemblies of Example 2(B), supra, we placed the
35 assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B. While the belt advanced initially at a

rate of about twenty feet per minute, we exposed the assemblies to ultraviolet radiation generated by the D fusion lamp of the F-300 B. We passed the assemblies under the fusion lamp light twelve times at that rate, pausing between every third or fourth pass to allow the assemblies to cool.

D. Use of Electrochromic Mirrors

We applied a potential of about 1.3 volts to one of the mirrors, and thereafter observed that the mirror colored rapidly and uniformly to a bluish purple color.

In addition, we observed that the high reflectance at the center portion of the mirror was about 57% reflectance which decreased to a low reflectance of about 9.3%. The response time for the reflectance to change from about 55% to about 20% was about 10 seconds when a potential of about 1.3 volts was applied thereto. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 50% reflectance in a response time of about 56 seconds under about zero applied potential. We noted the bleaching to be uniform, and the bleached appearance to be silvery.

Example 3

In this example, we compared the beneficial properties and characteristics of a polychromic solid film prepared using ferrocene as an anodic electrochromic compound, and manufactured within an exterior automotive mirror [Example 3(B)(1) and (D)(1), infra] and interior automotive mirrors [Example 3(B)(2) and (D)(2), infra]. We also installed an interior

automotive mirror as a rearview mirror in an automobile to observe its performance under conditions attendant with actual automotive use.

A. Preparation of Electrochromic Monomer Composition

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We prepared an electrochromic monomer composition comprising by weight about 4.4% EVC1O_4 (as a cathodic compound), about 2% ferrocene (as an anodic compound), both homogeneously dispersed in a combination comprising, in combination as the plasticizer component, about 48.6% propylene carbonate and about 8.8% cyanoethyl sucrose and, in combination as a monomer component, about 17.7% caprolactone acrylate and about 13.3% polyethylene glycol diacrylate (400). We also added about 0.9% benzoin *i*-butyl ether (as a photoinitiator) and about 4.4% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

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1. Exterior Automotive Mirror

We assembled exterior automotive mirrors from HW-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 3.5" x 5.5" x 74 μm , with a weather barrier of an epoxy resin coupled with spacers of about 74 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 3(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

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2. Interior Automotive Mirror

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We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 44 μm , with a weather barrier of an epoxy resin coupled with spacers of about 44 μm also applied.

15

We placed into these mirror assemblies the electrochromic monomer composition of Example 3(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

20

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

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Once the electrochromic monomer composition of Example 3(A), supra, was uniformly applied within each of the respective mirror assemblies of Example 3(B)(1) and (2), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 1(D), supra.

30

D. Use of Electrochromic Mirror

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We applied a potential of about 1.3 volts to one of the electrochromic mirrors of Example 3(B), supra, and to

two of the electrochromic mirrors of Example 3(C),
supra. Our observations follow.

1. Exterior Automotive Mirror

5 We observed that the electrochromic mirror colored
rapidly and uniformly to a blue color with a greenish
hue.

10 In addition, we observed that the high reflectance at
the center portion of the exterior mirror decreased
from about 80.5% to about 5.7%, with a change in the
reflectance of about 70% to about 20% in a response
time of about 5.0 seconds when a potential of about 1.3
volts was applied thereto. We made this determination
15 by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about
10% reflectance to about 60% reflectance in a response
time of about 9.2 seconds, under about a zero applied
20 potential. We noted the bleaching to be uniform, and
the bleached appearance to be silvery.

2. Interior Automotive Mirror

25 We observed that each of a first and second
electrochromic mirror colored rapidly and uniformly to
a blue color with a greenish hue.

30 In addition, we observed that for the first mirror the
high reflectance at the center portion of the interior
mirror decreased from about 80.2% to about 6.3%, with a
change in the reflectance of about 70% to about 20% in
a response time of about 3.1 seconds when a potential
of about 1.3 volts was applied thereto. The second
35 mirror exhibited comparable results, with the
reflectance decreasing from about 78.4% to about 7.5%

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in about 2.7 seconds. We made these determinations by the reflectometer described in Example 1, supra.

We also observed that the first mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 3.9 seconds under about a zero applied potential, and the second mirror bleached to the same extent in about 3.6 seconds. We noted the bleaching to be uniform, and the bleached appearance to be silvery.

We have successfully installed and operated such an electrochromic mirror in an automobile as a rearview mirror and achieved excellent results.

Example 4

In this example, we chose *t*-butyl ferrocene as the anodic electrochromic compound together with a monomer component containing the combination of a monomer and a commercially available ultraviolet curable formulation to illustrate the beneficial properties and characteristics of the polychromic solid films made therefrom and the electrochromic interior automotive mirrors manufactured therewith.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 3.9% EVC10_4 (as a cathodic compound), about 2.3% *t*-butyl ferrocene (as an anodic compound), both homogeneously dispersed in a combination comprising about 61.7% propylene carbonate (as a plasticizer) and, in combination as a monomer component, about 10.7% caprolactone acrylate and about 10.6% "SARBOX" acrylate resin (SB 500) (an ultraviolet

curable formulation). We also added about 1.3%
"IRGACURE" 184 (as a photoinitiator) and about 4.4%
"UVINUL" N 35 (as an ultraviolet stabilizing agent),
and thoroughly mixed this electrochromic monomer
composition to ensure that a homogeneous dispersion of
the components was achieved.

**B. Mirror Assembly With Electrochromic
Monomer Composition**

We assembled interior automotive mirrors from HWG-ITO
coated glass substrates (where the conductive surface
of each glass substrate faced one another), with both
the clear, front glass and the silvered, rear glass
having a sheet resistance of about 15 ohms per square.
The dimensions of the mirror assemblies were about 2.5"
x 10" x 53 μ m, with a weather barrier of an epoxy resin
coupled with spacers of about 53 μ m also applied.

We placed into these mirror assemblies the
electrochromic monomer composition of Example 4(A),
supra, using the vacuum backfilling technique [as
described in Varaprasad III, supra].

**C. Transformation of Electrochromic
Monomer
Composition Within Mirror
to Polychromic Solid Film**

Once the electrochromic monomer composition of Example
4(A), supra, was uniformly applied within the mirror
assemblies of Example 4(B), supra, we placed the
assemblies onto the conveyor belt of a Fusion UV Curing
System F-300 B, and exposed the assemblies to
ultraviolet radiation in the same manner as described
in Example 1(D), supra.

D. Use of Electrochromic Mirror

5 We applied a potential of about 1.3 volts to one of the electrochromic mirrors of Example 4(B) and (C), supra, and observed this mirror to color rapidly and uniformly to a blue color with a greenish hue.

10 In addition, we observed that the high reflectance at the center portion of the mirror was about 79.3% reflectance which decreased to a low reflectance of about 9.8% when about 1.3 volts was applied thereto. The response time for the reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.3 seconds. We made this
15 determination by the reflectometer described in Example 1, supra.

20 We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 3.0 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Examples 5 through 8

25 In Examples 5 through 8, we compared the beneficial properties and characteristics of polychromic solid films prepared from ferrocene, and three alkyl derivatives thereof, as the anodic electrochromic compound and manufactured within interior automotive
30 mirrors.

Example 5

35

**A. Preparation of Electrochromic
Monomer Composition**

We prepared an electrochromic monomer composition according to the present invention comprising about 3.5% EVClO_4 (as a cathodic compound), about 2.1% dimethyl ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 51.5% propylene carbonate (as a plasticizer) and about 34.3% "QUICK CURE" B-565 (as a monomer component). We also added about 8.6% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

**B. Mirror Assembly With Electrochromic
Monomer Composition**

We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 53 μm , with a weather barrier of an epoxy resin coupled with spacers of about 53 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 5(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

**C. Transformation of Electrochromic
Monomer
Composition Within Mirror
to Polychromic Solid Film**

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We prepared an electrochromic monomer composition according to the present invention comprising about

3.5% EVC10_4 (as a cathodic compound), about 2.3% *n*-butyl ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 51.3% propylene carbonate (as a plasticizer) and about 34.3% "QUICK CURE" B-565 (as a monomer component). We also added about 8.6% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly with Electrochromic Monomer Composition

We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 53 μm , with a weather barrier of an epoxy resin coupled with spacers of about 53 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 6(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 6(A), supra, was uniformly applied within the mirror assemblies of Example 6(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to

ultraviolet radiation in the same manner as described in Example 2(C), supra.

D. Use of Electrochromic Mirror

5 We applied a potential of about 1.3 volts to one of the electrochromic mirrors, and thereafter observed this mirror to color rapidly and uniformly to a blue color with a greenish hue.

10 In addition, we observed that the high reflectance at the center portion of the mirror was about 73.8% reflectance which decreased to a low reflectance of about 7.8% when about 1.3 volts was applied thereto. The response time for the reflectance to change from
15 about 70% to about 20% when that potential was applied thereto was about 2.5 seconds. We made this determination by the reflectometer described in Example 1, supra.

20 We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 4.3 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 7

A. Preparation of Electrochromic Monomer Composition

30 We prepared an electrochromic monomer composition according to the present invention comprising about 3.5% EVClO_4 (as a cathodic compound), about 2.3% *t*-butyl ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 51.3% propylene
35 carbonate (as a plasticizer) and about 34.3% "QUICK CURE" B-565 (as a monomer component). We also added

about 8.6% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

5

**B. Mirror Assembly with Electrochromic
Monomer Composition**

10

We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 53 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 53 μ m also applied.

15

We placed into these mirror assemblies the electrochromic monomer composition of Example 7(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

20

**C. Transformation of Electrochromic
Monomer
Composition Within Mirror
to Polychromic Solid Film**

25

Once the electrochromic monomer composition as of Example 7(A), supra, was uniformly applied within the mirror assemblies of Example 7(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 2(C), supra.

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D. Use of Electrochromic Mirror

5 We applied a potential of about 1.3 volts to one of the electrochromic mirrors, and thereafter observed this mirror to color rapidly and uniformly to a blue color with greenish hue.

10 In addition, we observed that the high reflectance at the center portion of the mirror was about 73.1% reflectance which decreased to a low reflectance of about 7.8% when about 1.3 volts was applied thereto. The response time for the reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.5 seconds. We made this
15 determination by the reflectometer described in Example 1, supra.

20 We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 4.3 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 8

25 **A. Preparation of Electrochromic Monomer Composition**

30 We prepared an electrochromic monomer composition according to the present invention comprising about 3.5% EVC10_4 (as a cathodic compound), about 1.8% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 51.8% propylene carbonate (as a plasticizer) and about 34.3% "QUICK CURE" B-565 (as a monomer component). We also added
35 about 8.6% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic

monomer composition to ensure that a homogeneous dispersion of the components was achieved.

**B. Mirror Assembly with Electrochromic
Monomer Composition**

5

We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 53 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 53 μ m also applied.

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We placed into these mirror assemblies the electrochromic monomer composition of Example 8(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

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**C. Transformation of Electrochromic
Monomer
Composition Within Mirror
to Polychromic Solid Film**

5 Once the electrochromic monomer composition of Example
8(A), supra, was uniformly applied within the mirror
assemblies of Example 8(B), supra, we placed the
assemblies onto the conveyor belt of a Fusion UV Curing
System F-300 B, and exposed the assemblies to
10 ultraviolet radiation in the same manner as described
in Example 2(C), supra.

D. Use of Electrochromic Mirror

15 We applied a potential of about 1.3 volts to one of the
electrochromic mirrors, and thereafter observed this
mirror to color rapidly and uniformly to a blue color
with a greenish hue.

20 In addition, we observed that the high reflectance at
the center portion of the mirror was about 72.7%
reflectance which decreased to a low reflectance of
about 7.9% when about 1.3 volts was applied thereto.
The response time for the reflectance to change from
25 about 70% to about 20% when that potential was applied
thereto was about 2.7 seconds. We made this
determination by the reflectometer described in
Example 1, supra.

30 We also observed that the mirror bleached from about
10% reflectance to about 60% reflectance in a response
time of about 4.8 seconds under about a zero applied
potential. We noted the bleaching to be uniform.

35

Example 9

A. Preparation of Electrochromic Monomer Composition

5 We prepared an electrochromic monomer composition comprising by weight about 3.9% EVC10_4 (as a cathodic compound), about 1.2% *t*-butyl ferrocene and about 1.0% DMPA (in combination as the anodic compound),
homogeneously dispersed in a combination comprising
10 about 45% propylene carbonate, about 8.9% cyanoethyl sucrose and about 8.9% 3-hydroxypropionitrile (in combination as a plasticizer component) and, in combination as a monomer component, about 17.7% caprolactone acrylate, about 11.5% polyethylene glycol
15 diacrylate (400) and about 1.8% 1,6-hexanediol diacrylate. We also added about 0.9% "IRGACURE" 184 (as a photoinitiator) and about 4.4% "UVINUL N 35" (as an ultraviolet stabilizing agent), and we thoroughly mixed this electrochromic monomer composition to ensure
20 that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly with Electrochromic Monomer Composition

25 We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with the clear, front glass and the silvered, rear glass having
30 a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 44 μm , with a weather barrier of an epoxy resin coupled with spacers of about 44 μm also applied.

35 We placed into these mirror assemblies the electrochromic monomer composition of Example 9(A),

supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

**C. Transformation of Electrochromic
Monomer
Composition Within Mirror
to Polychromic Solid Film**

Once the electrochromic monomer composition of Example 9(A), supra, was uniformly applied within the mirror assemblies of Example 9(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 1(D), supra.

D. Use of Electrochromic Mirror

We applied a potential of about 1.3 volts to one of the electrochromic mirrors, and thereafter observed that the mirror colored rapidly and uniformly to a bluish green color.

In addition, we observed that the high reflectance at the center portion of the mirror was about 78.2% decreased to a low reflectance of about 8.2%, with a change in the reflectance of about 70% to about 20% in a response time of about 1.9 seconds when a potential of about 1.3 volts was applied thereto. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 5.4 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 10

In this example, like Example 2, we chose to illustrate the sandwich lamination technique of manufacturing electrochromic devices to demonstrate its efficiency in the context of the present invention.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 3.0% EVC1O_4 (as a cathodic compound), about 1.9% t-butyl ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 31.7% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 31.7% "QUICK CURE" B-565 and about 31.7% Urethane Acrylate (Soft) (CN 953). We thoroughly mixed this electrochromic monomer composition to ensure that a homogenous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

We assembled rectangular mirrors by dispensing a portion of the electrochromic monomer composition of Example 10(A), supra, onto the conductive surface of a silvered "TEC-20" glass substrate onto which we also placed 150 μm glass beads, and then positioned thereover the conductive surface of a clear "TEC-20" glass substrate. We assembled these glass substrates, having dimensions of about 5.5" x 7", under moderate pressure to form an interpane distance between the glass substrates of about 150 μm . In this way, the electrochromic monomer composition was located between

the conductive surfaces of the two glass substrates of the mirror assemblies.

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 10(A), supra, was uniformly applied within the mirror assemblies of Example 10(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 2(C), supra.

D. Use of Electrochromic Mirrors

We applied a potential of about 1.3 volts to one of the electrochromic mirror, and thereafter observed that the mirror colored rapidly and uniformly to a greenish blue color.

In addition, we observed that the high reflectance at the center portion of the mirror was about 66.7% reflectance which decreased to a low reflectance of about 5.8%. The response time for the reflectance to change from about 60% to about 5.9% was about 30 seconds when a potential of about 1.3 volts was applied thereto. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 5.9% reflectance to about 60% reflectance in a response time of about 180 seconds under about zero applied potential.

Example 11

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Monomer Composition

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Monomer Composition

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D. Use of Electrochromic Mirror

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plasticizer component. Also, in this example, we chose to formulate the electrochromic monomer composition with an additional monomer having polyfunctionality as a compatibilizing agent for the polychromic solid film.

5

**A. Preparation of Electrochromic
Monomer Composition**

We prepared an electrochromic monomer composition according to the present invention comprising about
10 4.0% HVBF₄ (as a cathodic compound), about 1.7% RMPT (as an anodic compound), both homogeneously dispersed in a combination comprising, in combination as a plasticizer, about 10.2% propylene carbonate, about 17% benzyl acetone and about 14.7% cyanoethyl sucrose, and,
15 in combination as a monomer component, about 33.5% "QUICK CURE" B-565 and about 18.9% polyethylene glycol diacrylate (400). We thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

20

**B. Glazing Assembly with
Electrochromic Monomer Composition**

We constructed a glazing assembly consisting of a sun
25 roof model by dispensing a portion of the electrochromic monomer composition of Example 13(A), supra, onto the conductive surface of a "TEC-10" glass substrate onto which we also placed 100 μ m glass beads, and then positioned thereover another "TEC-10" glass
30 substrate, so that the electrochromic monomer composition was between and in contact with the conductive surface of the two glass substrates. We used "TEC-10" glass substrates having dimensions of about 6" x 16.5", with bus bars attached at the
35 lengthwise side of the substrates to create a distance therebetween of about 16.5". The interpane distance

between the "TEC-10" glass substrates was about 100 μ m.

**C. Transformation of Electrochromic
Monomer
Composition Within Glazing
Assembly to Polychromic Solid Film**

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Once the electrochromic monomer composition of Example 13(A), supra, was uniformly applied within the glazing assembly of Example 13(B), supra, we placed the assembly onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assembly to ultraviolet radiation in the same manner as described in Example 2(C), supra.

10

D. Use of Electrochromic Glazing Assembly

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We applied a potential of about 1.3 volts to the glazing assembly, and thereafter observed the assembly to color rapidly and uniformly to a bluish purple color.

20

In addition, we observed that the high transmission at the center portion of the glazing assembly was about 60.7% transmission which decreased to a low transmission of about 6.0% when about 1.3 volts was applied thereto. The response time for the transmission to change from about 60% to about 10% when that potential was applied thereto was about 3.8 minutes. We made this determination by the detection method described in Example 1, supra, except that the reflectometer was set in transmission mode.

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We also observed that the glazing assembly bleached from about 10% transmission to about 45% transmission in a response time of about 4.2 minutes under about a zero applied potential.

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Example 14

5 In this example, we chose to manufacture large area electrochromic mirrors, by the two hole filling technique, to demonstrate the beneficial properties and characteristics of the polychromic solid films within large truck mirrors.

A. Preparation of Electrochromic Monomer Composition

10 We prepared an electrochromic monomer composition comprising by weight about 1.9% EVC1O_4 (as a cathodic compound), about 1.2% RMPT (as an anodic compound), both homogeneously dispersed in a combination
15 comprising about 53.3% propylene carbonate (as a plasticizer) and about 43.6% "QUICK CURE" B-565 (as a monomer component). We thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly with Electrochromic Monomer Composition

20 We assembled large truck mirrors from FW-ITO glass substrates (where the conductive surface of each glass
25 substrate faced one another), with the clear, front glass and the silvered, rear glass having a sheet resistance of about 6 to about 8 ohms per square. The dimensions of the mirror assemblies were about 6.5" x
30 15" x 44 μm , with a weather barrier of an epoxy resin coupled with spacers of about 44 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 14(A),
35 supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

Composition Within Mirror to Polychromic Solid Film

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Mirror

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polyethylene glycol diacrylate (400) and "IRGACURE"
184, with stirring and bubbling nitrogen gas through
the combination, and initiating cure by exposing this
combination to a source of fluorescent light at room
temperature for a period of time of about 10 minutes.
5 At this point, we removed the source of fluorescent
light, and combined therewith the EVC10_4 , t-butyl
ferrocene and "UVINUL" N 35 to obtain a homogeneously
dispersed electrochromic monomer composition. We
monitored the extent of cure by the increase of
10 viscosity.

**B. Mirror Assembly With Electrochromic
Monomer Composition**

15 We assembled interior automotive mirrors with HWG-ITO
coated glass substrates (where the conductive surface
of each glass substrate faced one another), with the
clear, front glass and the silvered, rear glass having
a sheet resistance of about 15 ohms per square. The
20 dimensions of the mirror assemblies were about 2.5" x
10" x 53 μm , with a weather barrier of an epoxy resin
coupled with spacers of about 53 μm also applied.

We placed into these mirror assemblies the
25 electrochromic monomer composition of Example 15(A),
supra, using the vacuum backfilling technique [as
described in Varaprasad III, supra].

**C. Transformation of Electrochromic
Monomer
30 Composition Within Mirror
to Polychromic Solid Film**

Once the electrochromic monomer composition of Example
15(A), supra, was uniformly applied within the mirror
35 assemblies of Example 15(B), supra, we placed the
assemblies onto the conveyor belt of a Fusion UV Curing

System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 1(D), supra.

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D. Use of Electrochromic Mirrors

We applied a potential of about 1.3 volts to one of the mirrors, and thereafter observed that the mirror colored rapidly and uniformly to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 82.6% reflectance which decreased to a low reflectance of about 8.8%. The response time for the reflectance to change from about 70% to about 20% was about 2.5 seconds at about room temperature and about the same when the surrounding temperature was reduced to about -28 °C when a potential of about 1.3 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 1.9 seconds at about room temperature and of about 7.4 seconds when the surrounding temperature was reduced to about -28 °C under about zero applied potential.

Example 16

In this example, we chose to manufacture the polychromic solid film from a commercially available epoxy resin together with a cross-linking agent to illustrate enhanced prolonged coloration performance attained when such combinations are used in the electrochromic monomer composition.

**A. Preparation of Electrochromic
Monomer Composition**

5 We prepared an electrochromic monomer composition
comprising by weight about 4.7% HVBf_4 (as a cathodic
compound), about 1.7% ferrocene (as an anodic
compound), both homogeneously dispersed in a
combination comprising about 64.5% propylene carbonate
(as a plasticizer) and about 26.5% "CYRACURE" resin
10 UVR-6105 (as a monomer component) and about 1.2% 2-
ethyl-2-(hydroxymethyl)-1,3-propanediol (as a cross-
linking agent). We also added about 1.4% "CYRACURE"
UVI-6990 (as a photoinitiator), and thoroughly mixed
this electrochromic monomer composition to ensure that
15 a homogeneous dispersion of the components was
achieved.

**B. Mirror Assembly With Electrochromic
Monomer Composition**

20 We assembled interior automotive mirrors HWG-ITO coated
glass substrates (where the conductive surface of each
glass substrate faced one another), with the clear,
front glass and silvered, rear glass having a sheet
resistance of about 15 ohms per square. The dimensions
25 of the mirror assemblies were about 2.5" x 10" x 53 μm ,
with a weather barrier of an epoxy resin coupled with
spacers of about 53 μm also applied.

30 We placed into these mirror assemblies the
electrochromic monomer composition of Example 16(A),
supra, using the vacuum backfilling technique [as
described in Varaprasad III, supra].

**C. Transformation of Electrochromic
Monomer
Composition Within Mirror
to Polychromic Solid Film**

5 Once the electrochromic monomer composition of Example
16(A), supra, was uniformly applied within the mirror
assemblies of Example 16(B), supra, we placed the
assemblies onto the conveyor belt of a Fusion UV Curing
System F-300 B, and exposed the assemblies to
10 ultraviolet radiation in the same manner as described
in Example 1(D), supra.

D. Use of Electrochromic Mirror

15 We applied a potential of about 1.3 volts to one of the
electrochromic mirrors prepared according to Examples
16(B) and (C), supra, and observed this mirror to color
rapidly and uniformly to a blue color with a greenish
hue.

20 In addition, we observed that the high reflectance at
the center portion of the mirror was about 80.0%
reflectance which decreased to a low reflectance of
about 7.3% when about 1.3 volts was applied thereto.
The response time for the reflectance to change from
25 about 70% to about 20% when that potential was applied
thereto was about 2.9 seconds. We made this
determination by the reflectometer described in
Example 1, supra.

30 We also observed that the mirror bleached from about
10% reflectance to about 60% reflectance in a response
time of about 3.8 seconds under about a zero applied
potential. We noted the bleaching to be uniform.

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We further observed that the mirror bleached uniformly and satisfactorily after prolonged coloration in excess of about 8 hours.

Example 17

In this example, like Example 16, we chose to manufacture polychromic solid films from a commercially available epoxy resin together with a cross-linking agent to illustrate enhanced prolonged coloration performance attained when such combinations are used in the electrochromic monomer composition.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition according to the present invention comprising about 4.7% HVBF₄ (as a cathodic compound), about 1.4% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 64.6% propylene carbonate (as a plasticizer), about 17.5% "CYRACURE" resin UVR-6105 (as a monomer component) and about 10.1% "CARBOWAX" PEG 1450 (as a cross-linking agent). We also added about 1.4% "CYRACURE" UVI-6990 (as a photoinitiator), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly with Electrochromic Monomer Composition

We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The

dimensions of the mirror assemblies were about 2.5" x 10" x 53 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 53 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 17(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 17(A), supra, was uniformly applied within the mirror assemblies of Example 17(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 1(D), supra.

D. Use of Electrochromic Mirror

We applied a potential of about 1.3 volts to one of the electrochromic mirrors, and thereafter observed this mirror to color rapidly and uniformly to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 75.2% reflectance which decreased to a low reflectance of about 7.6% when about 1.3 volts was applied thereto. The response time for the reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.4 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 4.2 seconds under about a zero applied potential. We noted the bleaching to be uniform.

5 We further observed that the mirror bleached uniformly and satisfactorily after prolonged coloration in excess of about 8 hours.

10 **Example 18**

In this example, we chose ferrocene as the anodic electrochromic compound together with a monomer component containing the combination of a monofunctional monomer and a difunctional monomer to 15 illustrate the beneficial properties and characteristics of polychromic solid films made therefrom.

20 **A. Preparation of Electrochromic Monomer Composition**

We prepared an electrochromic monomer composition comprising by weight about 4.3% EVC10_4 (as a cathodic compound), about 1.9% ferrocene (as an anodic 25 compound), both homogeneously dispersed in a combination comprising about 55.9% propylene carbonate (as a plasticizer) and, in combination as a monomer component, about 12.7% caprolactone acrylate and about 17.2% polyethylene glycol diacrylate (400). We also 30 added about 3.5% benzoin *i*-butyl ether (as a photoinitiator) and about 4.3% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogenous dispersion of the components was achieved.

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**B. Mirror Assembly With Electrochromic
Monomer Composition**

We assembled interior automotive mirrors from HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 44 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 44 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 18(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

**C. Transformation of Electrochromic
Monomer
Composition Within Mirror
to Polychromic Solid Film**

Once the electrochromic monomer composition of Example 18(A), supra, was uniformly applied within the mirror assemblies of Example 18(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B. While the belt advanced initially at a rate of about fifty feet per minute, we exposed the assemblies to ultraviolet radiation generated by the D fusion lamp of the F-300 B. We passed these mirror assemblies under the fusion lamp fifteen times pausing for two minute intervals between every third pass, then nine times at that rate pausing for two minute intervals between every third pass, and finally six times at a rate of about twenty-five feet per minute pausing for two minute intervals after every other pass.

D. Use of Electrochromic Mirror

We applied a potential of about 1.5 volts to one of the electrochromic mirrors of Examples 18(B) and (C), supra, and observed this mirror to color rapidly and uniformly to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 77.1% reflectance which decreased to a low reflectance of about 7.9% when about 1.5 volts was applied thereto. The response time for the reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.8 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 2.6 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 19

In this example, we chose ferrocene as the anodic electrochromic compound together with a monomer component containing the combination of a monomer and a commercially available ultraviolet curable formulation to illustrate the beneficial properties and characteristics of polychromic solid films made therefrom.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 4.3% EVClO_4 (as a cathodic

compound), about 1.9% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 55.9% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 10.3% caprolactone acrylate, about 15.5% polyethylene glycol diacrylate (400) and about 4.3% "SARBOX" acrylate resin (SB 500). We also added about 3.5% benzoin *i*-butyl ether (as a photoinitiator) and about 4.3% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

We assembled interior automotive mirrors with HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 53 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 53 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 19(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 19(A), supra, was uniformly applied within the mirror assemblies of Example 19(B), supra, we placed the

assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 18(C), supra.

5

D. Use of Electrochromic Mirrors

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We applied a potential of about 1.5 volts to one of the mirrors, and thereafter observed that the mirror colored rapidly and uniformly to a blue color with a greenish hue.

15

In addition, we observed that the high reflectance at the center portion of the mirror was about 79.6% reflectance which decreased to a low reflectance of about 7.6%. The response time for the reflectance to change from about 70% to about 20% was about 2.2 seconds when a potential of about 1.5 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

20

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 2.5 seconds under about zero applied potential.

25

Example 20

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In this example, we chose to manufacture interior rearview mirrors from polychromic solid films prepared with a commercially available epoxy resin together with a cross-linking agent to illustrate enhanced prolonged coloration performance attained when such combinations are used in the electrochromic monomer composition.

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A. Preparation of Electrochromic Monomer Composition

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to Polychromic Solid Film

Once the electrochromic monomer composition of Example 20(A), supra, was uniformly applied within the mirror assemblies of Example 20(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 1(D), supra.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors prepared according to Examples 20(B) and (C), supra, and observed this mirror to color rapidly and uniformly to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 76.9% reflectance which decreased to a low reflectance of about 7.9% when about 1.4 volts was applied thereto. The response time for the reflectance to change from about 70% to about 20% when that potential was applied thereto was about 3.1 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 3.3 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 21

In this example, we illustrate that a prolonged application of a bleach potential -- i.e., a potential having a polarity opposite to that used to achieve color --, having a magnitude greater than about 0.2 volts, and preferably about 0.4 volts, may be used to enhance bleach speeds of electrochromic devices, such as automotive rearview mirrors, manufactured with polychromic solid films as the medium of variable reflectance.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 4.3% EVClO_4 (as a cathodic compound), about 1.9% ferrocene (as an anodic compound), both homogeneously dispersed in a combination comprising about 60.2% propylene carbonate (as a plasticizer) and, in combination as a monomer component, about 8.6% caprolactone acrylate, about 12.9% polyethylene glycol diacrylate (400) and about 4.3% "SARBOX" acrylate resin (SB 500). We also added about 3.4% "IRGACURE" 184 (as a photoinitiator) and about 4.3% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

We assembled interior automotive mirrors from HW-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the front, clear glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square.

The dimensions of the mirror assemblies were about 2.5" x 10" x 44 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 44 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 21(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 21(A), supra, was uniformly applied within the mirror assemblies of Example 21(B), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 1(D), supra.

D. Use of Electrochromic Mirror

We applied a potential of about -0.7 volts to one of the electrochromic mirrors of Examples 21(B) and (C), supra, and observed that mirror reflectance at the center portion of the mirror remained high at about 76%.

Upon reversing the polarity of the applied potential and increasing the magnitude to about +1.5 volts, we observed this mirror to color rapidly and uniformly to a blue color.

In addition, we observed that the high reflectance at the center portion of the mirror decreased to a low reflectance of about 7.8%, with the response time for

the reflectance to change from about 70% to about 20% when that potential was applied thereto being about 2.4 seconds. We made this determination by the reflectometer described in Example 1, supra.

5 We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 1.7 seconds under a potential of about -0.7 volts with a high reflectance of about 78% re-established. We noted that when a potential of about 10 zero volts to about -0.2 volts was applied to the mirror to bleach the mirror from the fully dimmed stated, the response time to achieve this effect was about 2.0 seconds. We also noted that when a potential having a greater magnitude, such as about -0.8 volts to 15 about -0.9 volts, was applied to the mirror, the color assumed by the polychromic solid film may be controlled. For instance, a slight blue tint may be achieved at that aforesated greater negative potential using the electrochromic system of this example so that 20 the bleached state of the electrochromic mirror may be matched to the color appearance of conventional non-electrochromic blue-tint mirrors commonly featured on luxury automobiles.

25 Example 22

In this example, we illustrate that a gradient opacity panel, such as that which may be used as an electrochromic shade band on an automobile windshield, 30 may be created by configuring the bus bars on the electrochromic assembly so they are affixed partially around, or along the opposite sides, of the assembly, thus creating a transition between the areas of the device to which voltage is applied and those where no 35 voltage is applied.

**A. Preparation of Electrochromic
Monomer Composition**

5 We prepared an electrochromic monomer composition
comprising by weight about 2.1% EVClO_4 (as a cathodic
compound), about 1.4% *t*-butyl ferrocene (as an anodic
compound), both homogeneously dispersed in a
combination of about 54.2% propylene carbonate (as the
plasticizer), and, in combination as a monomer
10 component, about 28.6% B-565 and about 13.8% Urethane
Acrylate (Soft) (CN 953). We thoroughly mixed this
electrochromic monomer composition to ensure that a
homogeneous dispersion of the components was achieved.

**B. Panel Assembly With Electrochromic
Monomer Composition**

15 We constructed a panel assembly containing an
electrochromic shade band by dispensing a portion of
the electrochromic monomer composition of Example
20 22(A), supra, onto the conductive surface of a HW-ITO
coated glass substrate having a sheet resistance of
about 15 ohms per square. Onto this substrate we also
placed 100 μm glass beads, and then positioned
thereover another HW-ITO coated glass substrate having
25 a sheet resistance of about 15 ohms per square so that
the electrochromic monomer composition was between and
in contact with the conductive surface of the two glass
substrates. The dimensions of the assembly were about
4.5" x 14", with an interpane distance between the
30 glass substrates of about 100 μm .

We connected bus bars along the 14" sides of the panel
assembly only about 4" inward from the edge of each of
the opposing 14" sides. We thereafter affixed
35 electrical leads to the bus bars.

Once the electrochromic monomer composition of Example 22(A), supra, was uniformly applied within the window panel assembly of Example 22(B), supra, we placed the assembly onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the panel assembly to ultraviolet radiation in the same manner as described in Example 2(C), supra.

Once the polychromic solid film was formed, we applied a weather barrier of epoxy resin along, and over, the glass joints to prevent entry of environmental contaminants. This weather barrier consisted of a bead of "ENVIBAR" UV 1244 ultraviolet curable adhesive followed by the application of "SMOOTH-ON" room temperature cure epoxy (commercially available from Smooth-On Inc., Gillette, New Jersey).

We applied a potential of about 1.3 volts to the panel assembly, and thereafter observed that only the 4" region through which an electric field was formed colored rapidly, uniformly and intensely to a blue color. We also observed that color extended beyond that 4" region for a distance of about 1" in a gradient opacity which changed gradually from an intense coloration immediately adjacent the bus bar/non-bus bar transition to a bleached appearance beyond that additional 1" region or thereabouts.

In addition, we observed that the high transmittance at
35 the center portion of the panel assembly was about
79.6% transmittance which decreased to a low

transmittance of about 7.6%. The response time for the transmittance to change from about 70% to about 20% was about 2.2 seconds when a potential of about 1.5 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra, except that the reflectometer was set in transmission mode.

We also observed that the panel assembly bleached from about 10% transmittance to about 60% transmittance in a response time of about 2.5 seconds under about zero applied potential.

Example 23

In this example, like Example 3, supra, we installed the interior automotive mirror as a rearview mirror in an automobile to observe its performance under conditions attendant with actual use.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 3.0% EVC10_4 (as a cathodic compound), about 1.3% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 62.6% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 8.9% caprolactone acrylate, about 13.4% polyethylene glycol diacrylate (400) and about 4.5% "SARBOX" acrylate resin (SB 500). We also added about 1.8% "IRGACURE" 184 (as a photoinitiator) and about 4.5% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogenous dispersion of the components was achieved.

35 In addition, we observed that the high reflectance at
the center portion of the mirror was about 72.0%

reflectance which decreased to a low reflectance of about 7.5%. The response time for the reflectance to change from about 70% to about 20% was about 3.5 seconds when a potential of about 1.5 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 5.2 seconds under about zero applied potential.

We have successfully installed and operated this mirror in an automobile as a rearview mirror and have achieved excellent results.

Example 24

In this example, we chose to illustrate the beneficial properties and characteristics of polychromic solid films manufactured within an electrochromic sun roof panel.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition according to the present invention comprising about 1.4% EVClO_4 (as a cathodic compound), about 0.9% *t*-butyl ferrocene (as an anodic compound), both homogeneously dispersed in a combination comprising about 39% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 39% "QUICK CURE" B-565 and about 19.53% Urethane Acrylate (Soft) (CN 953). We thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

**B. Preparation of Sun Roof Panel Assembly and
Placement of Electrochromic Monomer**

Composition Therein

5 We prepared the glass substrates for use in the glazing
assembly of this example by placing flat "TEC-20" glass
substrates (with a black ceramic frit band around their
perimeter edge regions), having dimensions of about 12"
x 16", onto the mold of a bending instrument at room
temperature under ambient conditions, and then
10 increasing the temperature of the substrates to be bent
to at least about 500 °C thereby causing the substrates
to conform to the shape of the mold.

15 We also placed, as a spacer means, black drafting tape
(Zipatone, Inc., Hillside, Illinois), having a width
of about 0.0625" and a thickness of about 150 μ m, onto
a conductive surface of one of the bent "TEC-20" glass
substrates in about 1.5" intervals across the width of
the substrate. At such intervals, we found the black
20 drafting tape to be positioned in an aesthetically
appealing manner, and to maintain uniformity of the
electrochromic media across the full dimensions of the
panel.

25 We assembled the sun roof panel by dispensing a portion
of the electrochromic monomer composition of Example
24(A), supra, onto the conductive surface of the
substrate to be used as the concave interior surface
(i.e., the Number 4 surface), and placed thereover the
30 conductive surface of the substrate bearing the spacer
means so that the electrochromic monomer composition
was between and in contact with the conductive surface
of the glass substrates. We then placed the panel
assembly in a vacuum bag, gently elevated the
35 temperature and evacuated substantially most of the air
from the vacuum bag. In this way, the electrochromic

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monomer composition dispersed uniformly between the substrates under the pressure from the atmosphere.

**C. Transformation of
Electrochromic
Monomer Composition
Into Polychromic Solid Film**

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We then placed the sun roof panel assembly (still contained in the vacuum bag) into a Sunlighter model 1530 UV chamber, equipped with three mercury lamps (commercially available from Test-Lab Apparatus Co., Milford, New Hampshire), and allowed the sun roof panel to remain exposed to the ultraviolet radiation emitted by the lamps for a period of time of about 30 minutes. The interpane distance between the "TEC-20" glass substrates was about 150 μ m.

20

We thereafter attached bus bars at the 12" side of the substrates to create a distance therebetween of about 16". We then attached electrical leads to the bus bars.

D. Use of Electrochromic Sun Roof Panel

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We applied a potential of about 1.3 volts to the glazing assembly, and thereafter observed the panel to color rapidly and uniformly to a bluish purple color.

30

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In addition, we observed that the high transmission at the center portion of the sun roof panel was about 67% transmission which decreased to a low transmission of about 5% when about 1.3 volts was applied thereto. The response time for the transmission to change from about 60% to about 10% when that potential was applied thereto was about 3 minutes. We made this determination by the detection method described in

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Example 1, supra, except that the reflectometer was set in transmission mode.

We also observed that the glazing assembly bleached from about 5% transmission to about 60% transmission in a response time of about 6.5 minutes under about a zero applied potential.

The ultraviolet stability, scatter safety performance and/or electrochromic performance, and reduction in transmittance of near-infrared radiation of sun roof panels manufactured in accordance with the teaching herein, may be augmented by using the methods taught in Lynam III and Lynam V, and in commonly assigned United States Patent 5,239,406 (Lynam)].

Example 25

In this example, we chose to illustrate the beneficial properties and characteristics of polychromic solid films manufactured within an electrochromic sun visor having a segmented design.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition according to the present invention comprising about 2.4% EVClO_4 (as a cathodic compound), about 1.6% ferrocene (as an anodic compound), both homogeneously dispersed in a combination comprising about 48% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 32% "QUICK CURE" B-565 and about 16% Urethane Acrylate (Soft) (CN 953). We thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

We assembled the sun visor of this example from FW-ITO coated glass substrates, having dimensions of about 4" x 14" and a sheet resistance of about 6 to about 8 ohms per square, onto which we previously placed deletion lines to form three individual segments. We created these deletion lines by screening a photo-resist material onto the glass substrate prior to depositing the ITO coating, and thereafter applying a coat of ITO onto the photo-resist coated substrate, and washing away the photoetched resist material using an organic solvent, such as acetone.

15 We assembled the sun visor by placing onto the 14" edges of the conductive surface of one of the FW-ITO glass substrates "KAPTON" high temperature polyamide tape (E.I. du Pont de Nemours and Company, Wilmington, Delaware), having a thickness of 70 μm . We then
20 dispensed a portion of the electrochromic monomer composition of Example 25(A), supra, onto that conductive surface and then placed thereover the conductive surface of another substrate so that the electrochromic monomer composition was between and in
25 contact with the conductive surface of the glass substrates. The interpane distance between the substrates was about 70 μm .

30

Once the electrochromic monomer composition of Example 25(A), supra, was uniformly applied within the sun visor assembly of Example 25(B), supra, we placed the assembly onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assembly to ultraviolet

radiation in the same manner as described in Example 2(C), supra.

Upon completion of the transformation process, we applied "ENVIBAR" UV 1244 to the glass edges and joints and again exposed the sun visor to ultraviolet radiation to further weather barrier protect the sun visor. We then applied "SMOOTH-ON" epoxy to those portions of the sun visor to form a final weather barrier about the sun visor.

D. Use of Electrochromic Sun Visor

We applied a potential of about 1.3 volts to the sun visor, and thereafter observed the sun visor to color rapidly and uniformly to a bluish purple color.

In addition, we observed that the high transmission at the center portion of the sun visor was about 74.9% transmission which decreased to a low transmission of about 2.5% when about 1.5 volts was applied thereto. The response time for the transmission to change from the high transmission state to about 10% when that potential was applied thereto was about 9 seconds. We made this determination by the detection method described in Example 1, supra, except that the reflectometer was set in transmission mode.

We also observed that the sun visor bleached from about 10% transmission to about 70% transmission in a response time of about 15 seconds under about a zero applied potential.

The segmented portions of the sun visor of this example may be made in a horizontal direction or a vertical direction, and individual segments may be activated by connection to an individual segment addressing means,

such as a mechanical switch, a photosensor, a touch sensor, including a touch activated glass panel, a voice activated sensor, an RF activated sensor and the like. In addition, segments may be activated individually or as pluralities by responding to glare from the sun, such as when the sun rises from and falls toward the horizon, or as it traverses the horizon. This sun visor, as well as other electrochromic glazings, such as windows, sun roofs and the like, may use automatic glare sensing means that involve single or multiple photosensors, such as those disclosed in United States Patent 5,148,014 (Lynam).

Example 26

In this example, we assembled an interior automotive mirror as a rearview mirror to be installed in an automobile to observe its performance under conditions attendant with actual use.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 3.6% EVC10_4 (as a cathodic compound), about 1.6% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 61.9% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 11.1% polyethylene glycol monomethacrylate (400), about 11.1% polyethylene glycol diacrylate (400) and about 4.4% "SARBOX" acrylate resin (SB 500). We also added about 1.8% "IRGACURE" 184 (as a photoinitiator) and about 4.4% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that

a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

5

We assembled an interior automotive mirror with HWG-ITO coated glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 53 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 53 μ m also applied.

10

15

We placed into these mirror assemblies the electrochromic monomer composition of Example 26(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

20

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

25

Once the electrochromic monomer composition of Example 26(A), supra, was uniformly applied within the mirror assembly of Example 26(B), supra, we placed the assembly onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assembly to ultraviolet radiation in the same manner as described in Example 1(D), supra.

30

D. Use of Electrochromic Mirror

35

We applied a potential of about 1.5 volts to the mirror, and thereafter observed rapid and uniform coloration to a blue color with a greenish hue.

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In addition, we observed that the high reflectance at the center portion of the mirror was about 72.0% reflectance which decreased to a low reflectance of about 7.4%. The response time for the reflectance to change from about 70% to about 20% was about 2.1 seconds when a potential of about 1.5 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 4.0 seconds under about zero applied potential.

Example 27

In this example, we assembled automotive mirrors for use with the 1993 Lincoln Continental automobile. Specifically, Example 27(A), infra, illustrates the manufacture and use of an interior rearview mirror, and Example 27(B), infra, illustrates the use of an exterior mirror, sized for driver-side and passenger-side applications, to be installed in the automobile.

A. 1993 Lincoln Continental Interior Rearview Mirror

1. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 3.6% EVC10_4 (as a cathodic compound), about 1.6% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 62% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 8.9% caprolactone acrylate, about 13.3% polyethylene glycol diacrylate (400) and about

4.4% "SARBOX" acrylate resin (SB 500). We also added about 1.8% "IRGACURE" 184 (as a photoinitiator) and about 4.4% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

2. Interior Rearview Mirror Assembly With Electrochromic Monomer Composition

We assembled an interior rearview mirror, with an interpane distance of 53 μ m, from HWG-ITO coated 093 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. We also applied a weather barrier of an epoxy resin coupled with spacers of about 53 μ m.

We placed into these mirror assemblies the electrochromic monomer composition of Example 27(A)(1), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

3. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 27(A)(1), supra, was uniformly applied within the mirror assembly of Example 27(A)(2), supra, we placed the assembly onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assembly to ultraviolet radiation in the same manner as described in Example 1(D), supra.

4. Use of Electrochromic Mirror

We applied a potential of about 1.5 volts to the mirror, and thereafter observed rapid and uniform coloration to a blue color with a greenish hue.

5 In addition, we observed that the high reflectance at the center portion of the mirror was about 76.5% reflectance which decreased to a low reflectance of about 7.4%. The response time for the reflectance to change from about 70% to about 20% was about 2.2
10 seconds when a potential of about 1.5 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response
15 time of about 2.7 seconds under about zero applied potential.

**B. 1993 Lincoln Continental Exterior
Mirrors -- Driver-Side and Passenger-Side**

20

**1. Preparation of Electrochromic
Monomer Composition**

We prepared an electrochromic monomer composition comprising by weight about 2.6% EVC10_4 (as a cathodic
25 compound), about 1.2% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 63% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 9% caprolactone acrylate, about 13.5%
30 polyethylene glycol diacrylate (400) and about 4.5% "SARBOX" acrylate resin (SB 500). We also added about 1.8% "IRGACURE" 184 (as a photoinitiator) and about 4.5% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic
35 monomer composition to ensure that a homogeneous dispersion of the components was achieved.

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2. **Exterior Mirror Assemblies With
Electrochromic Monomer Composition**

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We assembled exterior mirrors, with an interpane distance of 74 μm , from FW-ITO coated 063 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 6 to about 8 ohms per square. We also applied a weather barrier of an epoxy resin coupled with spacers of about 74 μm .

We placed into these mirror assemblies the electrochromic monomer composition of Example 27(B)(1), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

3. **Transformation of Electrochromic
Monomer Composition Within Mirror
to Polychromic Solid Film**

Once the electrochromic monomer composition of Example 27(B)(1), supra, was uniformly applied within the mirror assemblies of Example 27(B)(2), supra, we placed the assemblies onto the conveyor belt of a Fusion UV Curing System F-300 B, and exposed the assemblies to ultraviolet radiation in the same manner as described in Example 1(D), supra.

4. **Use of Electrochromic Mirrors**

We applied a potential of about 1.5 volts to one of the mirrors, and thereafter observed rapid and uniform coloration to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 72% reflectance which decreased to a low reflectance of

about 8%. The response time for the reflectance to change from about 70% to about 20% was about 3.9 seconds when a potential of about 1.5 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 4.0 seconds under about zero applied potential.

Example 28

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 6.31% HVSS (as a cathodic compound), about 1.63% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 47.48% propylene carbonate and about 8.63% 3-hydroxypropionitrile (as a plasticizer), and, in combination as a monomer component, about 12.95% caprolactone acrylate, about 8.63% polyethylene glycol diacrylate (400) and about 8.63% "SARBOX" acrylate resin (SB 501). We also added, in combination as photoinitiators, about 0.13% "IRGACURE" 184 and about 1.29% "CYRACURE" UVI-6990 and about 4.32% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

We assembled an interior rearview mirror, with an interpane distance of 53 μm , from HWG-ITO coated 093 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. We also applied a weather barrier of an epoxy resin coupled with spacers of about 53 μm .

We placed into these mirror assemblies the electrochromic monomer composition of Example 28(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 28(A), supra, was uniformly applied within the mirror assembly of Example 28(B), supra, we placed the assembly onto the conveyor belt of a Hanovia UV Curing System (Hanovia Corp., Newark, New Jersey), fitted with UV lamp 6506A431, with the intensity dial set at 300 watts. We exposed the assembly to ultraviolet radiation in a similar manner as described in Example 1(D), supra, by passing the assembly under the UV lamp with the conveyor speed set at about 20% to about 50% for about 120 to about 180 multiple passes.

D. Use of Electrochromic Mirror

We applied a potential of about 1.2 volts to the mirror, and thereafter observed rapid and uniform coloration to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 73.9% reflectance which decreased to a low reflectance of about 7.4%. The response time for the reflectance to change from about 70% to about 20% was about 3.9 seconds when a potential of about 1.2 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

Example 29

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 4.38% DSMVClO₄ (as a cathodic compound) and about 0.57% EHPVClO₄ (as a cathodic compound), about 1.62% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 56.74% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 13.10% caprolactone acrylate, about 8.73% polyethylene glycol diacrylate (400), about 4.37% "SARBOX" acrylate resin (SB 500E50) and about 4.37% "CYRACURE" resin UVR-6110. We also added, in combination as photoinitiators, about 0.44% "IRGACURE" 184 and about 1.31% "CYRACURE" UVI-6990 and about 4.37% "UVINUL" N 35 (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

We assembled an interior rearview mirror, with an interpane distance of 53 μ m, from HWG-ITO coated 093

glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. We also applied a weather barrier of an epoxy resin coupled with spacers of about 53 μ m.

We placed into these mirror assemblies the electrochromic monomer composition of Example 29(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 29(A), supra, was uniformly applied within the mirror assembly of Example 29(B), supra, we placed the assembly onto the conveyor belt of a Hanovia UV Curing System (Hanovia Corp., Newark, New Jersey, fitted with UV lamp 6506A431, with the intensity dial set at 300 watts. We exposed the assembly to ultraviolet radiation in a similar manner as described in Example 1(D), supra, by passing the assembly under the UV lamp with the conveyor speed set at about 20% to about 50% for about 120 to about 180 multiple passes.

D. Use of Electrochromic Mirror

We applied a potential of about 1.2 volts to the mirror, and thereafter observed rapid and uniform coloration to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 79.6% reflectance which decreased to a low reflectance of

about 6.7%. The response time for the reflectance to change from about 70% to about 20% was about 2.8 seconds when a potential of about 1.2 volts was applied thereto. We made that determination by the reflectometer described in Example 1, supra.

Example 30

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 4.42% DSMVClO₄ (as a cathodic compound) and about 0.59% EHPVClO₄ (as a cathodic compound), about 1.65% ferrocene (as an anodic compound), both homogeneously dispersed in a combination of about 48.67% propylene carbonate (as a plasticizer), and, in combination as a monomer component, about 13.27% caprolactone acrylate, about 8.85% polyethylene glycol diacrylate (400), about 8.85% "SARBOX" acrylate resin (SB 500E50) and about 8.85% "CYRACURE" resin UVR-6110. We also added, in combination as photoinitiators, about 0.44% "IRGACURE" 184 and about 1.77% "CYRACURE" UVI-6990 and about 2.65% 2-hydroxy-4-octoxybenzophenone (as an ultraviolet stabilizing agent), and thoroughly mixed this electrochromic monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror Assembly With Electrochromic Monomer Composition

We assembled an interior rearview mirror, with an interpane distance of 53 μ m, from HWG-ITO coated 093 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having

a sheet resistance of about 15 ohms per square. We also applied a weather barrier of an epoxy resin coupled with spacers of about 53 μm .

We placed into these mirror assemblies the electrochromic monomer composition of Example 30(A), supra, using the vacuum backfilling technique [as described in Varaprasad III, supra].

C. Transformation of Electrochromic Monomer Composition Within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 30(A), supra, was uniformly applied within the mirror assembly of Example 30(B), supra, we placed the assembly onto the conveyor belt of a Hanovia UV Curing System (Hanovia Corp., Newark, New Jersey, fitted with UV lamp 6506A431, with the intensity dial set at 300 watts. We exposed the assembly to ultraviolet radiation in a similar manner as described in Example 1(D), supra, by passing the assembly under the UV lamp with the conveyor speed set at about 20% to about 50% for about 120 to about 180 multiple passes.

D. Use of Electrochromic Mirror

We applied a potential of about 1.3 volts to the mirror, and thereafter observed rapid and uniform coloration to a blue color with a greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 71.9% reflectance which decreased to a low reflectance of about 6.9%. The response time for the reflectance to change from about 70% to about 20% was about 3.9 seconds when a potential of about 1.3 volts was applied

thereto. We made that determination by the reflectometer described in Example 1, supra.

Example 31

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**A. Preparation of Electrochromic Monomer
Composition**

10 We prepared an electrochromic monomer composition
comprising by weight about 0.36% HUVPF₆ (as a cathodic
compound), about 0.97% EVC1O₄ (as a cathodic compound),
about 0.17% Ferrocene (FE, an anodic compound), about
15 0.39% 5,10-dihydro-5,10-dimethylphenazine (as an anodic
compound), all homogeneously dispersed in a combination
of about 89.68% propylene carbonate (as plasticizer)
and, in combination as a monomer component, about 0.65%
HDT (an isocyanate) and about 3.08% Lexorez 1931-50 (a
20 polyol), and about 0.03% T-1 (a tin catalyst), and
about 4.67% Uvinul N 35 (a UV stabilizer). We
thoroughly mixed this monomer composition to ensure
that a homogeneous dispersion of the components was
achieved.

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**B. Mirror assembly with Electrochromic
Monomer Composition**

In this example, we assembled interior automotive
mirrors from TEC-15 glass substrates (where the
30 conductive surface of each glass substrate faced one
another), with both the clear, front glass and the
silvered, rear glass having a sheet resistance of about
15 ohms per square. The dimensions of the mirror
assemblies were about 2.5" x 10" x 125 μ m, with a
35 weather barrier of an epoxy resin coupled with spacers
of about 125 μ m also applied.

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We placed into these mirror assemblies the electrochromic monomer composition of Example 31 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

5 **C. Transformation of Electrochromic
Monomer Composition within Mirror to
Polychromatic Solid Film**

10 Once the electrochromic monomer composition of Example
31 (A), supra, was uniformly applied within the mirror
assemblies of Example 31 (B), supra, we placed the
assemblies overnight at room temperature during which
time the monomer composition reacted to form in situ
15 the solid polymer matrix film inside the mirror. These
mirror assemblies were then placed in an electrically
heated convection oven maintained at about 80°C for
about 2 hours.

20 **D. Use of Electrochromic Mirror**

We applied a potential of about 1.4 volts to one of the
electrochromic mirrors, and observed this mirror to
color rapidly and uniformly to a gray color with bluish
hue.

25 In addition, we observed that the high reflectance at
the center portion of the mirror was about 72.3 %
reflectance which decreased to a low reflectance of
about 7.1% when about 1.4 volts was applied to thereto.
30 The response time for reflectance to change from about
70% to about 20% when that potential was applied
thereto was about 2.1 seconds. We made this
determination by the reflectometer described in Example
1, supra.

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We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 7.0 seconds under about a zero applied potential. We noted the bleaching to be uniform.

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Example 32

**A. Preparation of Prepolymer Composition
that includes a Viologen containing
Polyol**

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We prepared viologen containing polyol through copolymerization of ESMVClO₄ with caprolactone acrylate according to the following procedure: We prepared a reaction mixture comprising by weight about 4.86% ESMVClO₄ (a viologen with vinyl functionality), about 1.94% UVI 6990 (a photoinitiator), about 0.97% Irgacure 184 (a photoinitiator), all homogeneously dispersed in a combination comprising about 43.69% caprolactone acrylate (an acrylate with hydroxyl functionality) and 48.54% propylene carbonate and placed it in a sealed glass container. We placed the sealed glass container on a conveyor belt of a Fusion UV Curing System F-300B. While the belt advanced at a rate of about 10 feet per minute, we exposed the reaction mixture to ultraviolet radiation generated by the D fusion lamp of the F 300B. We passed the sealed glass container containing the reaction mixture under the fusion lamp light twenty five times at that rate, pausing momentarily between the passes to allow the prepolymer composition to cool. We used the resulting prepolymer composition that includes a viologen containing polyol to prepare the electrochromic monomer composition.

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**B. Preparation of Electrochromic Monomer
Composition**

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We prepared an electrochromic monomer composition comprising by weight about 2.11% prepolymer composition of Example 32 (A), supra (as a cathodic compound and polyol), about 1.97% EVC1O_4 (as a cathodic compound), and about 1.01% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 76.65% propylene carbonate (as plasticizer) and in combination as a monomer component, about 2.68% HDT (an isocyanate) and about 15.52% Desmophen 1700 (a polyol), and about 0.06% T-9 (a tin catalyst). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

C. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled interior automotive mirrors from TEC-15 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 105 μm , with a weather barrier of an epoxy resin coupled with spacers of about 105 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 32 (B), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

D. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 32 (B), supra, was uniformly applied within the mirror assemblies of Example 32 (C), supra, we placed the assemblies overnight at room temperature during which time the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror. These mirror assemblies were then placed in an electrically heated convection oven maintained at about 80°C for about 2 hours.

E. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 64.1% reflectance which decreased to a low reflectance of about 6.5% when about 1.4 volts was applied to thereto. The response time for reflectance to change from about 60% to about 20% when that potential was applied thereto was about 2.6 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 50% reflectance in a response time of about 12.7 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 33

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.3% HHVPF₆ (as a cathodic compound), about 0.97% EVClO₄ (as a cathodic compound), about 0.59% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 89.71% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 0.65% HDT (an isocyanate) and about 3.08% Lexorez 1931-50 (a polyol), and about 0.03% T-9 (a tin catalyst), and about 4.67% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled interior automotive mirrors from TEC 15 and from HW-ITO glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 105 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 105 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 33 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 33 (A), supra, was uniformly applied within the mirror assemblies of Example 33 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 2 hours whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 71.8 % reflectance which decreased to a low reflectance of about 7.0% when about 1.4 volts was applied to thereto. The response time for reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.2 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 4.9 seconds under about a zero applied potential. We noted the bleaching to be uniform.

E. Stability and Cyclability of Electrochromic Devices Manufactured with Polychromic Solid Films

To demonstrate the cycle stability of the electrochromic mirrors assemblies of Example 33 (B and C), supra, we subjected the electrochromic mirrors made from TEC 15 glass substrates to 20 seconds color --- 20

seconds bleach cycles at different test temperatures required by automotive specifications. We have observed good cycle stability after about 85,000 cycles which include about 25,000 cycles at 70 °C, about 20,000 cycles at -30 °C, and about 40,000 cycles at room temperature. We observed, that the high reflectance of the mirror at the center portion of the mirror changed from 71.8% to 71.0% and that the low reflectance changed from 7.0% to 7.5% after about 85,000 cycles. We also observed that the response time for reflectance change from about 70% to about 20% changed from 2.2 seconds to 2.7 seconds and the response time for reflectance change from about 10% to about 60% changed from 4.9 seconds to 5.2 seconds after about 85,000 cycles.

To demonstrate the ultraviolet stability, we exposed the electrochromic mirror assemblies made from HW-ITO glass substrate of Example 33 supra, to at least about 2600 kJ/m² using a Xenon weatherometer as per SAE J1960. We observed, that the high reflectance of the mirror at the center portion of the mirror changed from 79.4% to 78.9% and that the low reflectance changed from 6.0% to 6.25% after exposure to ultraviolet radiation. We also observed that the response time for reflectance change from about 70% to about 20% changed from 1.6 seconds to 1.7 seconds and the response time for reflectance change from about 10% to about 60% changed from 4.1 seconds to 4.4 seconds after exposure to ultraviolet radiation.

To demonstrate the thermal stability of the electrochromic mirror assemblies of Example 33 (B and C), supra, we placed the mirror assemblies made from HW-ITO glass substrates in an electric oven maintained at about 85 °C for at least about 400 hours. We observed, that the high reflectance of the mirror at

the center portion of the mirror changed from 79% to 77% and that the low reflectance changed from 6.1 % to 5.7% after the heat test. We also observed that the response time for reflectance change from about 70% to about 20% changed from 1.5 seconds to 1.7 seconds and the response time for reflectance change from about 10% to about 60% changed from 4.1 seconds to 4.4 seconds after the heat test.

The environmental and overall performance the electrochromic mirrors was suitable for use in a vehicle.

Example 34

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.37% HUVPF₆ (as a cathodic compound), about 0.96% EVClO₄ (as a cathodic compound), about 0.59% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 89.65% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 0.65% HDT (an isocyanate) and about 3.08% Lexorez 1931-50 (a polyol), and about 0.03% T-1 (a tin catalyst), and about 4.67% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled exterior automotive mirrors using TEC 15 glass for the front substrate and

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a multi-layer metal reflector coated glass (consisting of about 200 angstroms of rhodium undercoated with about 1500 angstroms of chromium, and with the chromium being disposed between the rhodium layer and the glass surface so as to serve as an adhesion promoter layer such as is described in US Application No. 08/238,521 filed May 5, 1994, the disclosure of which is hereby incorporated by reference herein) for the rear substrate (where the conductive surface of each glass substrate faced one another), with the clear front glass having a sheet resistance of about 15 ohms per square and the rear multi-layered reflector coated glass having a sheet resistance of about 5 ohms per square. The dimensions of the mirror assemblies were about 3.5" x 7.5" x 105 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 105 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 34 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 34 (A), supra, was uniformly applied within the mirror assemblies of Example 34 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 2 hours whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

5 In addition, we observed that the high reflectance at the center portion of the mirror was about 56.3% reflectance which decreased to a low reflectance of about 7.0% when about 1.4 volts was applied to thereto. The response time for reflectance to change from about
10 55% to about 20% when that potential was applied thereto was 1.2 seconds. We made this determination by the reflectometer described in Example 1, supra.

15 We also observed that the mirror bleached from about 10% reflectance to about 50% reflectance in a response time of about 5.8 seconds under about a zero applied potential. We noted the bleaching to be uniform.

20 **Example 35**

A. Preparation of Electrochromic Monomer Composition

25 We prepared an electrochromic monomer composition comprising by weight about 1.09% HUVPF₆ (as a cathodic compound), about 0.58% EVC10₄ (as a cathodic compound), about 0.59% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 89.34% propylene carbonate (as
30 plasticizer) and, in combination as a monomer component, about 0.84% HDT (an isocyanate) and about 2.88% Lexorez 1931-50 (a polyol), and about 0.03% T-1 (a tin catalyst), and about 4.65% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer
35 composition to ensure that a homogeneous dispersion of the components was achieved.

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**B. Mirror assembly with Electrochromic
Monomer Composition**

5 In this example, we assembled interior automotive
mirrors from TEC-15 glass substrates (where the
conductive surface of each glass substrate faced one
another), with both the clear, front glass and the
silvered, rear glass having a sheet resistance of about
15 ohms per square. The dimensions of the mirror
10 assemblies were about 2.5" x 10" x 105 μm , with a
weather barrier of an epoxy resin coupled with spacers
of about 105 μm also applied.

15 We placed into these mirror assemblies the
electrochromic monomer composition of Example 35 (A),
supra, using the vacuum back filling technique (as
described in Varaprasad III, supra).

**C. Transformation of Electrochromic Monomer
Composition within Mirror to Polychromic
20 Solid Film**

Once the electrochromic monomer composition of Example
35 (A), supra, was uniformly applied within the mirror
assemblies of Example 35 (B), supra, we placed the
25 assemblies overnight at room temperature during which
time the monomer composition reacted to form in situ
the solid polymer matrix film inside the mirror. These
mirror assemblies were then placed in an electrically
heated convection oven maintained at about 80°C for
30 about 2 hours.

D. Use of Electrochromic Mirror

35 We applied a potential of about 1.4 volts to one of the
electrochromic mirrors, and observed this mirror to
color rapidly and uniformly to a gray color with
greenish hue.

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In addition, we observed that the high reflectance at the center portion of the mirror was about 72.1 % reflectance which decreased to a low reflectance of about 7.3% when about 1.4 volts was applied to thereto. The response time for reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.0 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 7.9 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 36

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.3% HHVPF₆ (as a cathodic compound), about 0.96% EVClO₄ (as a cathodic compound), about 0.59% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 84.13% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 1.38% HDT (an isocyanate) and about 7.96% Lexorez 1931-50 (a polyol), and about 0.01% T-9 (a tin catalyst), and about 4.67% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled interior automotive mirrors from TEC-15 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 105 μm , with a weather barrier of an epoxy resin coupled with spacers of about 105 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 36 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 36 (A), supra, was uniformly applied within the mirror assemblies of Example 36 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 2 hours whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 69.9% reflectance which decreased to a low reflectance of

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In this example, we assembled interior automotive mirrors from TEC-15 glass substrates (where the conductive surface of each glass substrate faced one

another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 105 μm , with a weather barrier of an epoxy resin coupled with spacers of about 105 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 37 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 37 (A), supra, was uniformly applied within the mirror assemblies of Example 37 (B), supra, we placed the assemblies overnight at room temperature during which time the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror. These mirror assemblies were then placed in an electrically heated convection oven maintained at about 80°C for about 2 hours.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 74.0 % reflectance which decreased to a low reflectance of about 7.5% when about 1.4 volts was applied to thereto.

The response time for reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.0 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 6.2 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 38

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.52% HUEVC10₄ (as a cathodic compound), about 0.77% EVC10₄ (as a cathodic compound), about 0.59% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 88.75% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 0.93% HDT (an isocyanate) and about 3.74% Lexorez 1931-50 (a polyol), and about 0.03% T-1 (a tin catalyst), and about 4.67% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled interior automotive mirrors from TEC-15 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the

silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 105 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 105 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 38 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 38 (A), supra, was uniformly applied within the mirror assemblies of Example 38 (B), supra, we placed the assemblies overnight at room temperature during which time the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror. These mirror assemblies were then placed in an electrically heated convection oven maintained at about 80°C for about 2 hours.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 72.9 % reflectance which decreased to a low reflectance of about 7.1% when about 1.4 volts was applied to thereto.

The response time for reflectance to change from about 70% to about 20% when that potential was applied thereto was about 2.0 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 5.4 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 39

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.3% HHVPF₆ (as a cathodic compound), about 0.96% EVClO₄ (as a cathodic compound), about 0.49% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), and about 0.13% THAc having been previously reduced by contacting with zinc (US Patent No. 5,500,760 issued March 19, 1996 the disclosure of which is incorporated by reference herein) (as an anodic compound), all homogeneously dispersed in a combination of about 85.34% propylene carbonate and about 0.91% acetic acid (as plasticizer) and, in combination as a monomer component, about 1.59% HDT (an isocyanate) and about 5.42 % Lexorez 1931-50 (a polyol), and about 0.19% T-9 (a tin catalyst), and about 4.67% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled interior automotive mirrors from TEC-15 glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 105 μm , with a weather barrier of an epoxy resin coupled with spacers of about 105 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 39 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 39 (A), supra, was uniformly applied within the mirror assemblies of Example 39 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 2 hours whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 67.4% reflectance which decreased to a low reflectance of

about 6.6% when about 1.4 volts was applied to thereto. The response time for reflectance to change from about 65% to about 20% when that potential was applied thereto was about 2.5 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 8.3 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 40

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.3% HHVPP₆ (as a cathodic compound), about 0.97% EVC10₄ (as a cathodic compound), about 0.59% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 89.71% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 0.65% HDT (an isocyanate) and about 3.08% Lexorez 1931-50 (a polyol), and about 0.03% T-9 (a tin catalyst), and about 4.67% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled exterior automotive mirrors using clear HW-ITO glass for the front substrate and chromium metal coated glass for the rear

substrate (where the conductive surface of each glass substrate faced one another), with the clear front glass having a sheet resistance of about 15 ohms per square and the rear chrome glass having a sheet resistance of 5 ohms per square. The dimensions of the mirror assemblies were about 3.5" x 7.5" x 105 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 105 μ m also applied.

We placed into these exterior mirror assemblies the electrochromic monomer composition of Example 40 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 40 (A), supra, was uniformly applied within the mirror assemblies of Example 40 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 2 hours whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror.

D. Use of Exterior Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 52.7% reflectance which decreased to a low reflectance of

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about 6.4 % when about 1.4 volts was applied to thereto. The response time for reflectance to change from high reflectance to about 23% when that potential was applied thereto was about 1.6 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from low reflectance to about 40% reflectance in a response time of about 6.9 seconds under about a zero applied potential. We noted the bleaching to be uniform.

**E. Stability and Cyclability of
Electrochromic Devices Manufactured with
Polychromic Solid Films**

To demonstrate the electrical stability of the mirror assemblies of Example 40 (B and C), supra, we applied 1.4 volts and continuously colored the electrochromic mirrors for at least about 300 hours at room temperature. We observed that the high reflectance changed from 52.7% to 52.2% and the low reflectance remained unchanged at 6.4% after the continuous coloration test. We observed that the response time for reflectance to change from high reflectance to about 23% changed from 1.6 seconds to 2.0 seconds after the continuous coloration test and also that the response time for the mirror to bleach from low reflectance to about 40% reflectance remained steady at about 6.9 seconds before and after the continuous coloration test.

To demonstrate the cyclability of the mirror assemblies of Example 40 (B and C), supra, we applied 1.4 volts and continuously colored the electrochromic mirrors for at least about 300 hours at room temperature.

To demonstrate the cycle stability of the electrochromic mirrors assemblies of Example 40 (B and C), supra, we subjected the electrochromic mirrors to 20 seconds color --- 20 seconds bleach cycles at different test temperatures required by automotive specifications. We observed good cycle stability after about 80,000 cycles which include about 30,000 cycles at 70 °C, and about 50,000 cycles at room temperature. We observed, that the high reflectance of the mirror at the center portion of the mirror changed from 53.22 to 51.1% and that the low reflectance changed from 6.5% to 7.1% after the cycle test. We also observed that the response time for reflectance change from high reflectance to about 23% remained constant at about 1.9 seconds after the cycle test and the response time for reflectance change from low reflectance to about 40% changed from 5.7 seconds to 5.5 seconds after the cycle test.

Example 41

In this example, we chose to illustrate the beneficial properties and characteristics of the polychromic solid films manufactured within electrochromic glazings, or that may be used as small area transmissive devices, such as optical filters and the like.

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.37% HUVPF₆ (as a cathodic compound), about 0.96% EVC10₄ (as a cathodic compound), about 0.59% 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 89.65% propylene carbonate (as plasticizer) and, in combination as a monomer

component, about 0.65% HDT (an isocyanate) and about 3.08% Lexorez 1931-50 (a polyol), and about 0.03% T-1 (a tin catalyst), and about 4.67% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

**B. Glazing Assembly with Electrochromic
Monomer Composition**

In this example, we assembled electrochromic glazings from clear TEC 15 glass substrates (where the conductive surface of each glass substrate faced one another), with the glass having a sheet resistance of about 15 ohms per square. The dimensions of the glazing assemblies were about 2.5" x 10" x 105 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 105 μ m also applied.

We placed into these glazing assemblies the electrochromic monomer composition of Example 41 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

**C. Transformation of Electrochromic Monomer
Composition within Glazing to Polychromic
Solid Film**

Once the electrochromic monomer composition of Example 41 (A), supra, was uniformly applied within the glazing assemblies of Example 41 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 2 hours whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the glazing assemblies.

D. Use of Electrochromic Glazing

We applied a potential of about 1.4 volts to one of the electrochromic glazings of Example 41 (B and C), supra. We observed that the electrochromic glazings colored rapidly and uniformly to a gray color with greenish hue.

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In addition, we observed that the high transmission at the center portion of the glazing was about 77.1% transmission which decreased to a low transmission of about 10.3% when about 1.4 volts was applied to thereto. The response time for transmission to change from about 70% to about 20% when that potential was applied thereto was 4 seconds. We made this determination by the reflectometer described in Example 1, supra.

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We also observed that the glazing bleached from about 10% transmission to about 70% transmission in a response time of about 7.7 seconds under about a zero applied potential. We noted the bleaching to be uniform.

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Example 42

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A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.3% DVAVPF₆ (as a cathodic compound), about 1.15% EVC10₄ (as a cathodic compound), about 0.69 % 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 86.63% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 0.93 % HDT (an isocyanate) and about 5.59% Lexorez 1931-50 (a polyol), and about 0.05% dibutyltin dilaurate (a tin catalyst), and about 4.66%

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Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

5 **B. Mirror assembly with Electrochromic
 Monomer Composition**

10 In this example, we assembled interior automotive
 mirrors from TEC-15 glass substrates (where the
 conductive surface of each glass substrate faced one
15 another), with both the clear, front glass and the
 silvered, rear glass having a sheet resistance of about
 15 ohms per square. The dimensions of the mirror
 assemblies were about 2.5" x 10" x 105 μ m, with a
20 weather barrier of an epoxy resin coupled with spacers
 of about 105 μ m also applied.

 We placed into these mirror assemblies the
 electrochromic monomer composition of Example 42 (A),
25 supra, using the vacuum back filling technique (as
 described in Varaprasad III, supra).

**C. Transformation of Electrochromic Monomer
 Composition within Mirror to Polychromic
25 Solid Film**

 Once the electrochromic monomer composition of Example
 42 (A), supra, was uniformly applied within the mirror
 assemblies of Example 42 (B), supra, we placed the
30 assemblies in an electrically heated convection oven
 maintained at about 60°C for about 1 hour whereupon the
 monomer composition reacted to form in situ the solid
 polymer matrix film inside the mirror.

35 **D. Use of Electrochromic Mirror**

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with greenish hue.

5 In addition, we observed that the high reflectance at the center portion of the mirror was about 68.0% reflectance which decreased to a low reflectance of about 6.7% when about 1.2 volts was applied to thereto. The response time for reflectance to change from about 10 60% to about 20% when that potential was applied thereto was about 2.4 seconds. We made this determination by the reflectometer described in Example 1, supra.

15 We also observed that the mirror bleached from about 10% reflectance to about 60% reflectance in a response time of about 5.7 seconds under about a zero applied potential. We noted the bleaching to be uniform.

20 Example 43

**A. Preparation of Electrochromic Monomer
Composition**

25 We prepared an electrochromic monomer composition comprising by weight about 2.18% HUVPF₆ (as a cathodic compound), about 0.58 % 5,10-dihydro-5,10-dimethylphenazine (as an anodic compound), all homogeneously dispersed in a combination of about 30 88.87% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 1.3 % HDT (an isocyanate) and about 2.41% Lexorez 1931-50 (a polyol), and about 0.03% T-1 (a tin catalyst), and about 4.63% 35 Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

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**B. Mirror assembly with Electrochromic
Monomer Composition**

In this example, we assembled interior automotive mirrors from TEC-15 glass substrates (where the
5 conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 105 μm , with a
10 weather barrier of an epoxy resin coupled with spacers of about 105 μm also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 43 (A),
15 supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

**C. Transformation of Electrochromic Monomer
Composition within Mirror to Polychromic
20 Solid Film**

Once the electrochromic monomer composition of Example 43 (A), supra, was uniformly applied within the mirror
25 assemblies of Example 43 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 1 hour whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror.

D. Use of Electrochromic Mirror

We applied a potential of about 1.4 volts to one of the electrochromic mirrors, and observed this mirror to
35 color rapidly and uniformly to a gray color with greenish hue.

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In addition, we observed that the high reflectance at the center portion of the mirror was about 71.2% reflectance which decreased to a low reflectance of about 12.5% when about 1.2 volts was applied to thereto. The response time for reflectance to change from about 70% to about 20% when that potential was applied thereto was about 5.3 seconds. We made this determination by the reflectometer described in Example 1, supra.

We also observed that the mirror bleached from about 15% reflectance to about 50% reflectance in a response time of about 12.0 seconds under about a zero applied potential. We noted the bleaching to be uniform.

Example 44

A. Preparation of Electrochromic Monomer Composition

We prepared an electrochromic monomer composition comprising by weight about 0.66% HVSS (as a cathodic compound), about 1.52% EVC10_4 (as a cathodic compound), about 0.17% ferrocene (as an anodic compound), about 0.74% phenothiazine (as an anodic compound) all homogeneously dispersed in a combination of about 87.6% propylene carbonate (as plasticizer) and, in combination as a monomer component, about 4.61% dipentaerythritol pentaacrylate. We also added about 0.09% 1,1'-azobiscyclohexanecarbonitrile (as an initiator), about 4.61% Uvinul N 35 (a UV stabilizer). We thoroughly mixed this monomer composition to ensure that a homogeneous dispersion of the components was achieved.

B. Mirror assembly with Electrochromic Monomer Composition

In this example, we assembled interior automotive mirrors from HW-ITO glass substrates (where the conductive surface of each glass substrate faced one another), with both the clear, front glass and the silvered, rear glass having a sheet resistance of about 15 ohms per square. The dimensions of the mirror assemblies were about 2.5" x 10" x 125 μ m, with a weather barrier of an epoxy resin coupled with spacers of about 125 μ m also applied.

We placed into these mirror assemblies the electrochromic monomer composition of Example 44 (A), supra, using the vacuum back filling technique (as described in Varaprasad III, supra).

C. Transformation of Electrochromic Monomer Composition within Mirror to Polychromic Solid Film

Once the electrochromic monomer composition of Example 44 (A), supra, was uniformly applied within the mirror assemblies of Example 44 (B), supra, we placed the assemblies in an electrically heated convection oven maintained at about 80°C for about 2 hour whereupon the monomer composition reacted to form in situ the solid polymer matrix film inside the mirror.

D. Use of Electrochromic Mirror

We applied a potential of about 1.3 volts to one of the electrochromic mirrors, and observed this mirror to color rapidly and uniformly to a gray color with bluish hue.

In addition, we observed that the high reflectance at the center portion of the mirror was about 65% reflectance which decreased to a low reflectance of

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about 6% when about 1.3 volts was applied to thereto. We made this determination by the reflectometer described in Example 1, supra. We noted that the response time to color and also the response time to bleach the mirror was suitable for use in a vehicle.

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While we have provided the above examples of the foregoing invention for illustrative purposes employing preferred electrochromic compounds, monomer components and plasticizers, and other components it is to be understood that variations and equivalents of each of the prepared electrochromic monomer compositions identified herein will provide suitable, if not comparable, results when viewed in connection with the results gleaned from these examples. Without undue experimentation, those of ordinary skill in the art will find it readily apparent to prepare polychromic solid film with the beneficial properties and characteristics desirable for the specific application armed with the teaching herein disclosed. And, it is intended that such equivalents be encompassed by the claims which follow hereinafter.

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